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**CEE 697z**

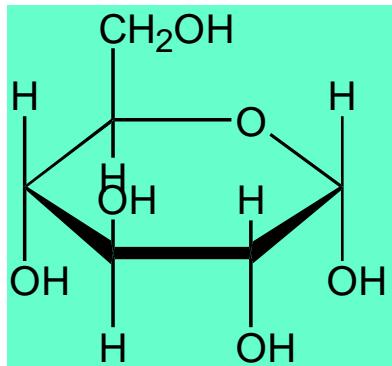
*Organic Compounds in Water and  
Wastewater*

Origins of NOM II

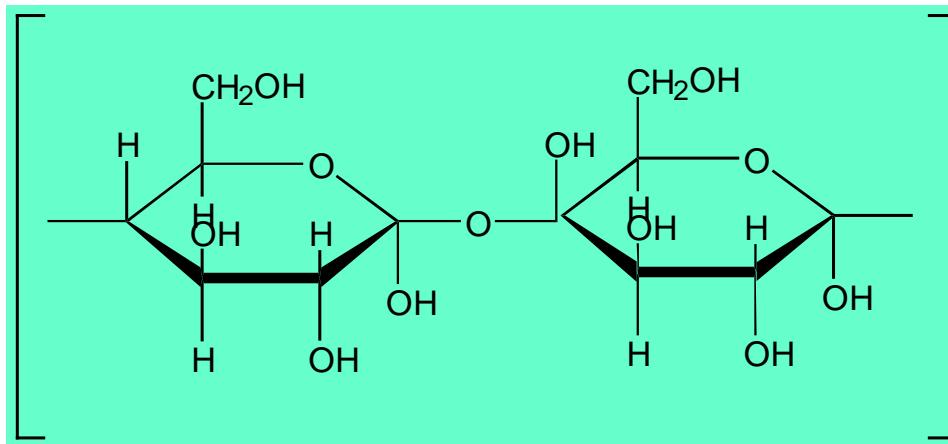
Lecture #5

# Carbohydrates

- empirical formula:  $C_x(H_2O)_y$

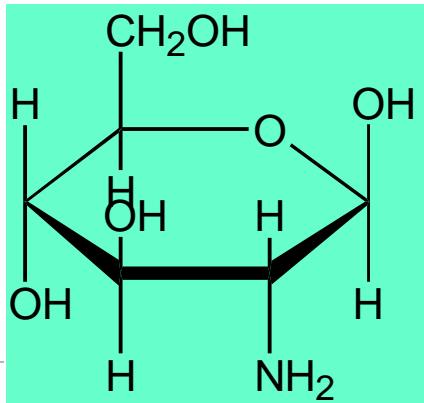


Glucose (monosaccharide)



Cellulose (polysaccharide)

Glucosamine (amino sugar)



# Carbohydrates, cont.

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- ▶ **Nomenclature**
  - ▶ Monosaccharide: 1 simple sugar
    - ▶ 1% of DOC
  - ▶ Oligosaccharide:  $\leq 10$  simple sugars
  - ▶ Polysaccharide:  $> 10$  simple sugars
    - ▶ 5% of DOC
- ▶ **Special interest in distribution systems**
  - ▶ Food for microbial regrowth
  - ▶ Major constituents of:
    - ▶ soluble metabolic byproducts
    - ▶ biofilms



# Carbohydrates, cont.

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- ▶ Function in plants
  - ▶ Structural – cell walls
    - ▶ Cellulose (~10,000 D-glucose units)
      - Most abundant natural organic compound
      - Mostly in higher plants; some algae have none
    - ▶ Hemicelluloses (50-2000 monosaccharides of many types)
      - Forms a matrix around cellulose fibers in cell walls
    - ▶ Chitin (N-acetyl-D-glucosamine units)
      - Second most abundant natural organic (~tied with lignin)
      - Role of cellulose in most fungi, some algae & arthropods
    - ▶ Murein or “peptidoglycan”, a major group of Acylheteropolysaccharides
      - N-acetyl-D-glucosamine & N-acetylmuramic acid cross linked by AA chains
      - Dominant in Eubacteria: up to 75% of bacterial dry mass
  - ▶ Energy – polysaccharides
    - ▶ Starch in plants (80% amylopectin, 20% amylose)
  - ▶ Anti-dessicants

