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CEE 697z

Organic Compounds in Water and Wastewater

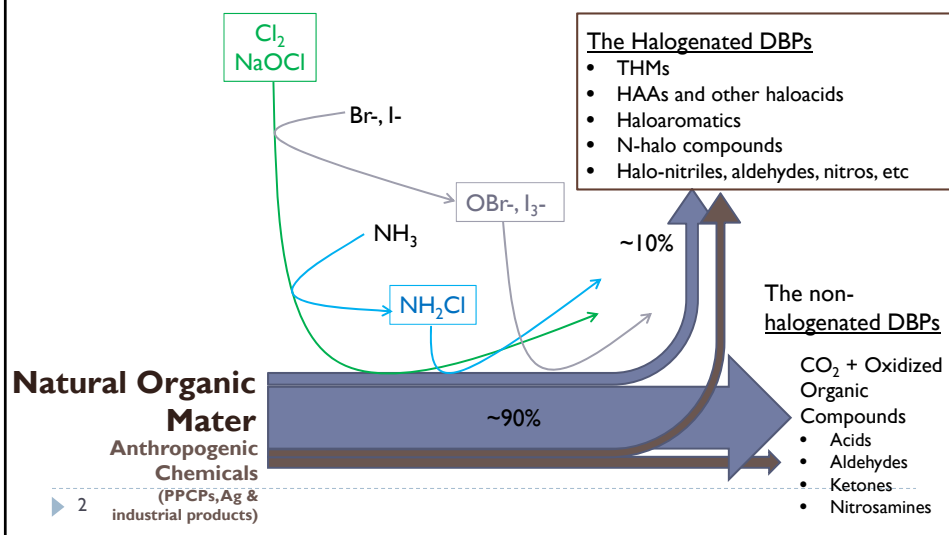
NOM and DBPs
Special Lecturer: Rassil El Sayess

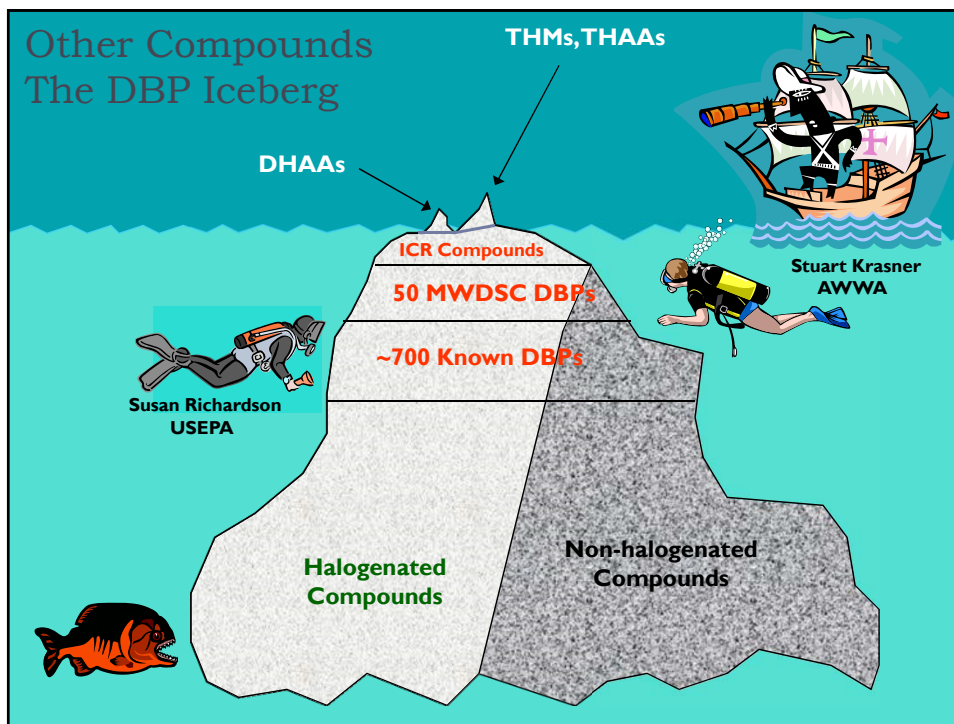
Lecture #9

http://www.ecs.umass.edu/eve/research/nyc_chloramines/literature.html

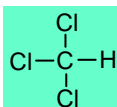
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Formation of Cl₂-driven DBPs

