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CEE 697z
*Organic Compounds in Water and
 Wastewater*

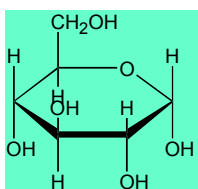
Origins of NOM II

Lecture #5

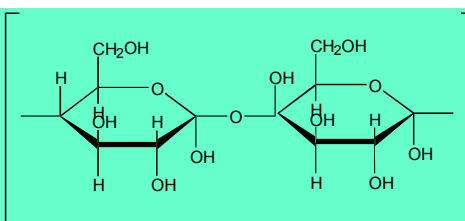
Dave Reckhow - Organics In W & WW

Carbohydrates

- empirical formula: $C_x(H_2O)_y$

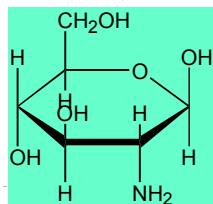


Glucose (monosaccharide)



Cellulose (polysaccharide)

Glucosamine (amino sugar)



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Carbohydrates, cont.

- ▶ **Nomenclature**
 - ▶ Monosaccharide: 1 simple sugar
 - ▶ 1% of DOC
 - ▶ Oligosaccharide: ≤ 10 simple sugars
 - ▶ Polysaccharide: > 10 simple sugars
 - ▶ 5% of DOC
- ▶ **Special interest in distribution systems**
 - ▶ Food for microbial regrowth
 - ▶ Major constituents of:
 - ▶ soluble metabolic byproducts
 - ▶ biofilms



Carbohydrates, cont.

- ▶ **Function in plants**
 - ▶ Structural – cell walls
 - ▶ Cellulose (~10,000 D-glucose units)
 - Most abundant natural organic compound
 - Mostly in higher plants; some algae have none
 - ▶ Hemicelluloses (50-2000 monosaccharides of many types)
 - Forms a matrix around cellulose fibers in cell walls
 - ▶ Chitin (N-acetyl-D-glucosamine units)
 - Second most abundant natural organic (~tied with lignin)
 - Role of cellulose in most fungi, some algae & arthropods
 - ▶ Murein or “peptidoglycan”, a major group of Acylheteropolysaccharides
 - N-acetyl-D-glucosamine & N-acetylmuramic acid cross linked by AA chains
 - Dominant in Eubacteria: up to 75% of bacterial dry mass
 - ▶ Energy – polysaccharides
 - ▶ Starch in plants (80% amylopectin, 20% amylose)
 - ▶ Anti-desiccants



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Carbohydrates, cont.

Algae etc.,
Heteropolysaccharides
Nitrogen-containing

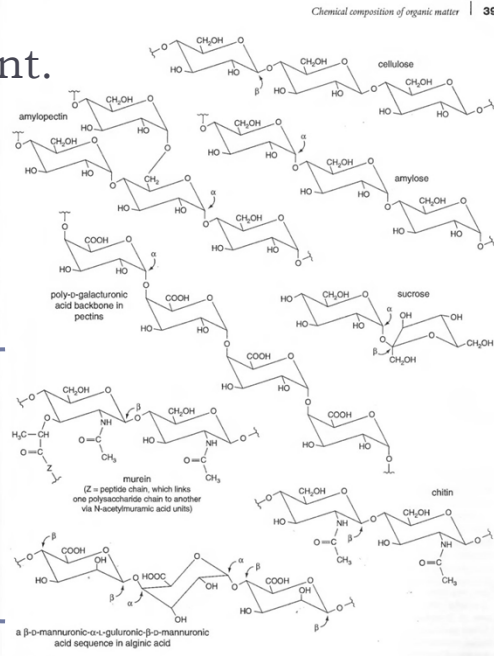


Fig. 2.9 Some important carbohydrates (showing configuration at C-1 in monosaccharide units).

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Carbohydrates, cont.

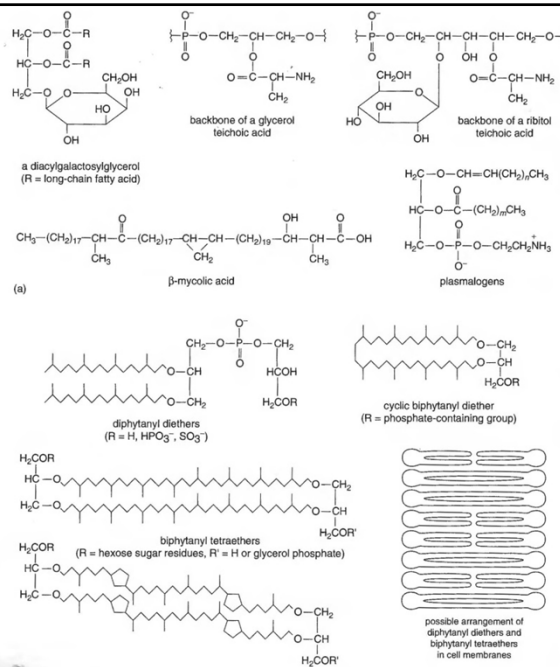


Fig. 2.15 (a) Examples of lipids in the membranes and cell walls of eubacteria. (b) Phytanyl ether lipids in archaebacterial cell membranes.