# Carbohydrates, cont.

Algae etc.,  
Heteropolysaccharides  
Nitrogen-containing

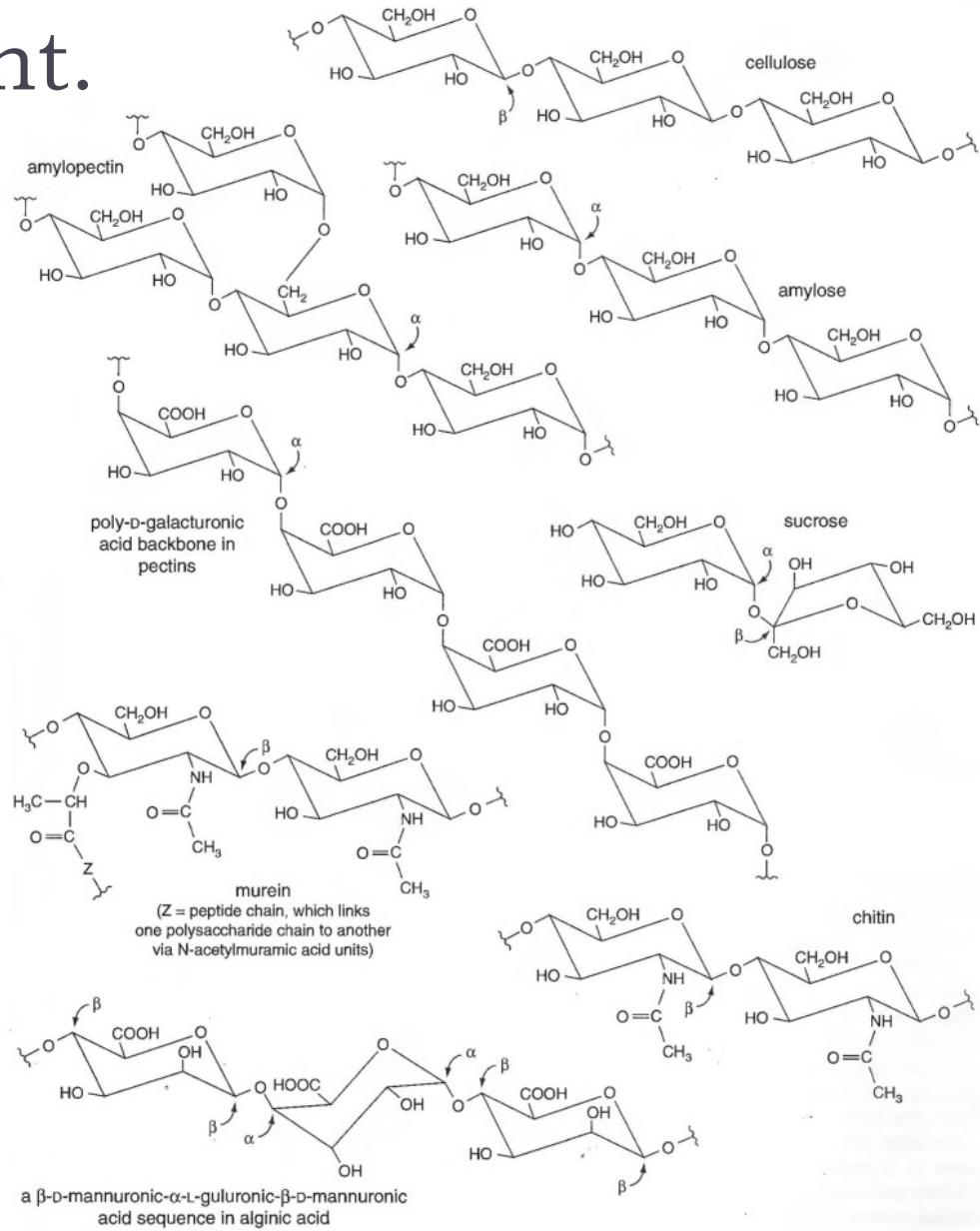
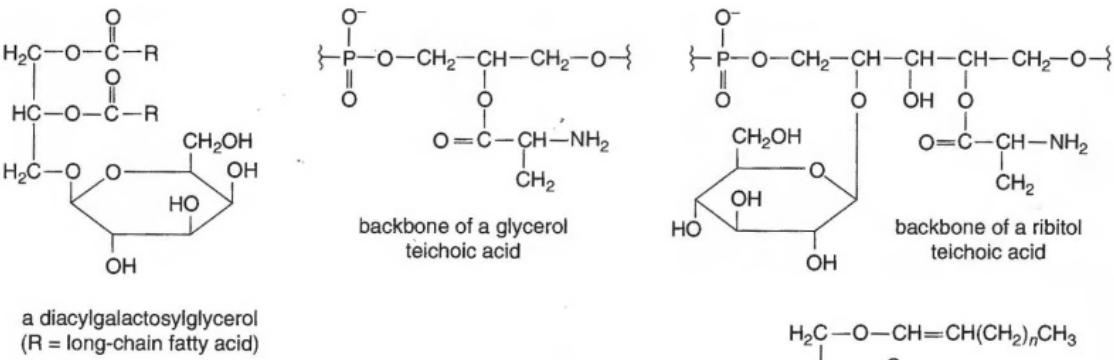