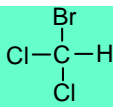




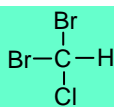
The Trihalomethanes (THMs)



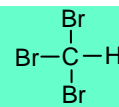
Chloroform



Bromodichloromethane



Chlorodibromomethane

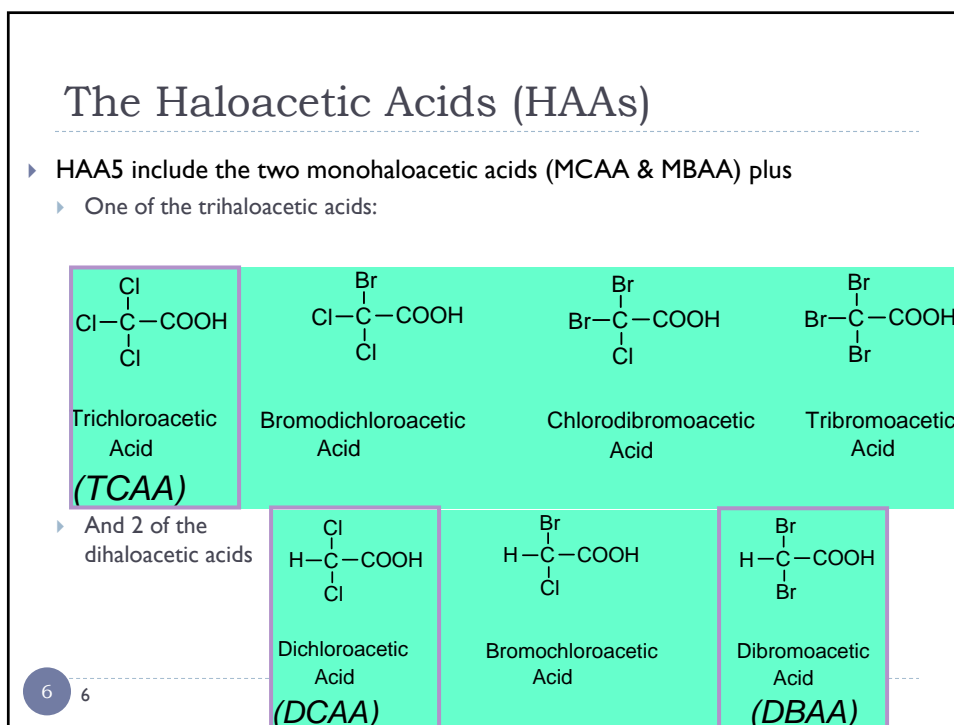
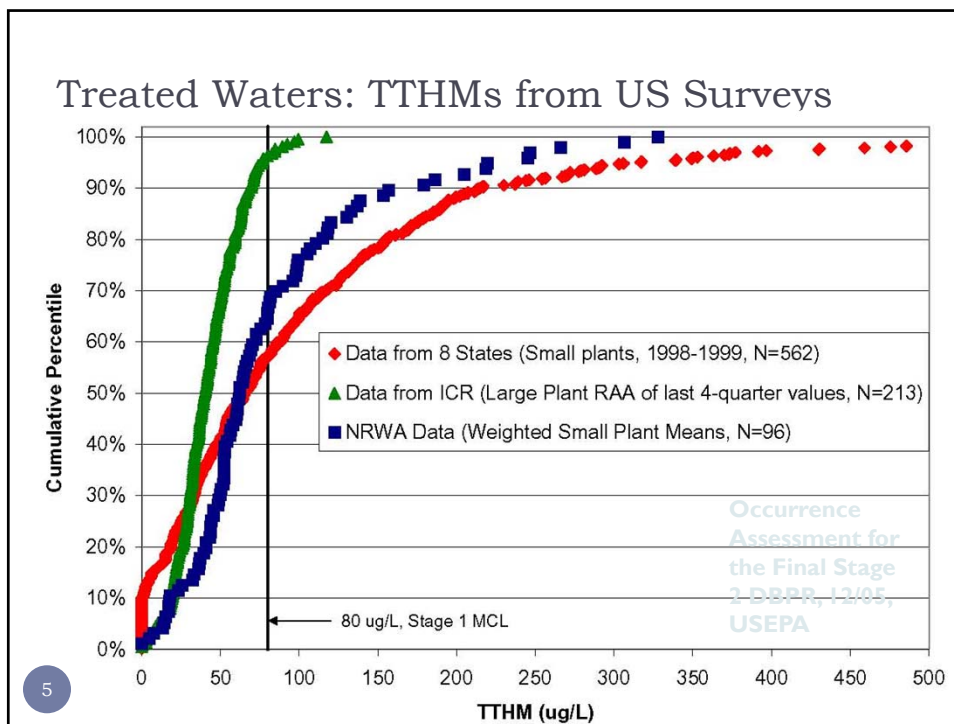