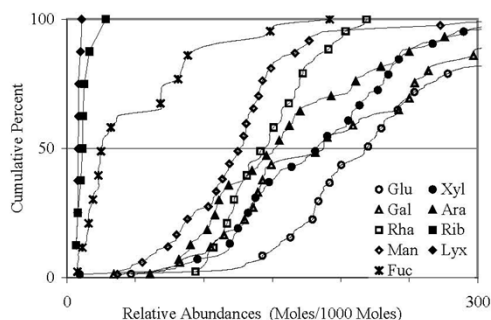
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Acylheteropolysaccharides (APS)

- ▶ 10-35% of river and lake water DOC
- ▶ Produced by algae in fresh and salt waters
- ▶ Similar to structural polysaccharides?
- ▶ Comprised of a nearly fixed ratio of simple sugars, acetate and lipids
- ▶ Refractory like humic substances

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Sugars in Natural Waters



| Sugar | Obs. | Relative Abundances (Moles/1000 Moles) | | | |
|-----------------|------|--|--------|------|-----------|
| | | Range | Median | Mean | Std. Dev. |
| Glucose (Glu) | 71 | 47 - 591 | 220 | 264 | 140 |
| Galactose (Gal) | 85 | 35 - 875 | 188 | 204 | 113 |
| Rhamnose (Rha) | 43 | 94 - 219 | 147 | 147 | 33 |
| Mannose (Man) | 84 | 37 - 357 | 127 | 126 | 51 |
| Fucose (Fuc) | 43 | 8 - 192 | 26 | 50 | 46 |
| Xylose (Xyl) | 84 | 9 - 400 | 183 | 184 | 67 |
| Arabinose (Ara) | 87 | 61 - 455 | 153 | 167 | 77 |
| Ribose (Rib) | 8 | 7 - 28 | 12 | 13 | 7 |
| Lyxose (Lyx) | 8 | 7 - 11 | 8 | 9 | 1 |

| | Obs. | Concentration | | | |
|---|------|---------------|--------|------|-----------|
| | | Range | Median | Mean | Std. Dev. |
| Total Sugars ($\mu\text{mol L}^{-1}$) | 104 | 0.08 - 20.0 | 2.6 | 3.1 | 2.7 |
| % DOC as Sugars | 95 | 0.08 - 35.1 | 3.0 | 7.1 | 8.4 |

From:
Perdue & Ritchie, 2004

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At neutral pH's most lose H⁺

Fatty Acids

• maybe 4% of DOC
 • other mixed acids may account for 2%

$\text{CH}_3\text{-COO}^-$

H-COOH $\text{CH}_3\text{-COOH}$ $\text{CH}_3\text{-CH}_2\text{-COOH}$
 Formic Acid Acetic Acid Propionic Acid

$\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-COOH}$ $\text{H}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-COOH}$
 Butyric Acid Valeric Acid

Common Volatile Fatty Acids in Natural Waters

▶ 9

Amino Acids and Proteins

▶ **Simple Amino Acids**

- ▶ some may form THMs and HANs

$\begin{array}{c} \text{NH}_2 \\ | \\ \text{H}_2\text{C}-\text{C}-\text{COOH} \\ | \\ \text{H} \end{array}$
Alanine

▶ **Proteins**

- ▶ much larger, comprised many AAs

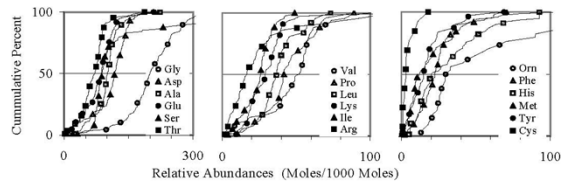
$\begin{array}{c} \text{NH}_2 \\ | \\ \text{HO}-\text{C}_6\text{H}_4-\text{C}-\text{C}-\text{COOH} \\ | \quad | \\ \text{H}_2 \quad \text{H} \end{array}$
Tyrosine

◆ **Special interests in DWT**

- nutrients for bacterial regrowth
- role in chlorine decay and DBP formation

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Amino Acids



| Amino Acid | THAA ^a | TFAA ^a | Relative Abundances (Moles/1000 Moles) | | | |
|---------------------|-------------------|-------------------|--|--------|------|-----------|
| | | | Range | Median | Mean | Std. Dev. |
| Glycine (Gly) | 69 | 19 | 26 - 450 | 200 | 206 | 64 |
| Aspartic (Asp) | 69 | 19 | 20 - 212 | 117 | 113 | 38 |
| Alanine (Ala) | 69 | 18 | 4 - 223 | 102 | 98 | 37 |
| Glutamic (Glu) | 69 | 17 | 14 - 208 | 88 | 84 | 36 |
| Serine (Ser) | 69 | 19 | 31 - 483 | 88 | 126 | 94 |
| Threonine (Thr) | 69 | 19 | 2 - 187 | 73 | 70 | 33 |
| Valine (Val) | 69 | 16 | 2 - 145 | 52 | 49 | 24 |
| Proline (Pro) | 50 | 2 | 10 - 89 | 42 | 45 | 17 |
| Leucine (Leu) | 68 | 19 | 3 - 73 | 37 | 38 | 14 |
| Ornithine (Orn) | 25 | 9 | 13 - 190 | 31 | 55 | 45 |
| Lysine (Lys) | 69 | 9 | 9 - 149 | 29 | 31 | 18 |
| Isoleucine (Ile) | 67 | 19 | 4 - 49 | 26 | 24 | 10 |
| Phenylalanine (Phe) | 68 | 19 | 2 - 70 | 23 | 24 | 14 |
| Histidine (His) | 56 | 18 | 5 - 93 | 20 | 28 | 20 |
| Arginine (Arg) | 52 | 15 | 2 - 117 | 17 | 26 | 25 |
| Tyrosine (Tyr) | 61 | 19 | 0.5 - 69 | 15 | 17 | 14 |
| Methionine (Met) | 60 | 16 | 0.2 - 108 | 12 | 21 | 20 |
| Cysteine (Cys) | 13 | 0 | 1 - 18 | 3 | 5 | 5 |

From:
Perdue & Ritchie, 2004

| | Obs. | Range | Concentration | | |
|---------------------------------|------|-------------|---------------|------|-----------|
| | | | Median | Mean | Std. Dev. |
| THAA ($\mu\text{mol L}^{-1}$) | 51 | 0.12 - 23.2 | 1.3 | 4.1 | 6.2 |
| TFAA ($\mu\text{mol L}^{-1}$) | 21 | 0.05 - 1.8 | 0.3 | 0.6 | 0.5 |
| % DOC as THAA | 59 | 0.42 - 10.4 | 1.8 | 2.2 | 1.8 |
| % DOC as TFAA | 14 | 0.02 - 1.2 | 0.1 | 0.3 | 0.4 |