Fig. 2.9 Some important carbohydrates (showing configuration at C-1 in monosaccharide units).

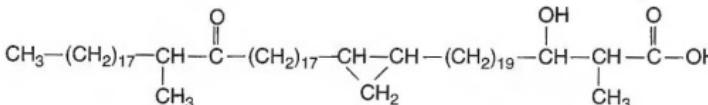
# Carbohydrates, cont.



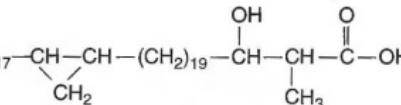
a diacylgalactosylglycerol  
(R = long-chain fatty acid)

backbone of a glycerol teichoic acid

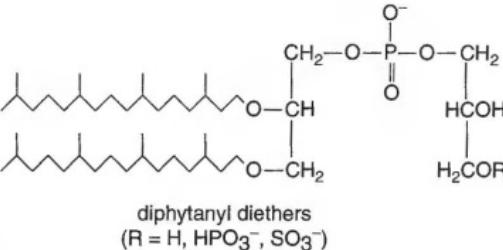
backbone of a ribitol teichoic acid



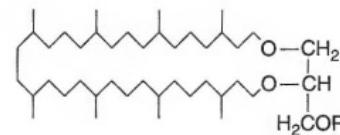
(a)



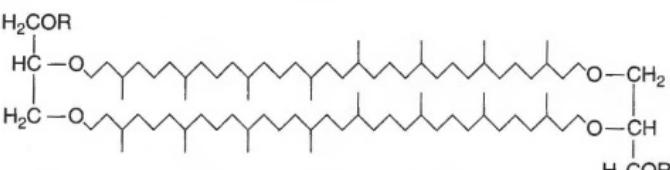
plasmalogens



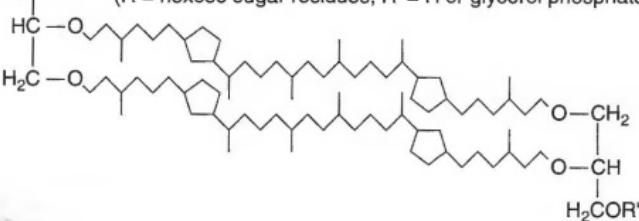
diphytanyl diethers  
(R = H,  $\text{HPO}_3^{2-}$ ,  $\text{SO}_3^{2-}$ )



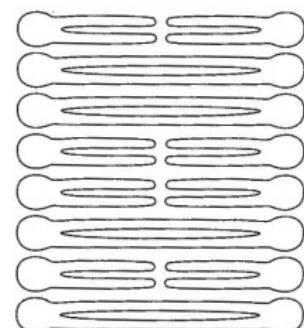
cyclic biphytanyl diether  
(R = phosphate-containing group)



biphytanyl tetraethers  
(R = hexose sugar residues, R' = H or glycerol phosphate)



(b)



possible arrangement of  
diphytanyl diethers and  
biphytanyl tetraethers  
in cell membranes