Bromoform

- Published in Dutch journal H₂O, Aug 19, 1972 issue
- Deduced that they were formed as byproducts of chlorination
- Proposed chemical pathways



1921-2010



Regulated Compounds

- ▶ THMs
 - ▶ HAA5
 - ▶ Bromate
 - ▶ Chlorite
 - ▶ The regulated compounds are
 - ▶ Common “end products” produced by almost all precursors
 - ▶ Chemically very stable
 - ▶ This is not typical of other DBPs
-

DBP Precursor Materials

General Groups

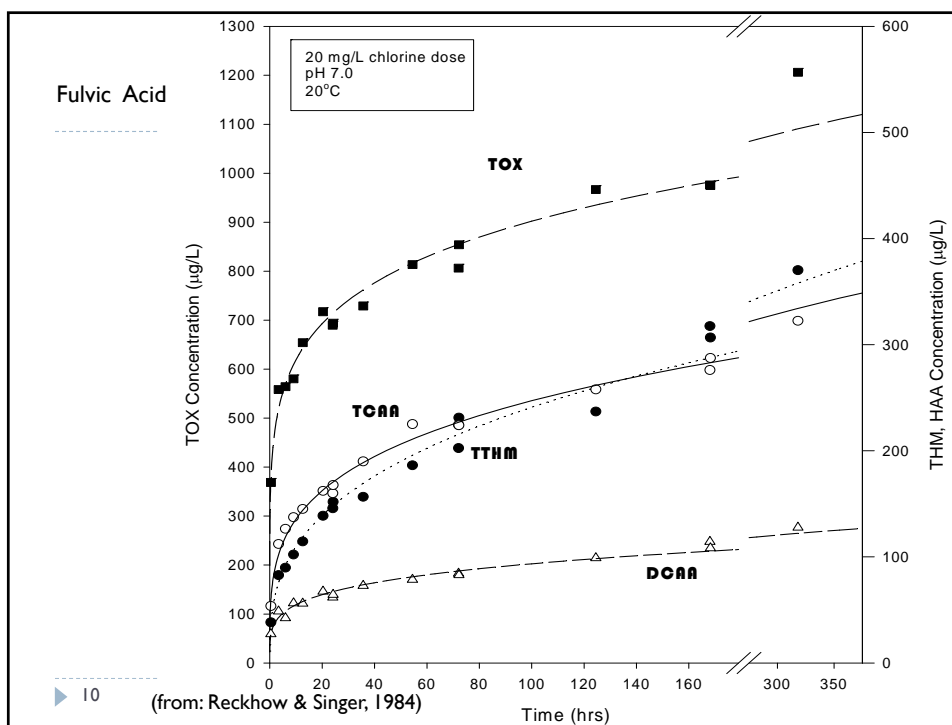
- ▶ Bulk NOM
 - ▶ Hydrophobic NOM
 - ▶ Acids (Fulvics & Humics)
 - ▶ Neutrals
 - ▶ Bases
 - ▶ Hydrophilic NOM
 - ▶ Acids, Bases, Neutrals
 - ▶ Mesophilic NOM
 - ▶ Acids, Bases, Neutrals
 - ▶ Soluble Metabolics
-