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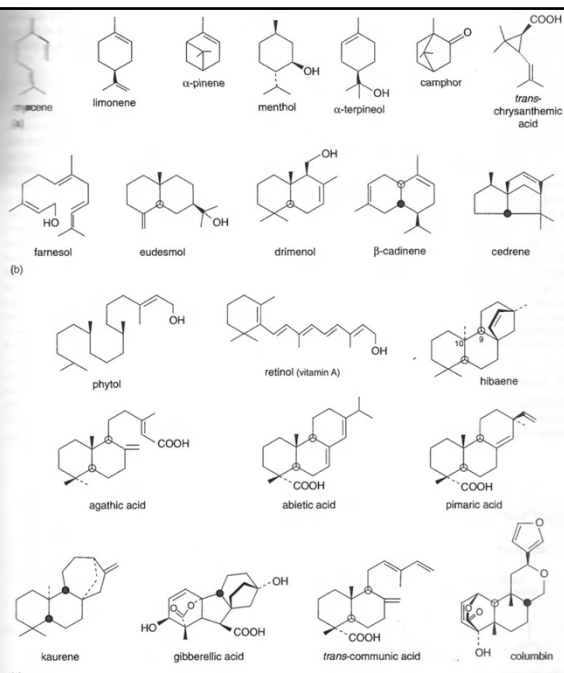
Terpenes and Terpenoids

- The **terpenoids**, sometimes called **isoprenoids**, are a large and diverse class of naturally occurring organics similar to [terpenes](#), derived from five-carbon [isoprene](#) units assembled and modified in thousands of ways.
 - Terpenoids can be thought of as modified terpenes, wherein methyl groups have been moved or removed, or oxygen atoms added.
- Plant terpenoids are used extensively for their aromatic qualities. They play a role in traditional herbal remedies and are under investigation for [antibacterial](#), [antineoplastic](#), and other [pharmaceutical](#) functions. Terpenoids contribute to the scent of [eucalyptus](#), the flavors of [cinnamon](#), [cloves](#), and [ginger](#), the yellow color in [sunflowers](#), and the red color in [tomatoes](#).
- Terpenoids can be classified according to the number of isoprene units used:
 - [Hemiterpenoids](#), 1 isoprene unit (5 carbons)
 - [Monoterpenoids](#), 2 isoprene units (10C)
 - [Sesquiterpenoids](#), 3 isoprene units (15C)
 - [Diterpenoids](#), 4 isoprene units (20C) (e.g. [ginkgolides](#))
 - [Sesterterpenoids](#), 5 isoprene units (25C)
 - [Triterpenoids](#), 6 isoprene units (30C) (e.g. [sterols](#))
 - [Tetraterpenoids](#), 8 isoprene units (40C) (e.g. [carotenoids](#))
 - [Polyterpenoid](#) with a larger number of isoprene units

From Wikipedia

► 12

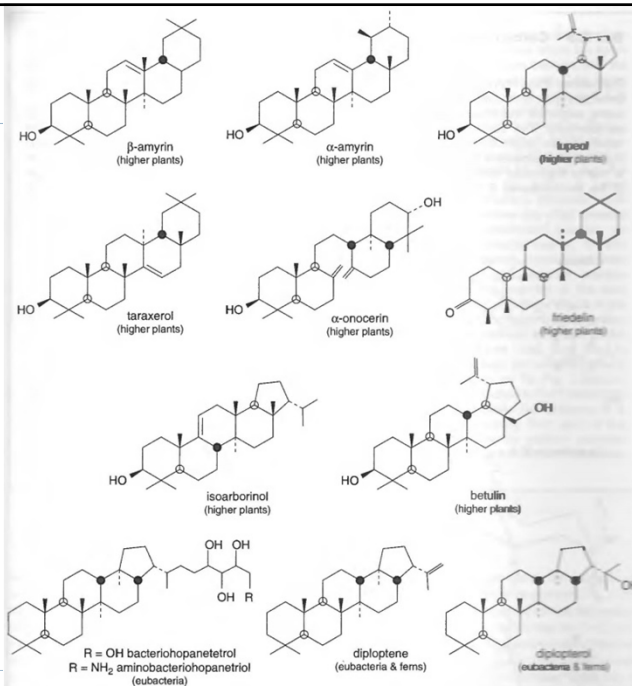
Terpenoids



▶ 13

Fig. 2.17 Examples of (a) monoterpenoids, (b) sesquiterpenoids and (c) diterpenoids.

Terpenoids, cont.



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Fig. 2.19 Some geochemically important polycyclic triterpenoids and their major sources.

Iron Complexation

Iron is a vital component of the proteins involved in photosynthesis, respiration and nitrogen fixation, but under oxic conditions it is predominantly found in its extremely low solubility Fe(III) form and so is not readily available for uptake by organisms. Unlike anoxic environments, in which soluble Fe(II) is generally abundant, the well oxygenated surface waters of the open oceans can contain as little as 10^{-8}g l^{-1} of

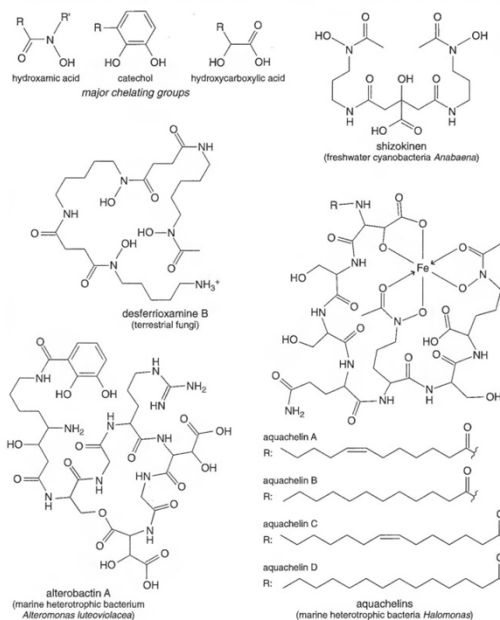


Fig. 3.11 Major iron(III)-chelating groups in siderophores and structures of some siderophores.