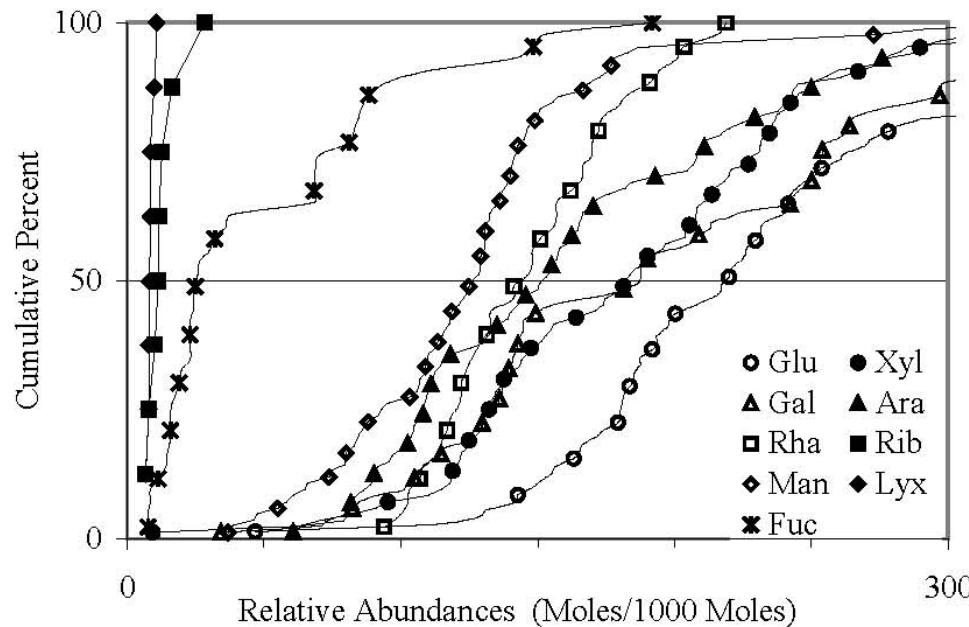
**Fig. 2.15** (a) Examples of lipids in the membranes and cell walls of eubacteria. (b) Phytanyl ether lipids in archaeabacterial cell membranes.

# Acylheteropolysaccharides (APS)

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- ▶ 10-35% of river and lake water DOC
- ▶ Produced by algae in fresh and salt waters
- ▶ Similar to structural polysaccharides?
- ▶ Comprised of a nearly fixed ratio of simple sugars, acetate and lipids
- ▶ Refractory like humic substances

# Sugars in Natural Waters



Sugar	Obs.	Relative Abundances (Moles/1000 Moles)			
		Range	Median	Mean	Std. Dev.
Glucose (Glu)	71	47 - 591	220	264	140
Galactose (Gal)	85	35 - 875	188	204	113
Rhamnose (Rha)	43	94 - 219	147	147	33
Mannose (Man)	84	37 - 357	127	126	51
Fucose (Fuc)	43	8 - 192	26	50	46
Xylose (Xyl)	84	9 - 400	183	184	67
Arabinose (Ara)	87	61 - 455	153	167	77
Ribose (Rib)	8	7 - 28	12	13	7
Lyxose (Lyx)	8	7 - 11	8	9	1

From:  
Perdue & Ritchie, 2004

	Obs.	Concentration			
		Range	Median	Mean	Std. Dev.
Total Sugars ( $\mu\text{mol L}^{-1}$ )	104	0.08 - 20.0	2.6	3.1	2.7
% DOC as Sugars	95	0.08 - 35.1	3.0	7.1	8.4