Specific Structures

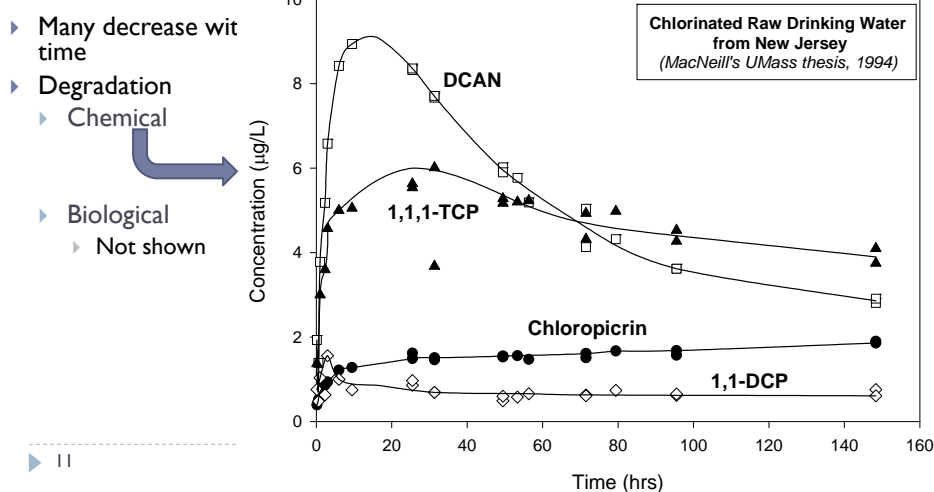
- ▶ Lignin
 - ▶ Carbohydrates
 - ▶ Proteins & Amino Acids
 - ▶ Terpenoids
 - ▶ Fatty Acids
 - ▶ Tannins
 - ▶ Anthropogenics
 - ▶ Ranitidine
-

DBP Data - Availability

- ▶ **Based on precursors**
 - ▶ Bulk NOM: most data, from raw & treated waters
 - ▶ NOM Fractions: some data
 - ▶ Specific Structures: far less data
- ▶ **Based on type of DBP**
 - ▶ Regulated compounds (THMs & HAAs)
 - ▶ Extensive Data, especially for bulk NOM
 - ▶ Common unregulated compounds
 - ▶ Moderate level, especially from ICR and selected "studies"
 - ▶ Emerging unregulated compounds
 - ▶ Very little data



Some Common Unregulated DBPs



Model Compound Studies

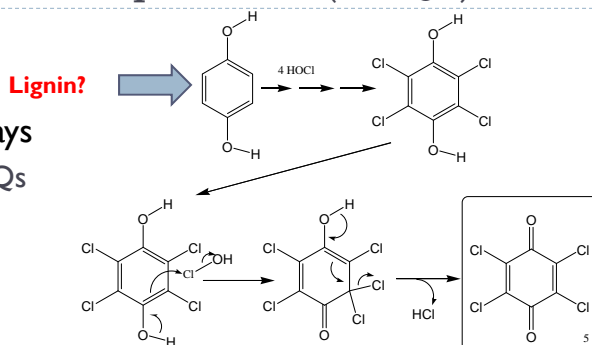
- ▶ Model compounds
 - ▶ Synthetically prepared in the lab: water that has been spiked with certain compounds
 - ▶ Most have been used to assess formation of regulated DBPs (THMs & HAAs)
 - ▶ Some have been conducted to find new DBPs and especially intermediates formed along the way to the final byproducts
- ▶

Lignin: Halobenzoquinones (HBQs)

- ▶ Many pathways
 - ▶ Plants to HQs

- ▶ Toxicity

- ▶ HQs are known to be reactive and damaging to DNA
- ▶ Postulated to be bladder carcinogen of high potency



Bull et al., 2006

▶ 13

Halobenzoquinones (cont.)