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Van Krevlin Plot

Table 4.5 Coal analysis (ASTM procedures; after Ward 1964)

| | |
|-----------------------------------|---|
| proximate analyses | |
| moisture | drying at 110 °C (avoiding oxidation and decomposition) |
| volatile matter | volatiles liberated at 950 °C in absence of air, excluding moisture |
| ash | inorganic residue from combustion |
| fixed carbon | C remaining after volatiles determination (i.e. coking potential) |
| ultimate analyses | |
| elemental analysis | C, H, N and S content |
| oxygen | determined by difference (i.e. total minus C, H, N, S content, and correcting for inorganic mineral content); more rarely determined directly |
| examples of other analyses | |
| forms of sulphur | organic, sulphide, native S, sulphate |
| other elements | e.g. trace elements, phosphorus, chlorine, carbonate CO ₂ |
| relative density | depends on ash content and maturity |
| specific energy | energy liberated upon combustion |

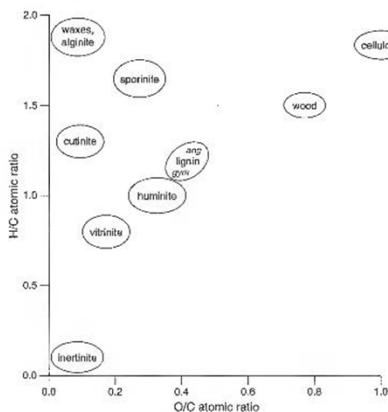
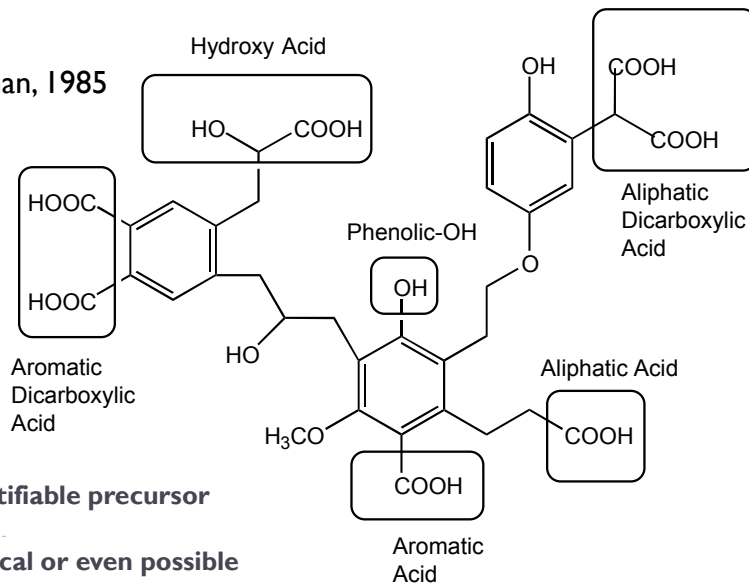


Fig. 4.3 Chemical composition of major coal macerals and plant tissues plotted on a van Krevlin diagram (after Tissot & Welte 1984; Hedges et al. 1985). Lignin incorporates differences between angiosperms (ang) and gymnosperms (gym) (after Hatcher 1990).

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Putting it all together?

From Thurman, 1985



- ▶ Many identifiable precursor structures
- ▶ Not practical or even possible

Concentrations: Pedogenic

- ▶ Land Sources
 - ▶ From Woody & non-woody plants, lignin, etc.
 - ▶ Depends on vegetation, soil, hydrology
- ▶ Attenuated by adsorption to clay soils
 - ▶ Parallel watersheds in Australia (Cotsaris et al., 1994 [Chamonix proceedings])
 - ▶ Clearwater Creek, high clay content: 2.5 mg/L TOC
 - ▶ Redwater Creek, sandy soil: 31.7 mg/L TOC

Concentrations: Aquagenic

- ▶ Algal & aquatic plant Sources
 - ▶ Depend on nutrient levels / trophic state
- ▶ Concentrations in Lakes (mg/L) (Thurman, 1985)

| Trophic State | Mean DOC | Range |
|---------------|----------|-------|
| Oligotrophic | 2 | 1-3 |
| Mesotrophic | 3 | 2-4 |
| Eutrophic | 10 | 3-34 |
| Dystrophic | 30 | 20-50 |

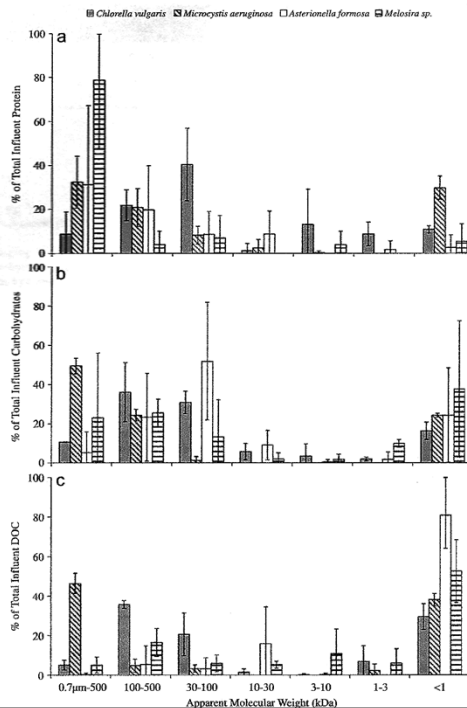
- ▶ Groundwater
 - ▶ No algae

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MW vs type

- ▶ Algogenic organic matter (AOM)
 - ▶ Proteins & carbohydrates
 - ▶ Large polymers with monomers