- ▶ Identified following QSAR deductive reasoning
 - ▶ SPE - LC/MS/MS method: Zhao et al., 2010
- ▶ Little occurrence data:
 - ▶ U Alberta: 7 samples in 2 publications
 - ▶ Dichloro (DCBQ): 14 ng/L median (165 ng/L max)
 - ▶ Others much lower
 - ▶ UMass: several dozen samples - unpublished
 - ▶ Dichloro: 306 ng/L high value

▶ 14

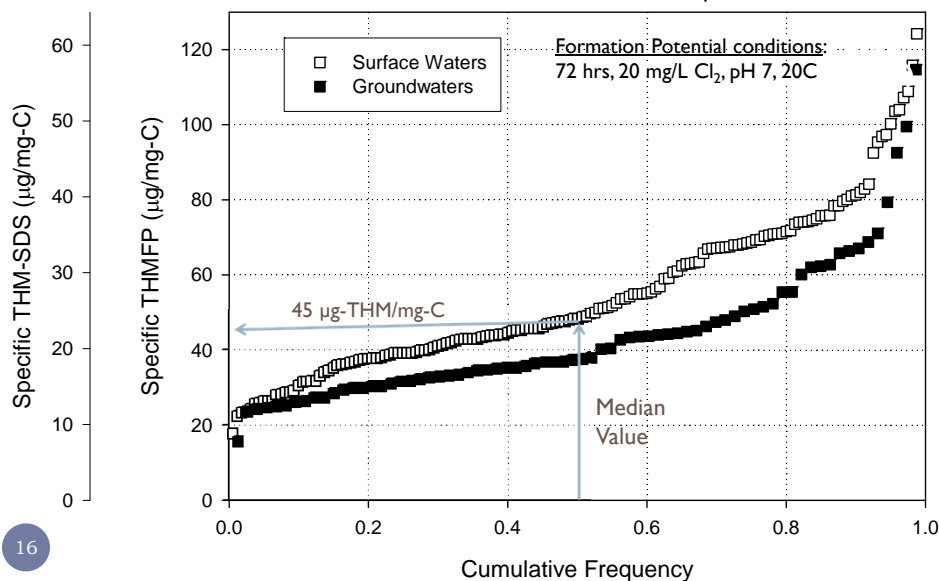
Formation Potential

- ▶ Experiments designed to maximize exposure of water to chlorine (in this case) under optimal conditions and measure the concentration of DBP for a specified duration
- ▶ Disinfection by-product formation potential (DBP-FM):
72 hr, 20 mg/L Cl₂ dose, pH 7, 20C
- ▶ Simulated distribution system (SDS) test: 24 hr, 4 mg/L Cl₂ dose, 20C and pH 7

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THM-FP

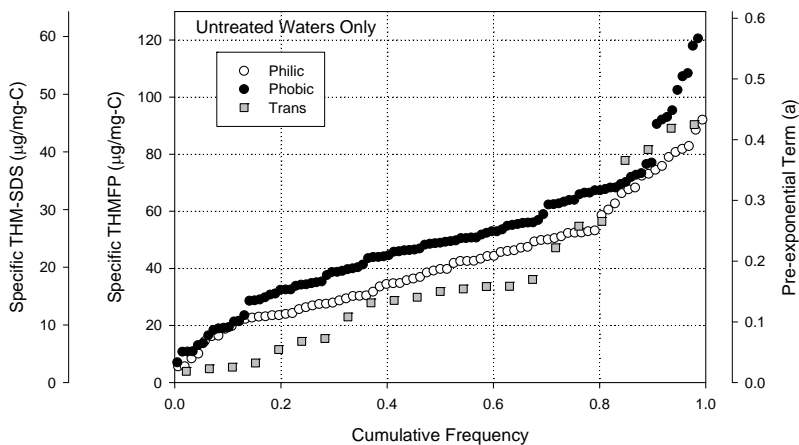
From: Reckhow et al., 2007
WRF Report #91186