Henderson et al., 2008

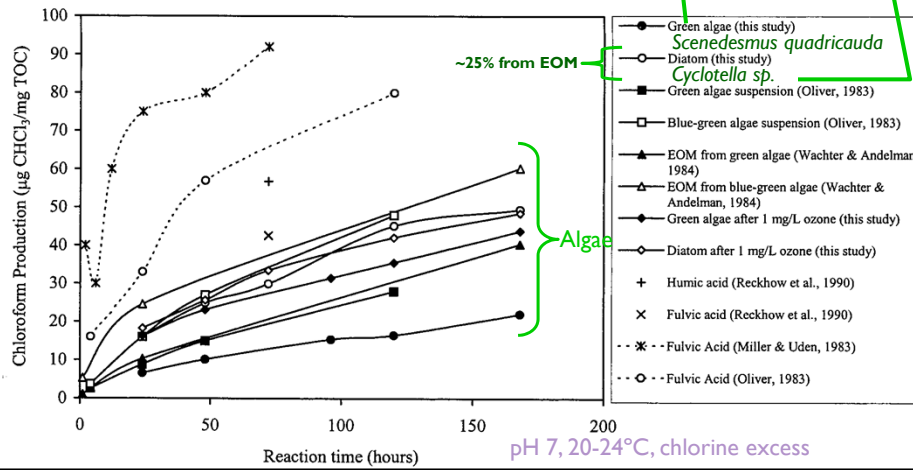
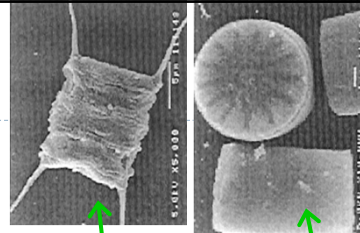


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Algae as THM Precursors

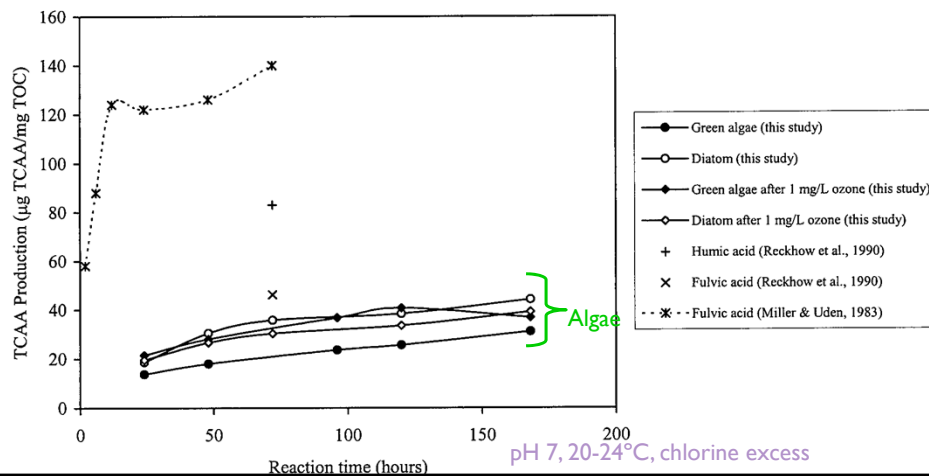
► From: Plummer & Edzwald, 2001

► [ES&T:35:3661]



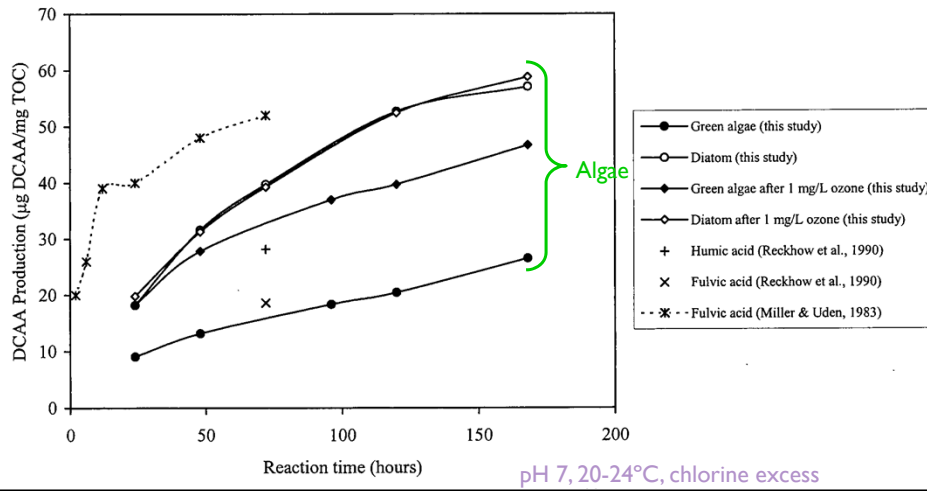
Algae as TCAA Precursors

► Not much impact?