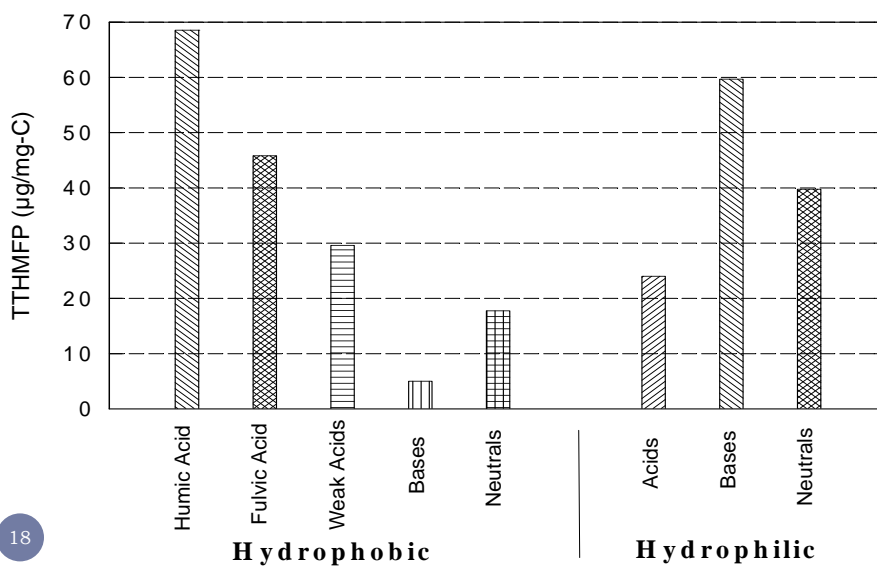
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FP and SDS for NOM Fractions

- ▶ Cumulative Frequency Plot for THM Precursor Content in Major RW Fractions

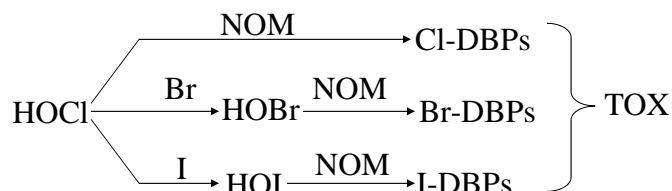


Formation Potentials of NOM Fractions



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TOX Formation



$$\text{TOX} = \text{TOCl} + \text{TOBr} + \text{TOI}$$

Other disinfectants: NH_2Cl , O_3 , ClO_2

From: Guanghui Hua; 2004 WQTC

David Reckhow

What do we know so far?

- ▶ Approximately 50% of the TOX formed by drinking water chlorination is not accounted for → concern about the identity and concentrations of DBPs
- ▶ Not feasible to account for each and every compound that might be formed in disinfected water
- ▶ TOX: A surrogate measure for organically-bound halogenated DBPs in a disinfected water sample.
- ▶ Comparing the TOX values with the halides attributed to the identified DBPs: allow for the estimation of the unidentified TOX
- ▶ TOX analyzers: used to quantify amounts of organically-bound chlorine, bromine and iodine in raw and disinfected water samples

TOX: Known & Unknown

Data from the Mills Plant (CA) August 1997 (courtesy of Stuart Krasner)

But, the Bad Stuff is probably somewhere here?