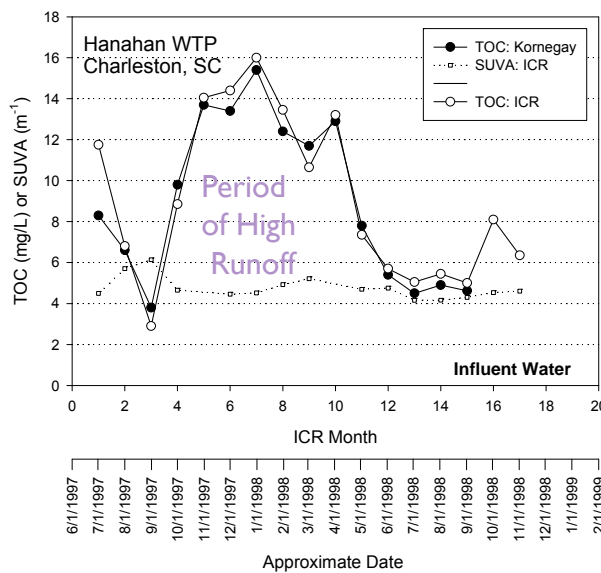
Algae as DCAA Precursors

▶ Are Algae important sources of dihalo-AA precursors?



Annual TOC Cycles

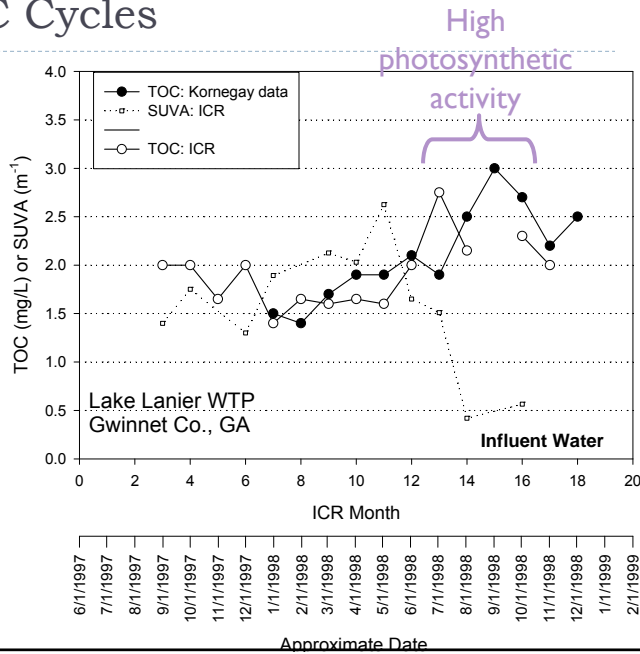
- ▶ Edisto River
 - ▶ Former source for Charleston's (SC) Hanahan WTP
 - ▶ Flushing of TOC during high rainfall months (cold period)



Annual TOC Cycles

▶ Lake Lanier

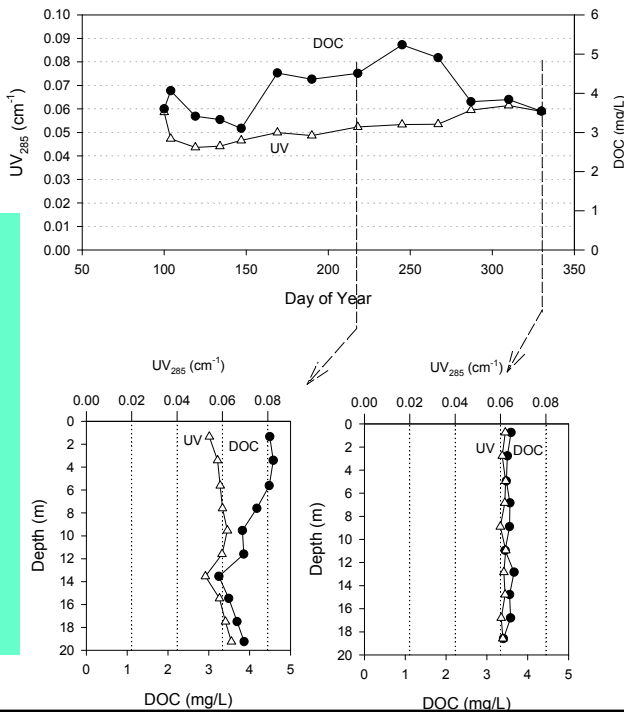
- ▶ Source for Gwinnett Co.'s (GA) Lanier WTP
- ▶ High clay content in watershed