Unknown Organic Halogen 64%

Haloketones

Chloropicrin

Regulated DBPs

Trihalomethanes 20%

TTHMs

Haloacetonitriles 2%

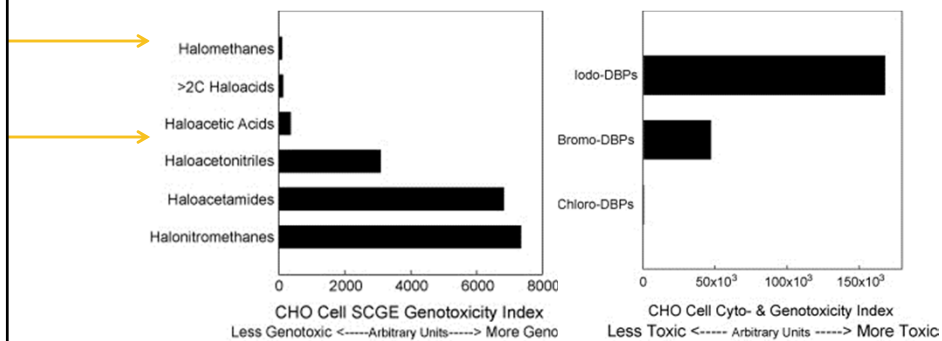
Chloral Hydrate 1%

Sum of 5 Haloacetic Acids 10%

Bromochloroacetic Acid 3%

▶ 21

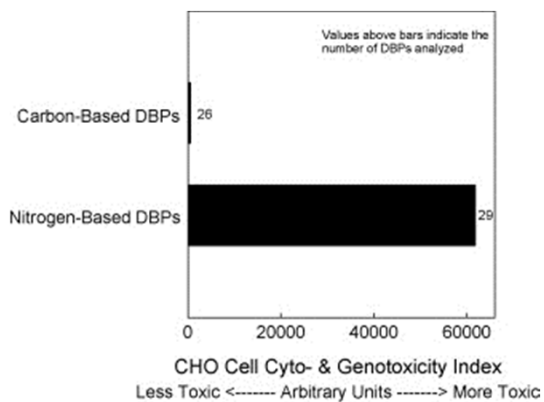
Cyto- and Geno-Toxicity of DBP classes



Occurrence, genotoxicity, and carcinogenicity of regulated and emerging disinfection by-products in drinking water: A review and roadmap for research (Richardson et al., 2007)

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C- and N-based DBPs



Occurrence, genotoxicity, and carcinogenicity of regulated and emerging disinfection by-products in drinking water: A review and roadmap for research
(Richardson et al., 2007)

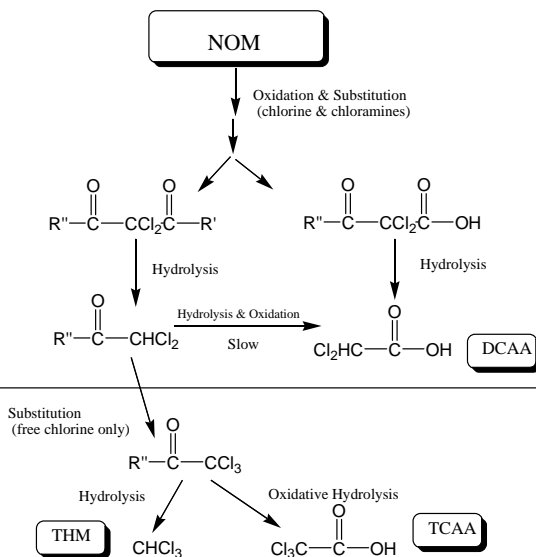
Final disinfectant

- ▶ Drinking water treatment plants usually employ a chemical as a final disinfectant
- ▶ Common oxidative chemicals
 - ▶ Free chlorine
 - ▶ Chloramines
 - ▶ Chlorine dioxide
 - ▶ Manganese oxide
 - ▶ Potassium permanganate

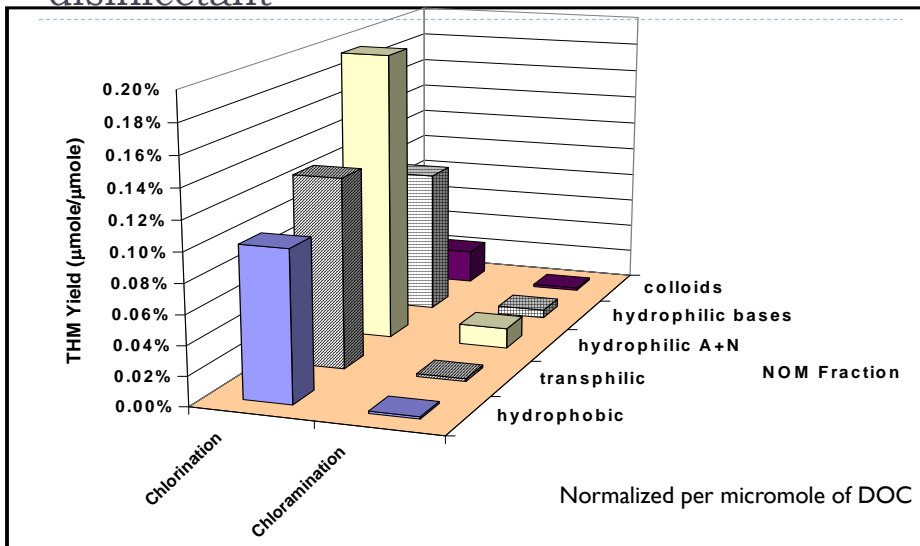
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Use of chloramine vs chlorine as final disinfectant

- ▶ Less formation of regulated DBPs
 - ▶ THMs & HAAs
- ▶ Hydrolysis and oxidation is slow which minimizes further oxidation to TXAA
- ▶ Dihalo products, but little trihalo

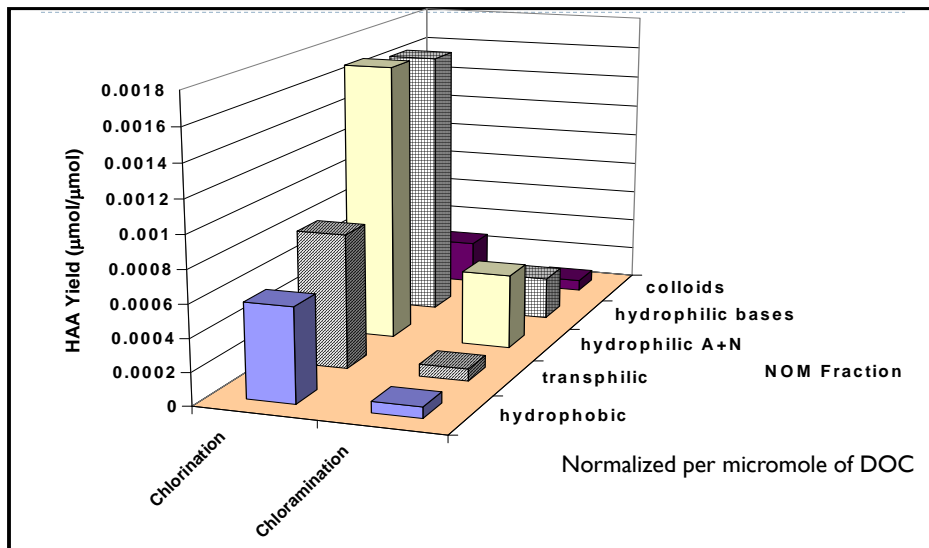


Use of chloramine vs chlorine as final disinfectant



DBP Formation Reactivities of NOM Fractions of a Low-Humic Water[®] by Hwang, Scrimanti & Krasner

Use of chloramine vs chlorine as final disinfectant



▶ DBP Formation Reactivities of NOM Fractions of a Low-Humic Water" by Hwang, Scilimenti & Krasner

Final Thought

- ▶ US federal and state environmental agencies still only regulate four THMs and five HAAs (none of which include iodinated species) in addition to bromate and chlorite.
- ▶ How to change that?
- ▶ Literature is lacking in studies conducted on treated drinking waters that are not spiked with model compounds
 - ▶ → attention should be put in that direction.
- ▶ Focus on quantifying more harmful compounds or TOI/TOBr in drinking water
- ▶ With the recent advances in analytical techniques, it is possible to have data that will supplement existing and ongoing epidemiological/toxicological evidence.
- ▶ Once enough concrete evidence is generated, regulatory agencies will have no choice but to improve on current regulations.

▶

▶ [To next lecture](#)