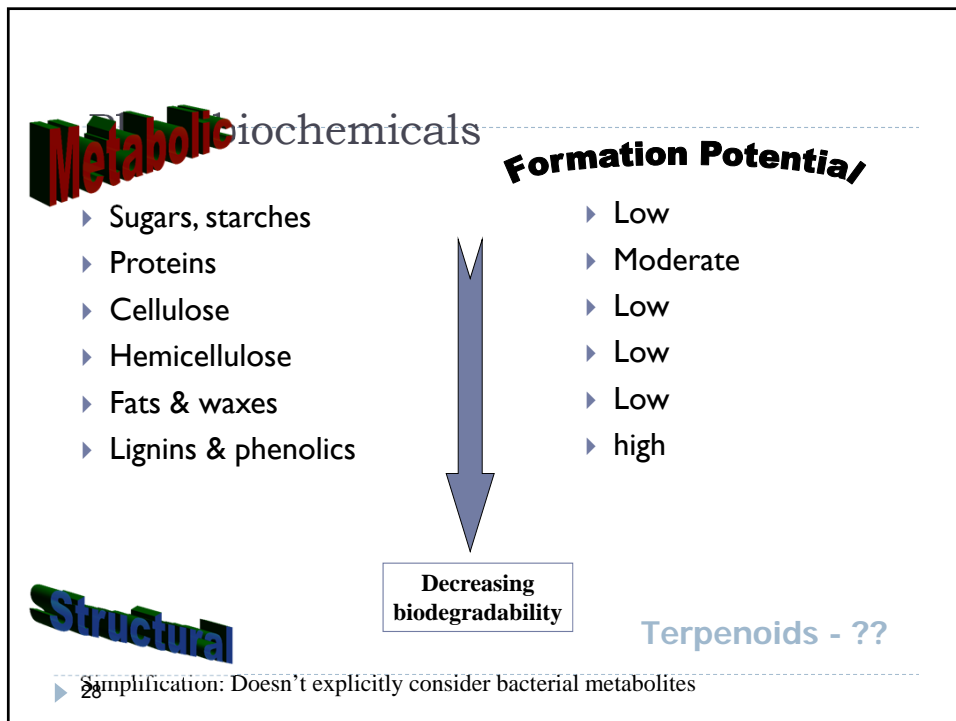
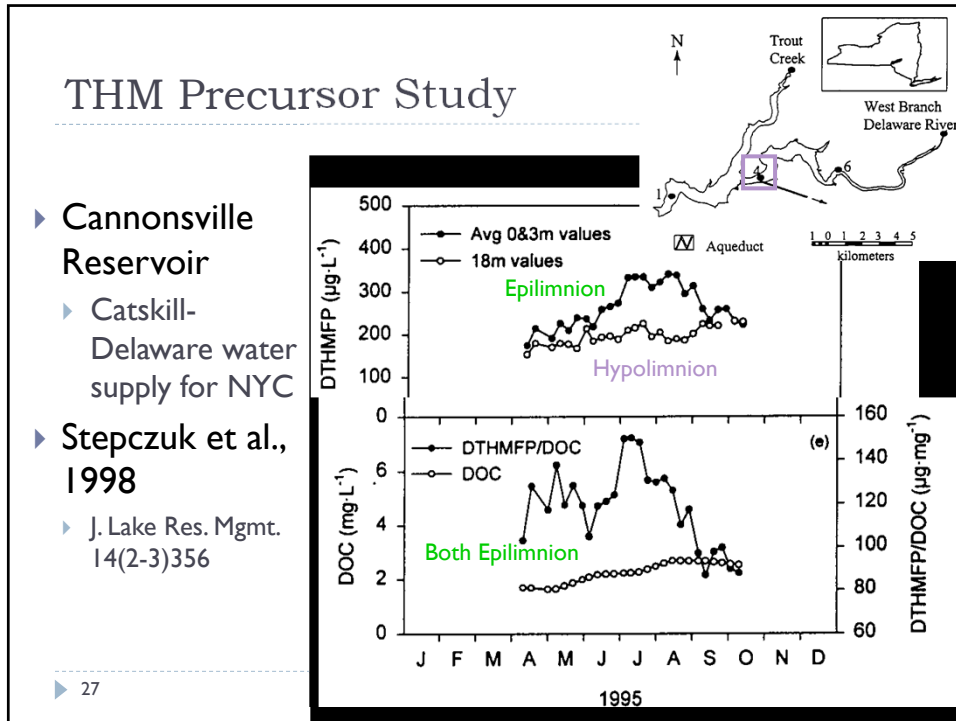
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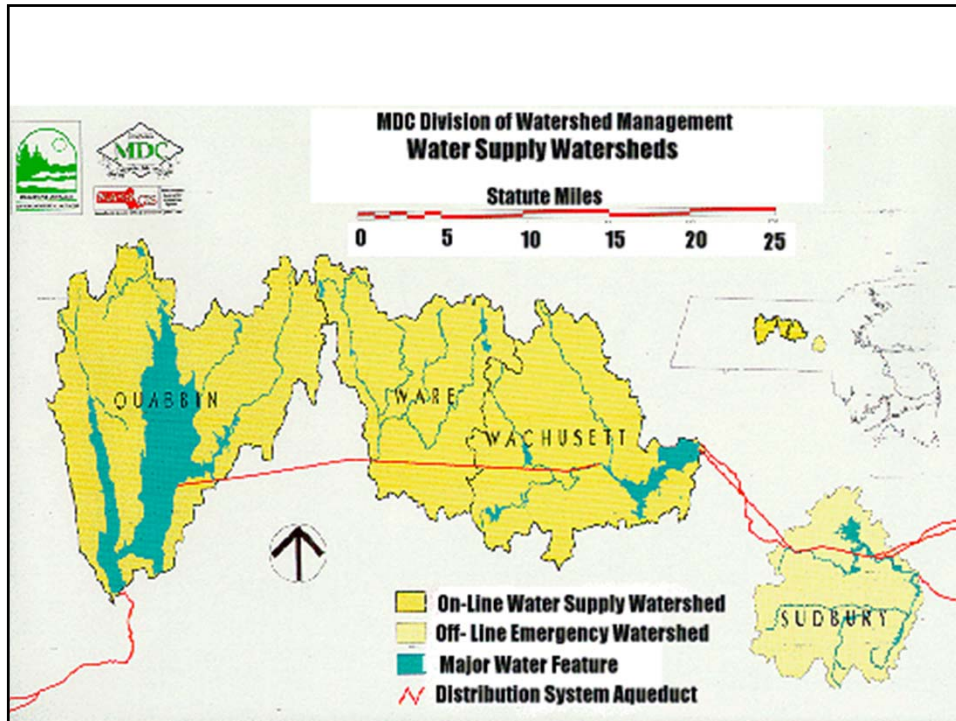
Spatial and Temporal Distribution of DOC and UV absorbing Substances in Lake Bret

(from Zumstein & Buffle, 1989; and Krasner et al., 1996)



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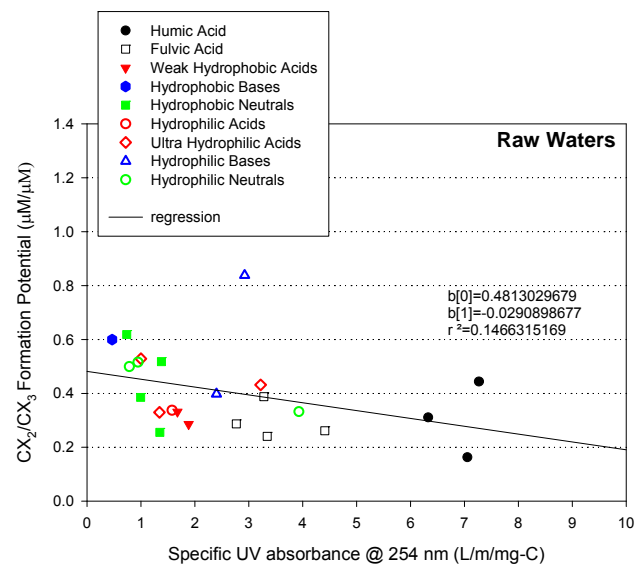




Dihalo and Trihalo DBPs

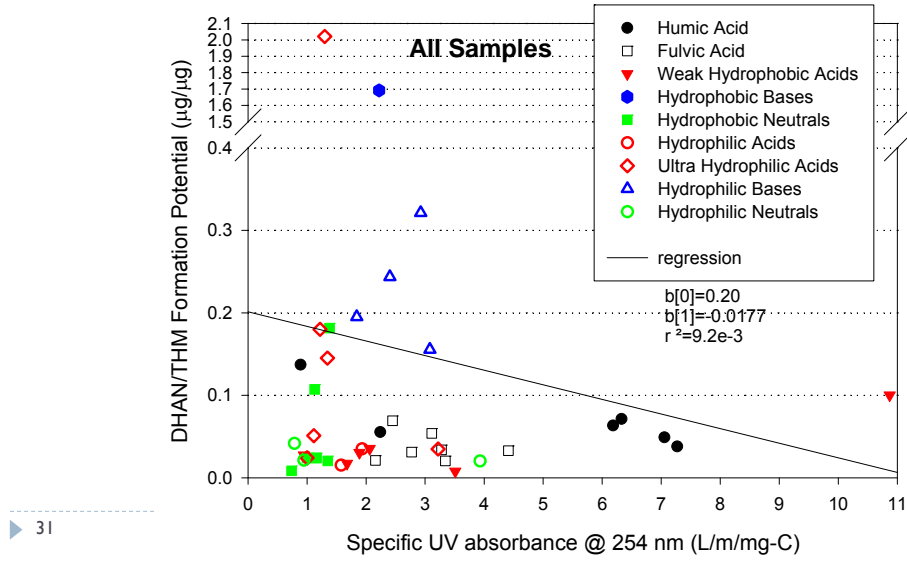
► NOM Fractions

- Evidence for greater importance of dihalo species in non-lignin based NOM



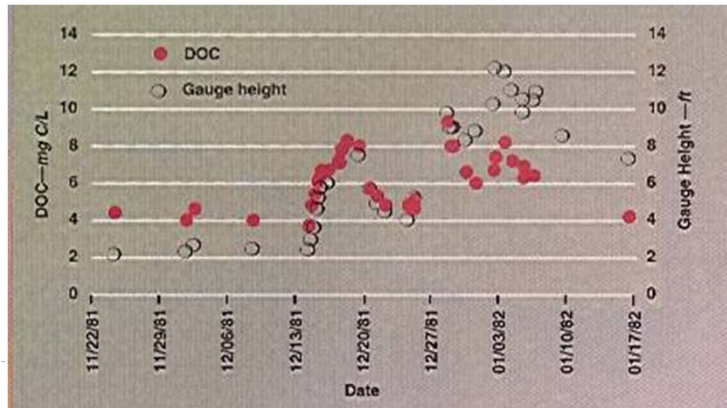
► 30

DHAN/THM Ratio vs SUVA

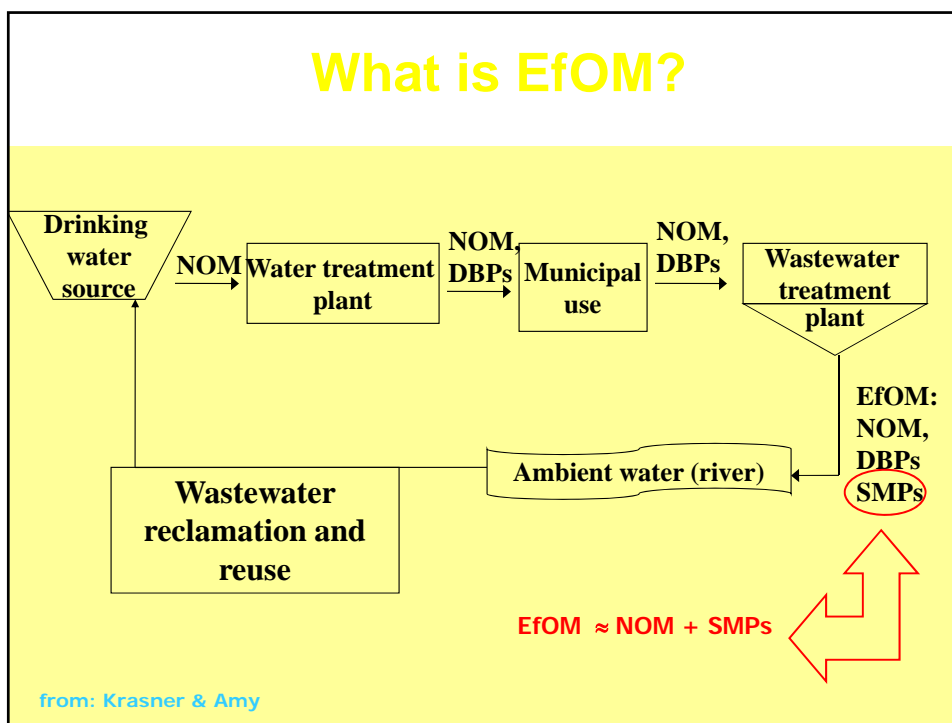


DOC and runoff

- ▶ Ogeechee River (GA)
 - ▶ From Aiken & Cotsaris, 1995
 - [JAWWA 87(1)36]



What is EfOM?



▶ [To next lecture](#)