At neutral pH's most lose H<sup>+</sup>



## Fatty Acids

- maybe 4% of DOC
- other mixed acids may account for 2%



Formic Acid



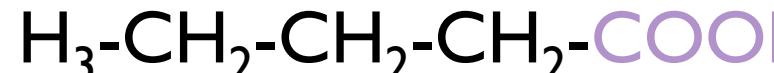
Acetic Acid



Propionic Acid



Butyric Acid



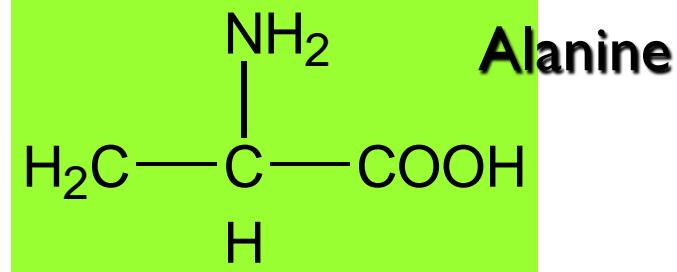
Valeric Acid

*Common Volatile Fatty Acids in Natural Waters*

# Amino Acids and Proteins

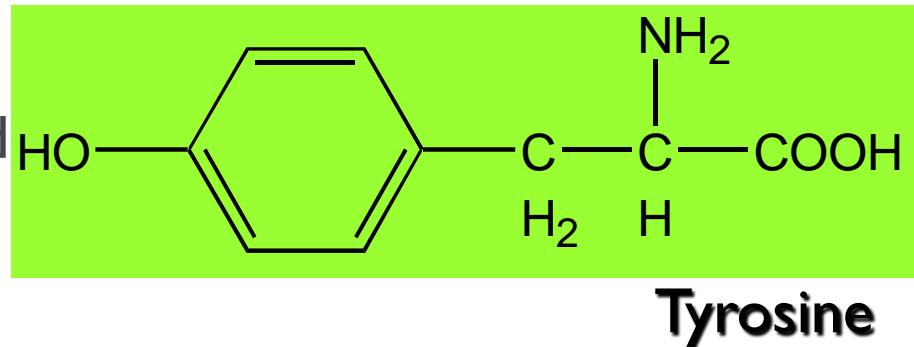
- ▶ Simple Amino Acids

- ▶ some may form THMs and HANs



- ▶ Proteins

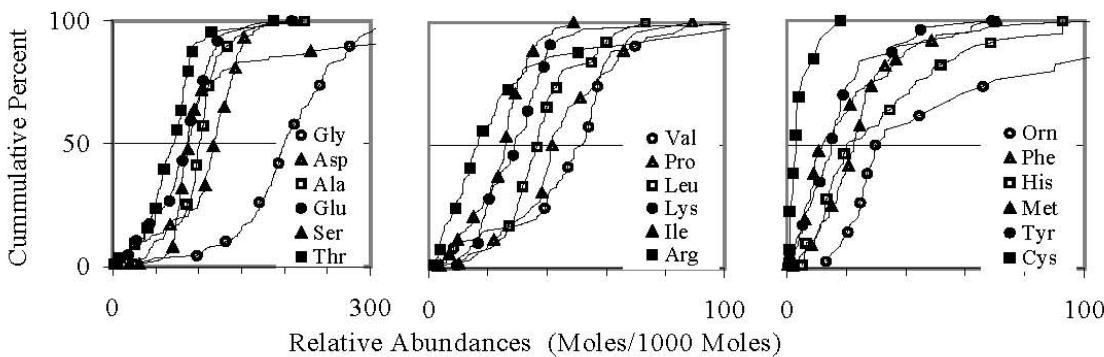
- ▶ much larger, comprised of many AAs



- ◆ Special interests in DWT

- nutrients for bacterial regrowth
  - role in chlorine decay and DBP formation

# Amino Acids



Amino Acid	Relative Abundances (Moles/1000 Moles)					
	THAA <sup>a</sup>	TFAA <sup>a</sup>	Range	Median	Mean	Std. Dev.
Glycine (Glu)	69	19	26 - 450	200	206	64
Aspartic (Asp)	69	19	20 - 212	117	113	38
Alanine (Ala)	69	18	4 - 223	102	98	37
Glutamic (Glu)	69	17	14 - 208	88	84	36
Serine (Ser)	69	19	31 - 483	88	126	94
Threonine (Thr)	69	19	2 - 187	73	70	33
Valine (Val)	69	16	2 - 145	52	49	24
Proline (Pro)	50	2	10 - 89	42	45	17
Leucine (Leu)	68	19	3 - 73	37	38	14
Ornithine (Orn)	25	9	13 - 190	31	55	45
Lysine (Lys)	69	9	9 - 149	29	31	18
Isoleucine (Ile)	67	19	4 - 49	26	24	10
Phenylalanine (Phe)	68	19	2 - 70	23	24	14
Histidine (His)	56	18	5 - 93	20	28	20
Arginine (Arg)	52	15	2 - 117	17	26	25
Tyrosine (Tyr)	61	19	0.5 - 69	15	17	14
Methionine (Met)	60	16	0.2 - 108	12	21	20
Cysteine (Cys)	13	0	1 - 18	3	5	5

From:  
Perdue & Ritchie, 2004

	Obs.	Range	Concentration		
			Median	Mean	Std. Dev.
THAA ( $\mu\text{mol L}^{-1}$ )	51	0.12 - 23.2	1.3	4.1	6.2
TFAA ( $\mu\text{mol L}^{-1}$ )	21	0.05 - 1.8	0.3	0.6	0.5
% DOC as THAA	59	0.42 - 10.4	1.8	2.2	1.8
% DOC as TFAA	14	0.02 - 1.2	0.1	0.3	0.4

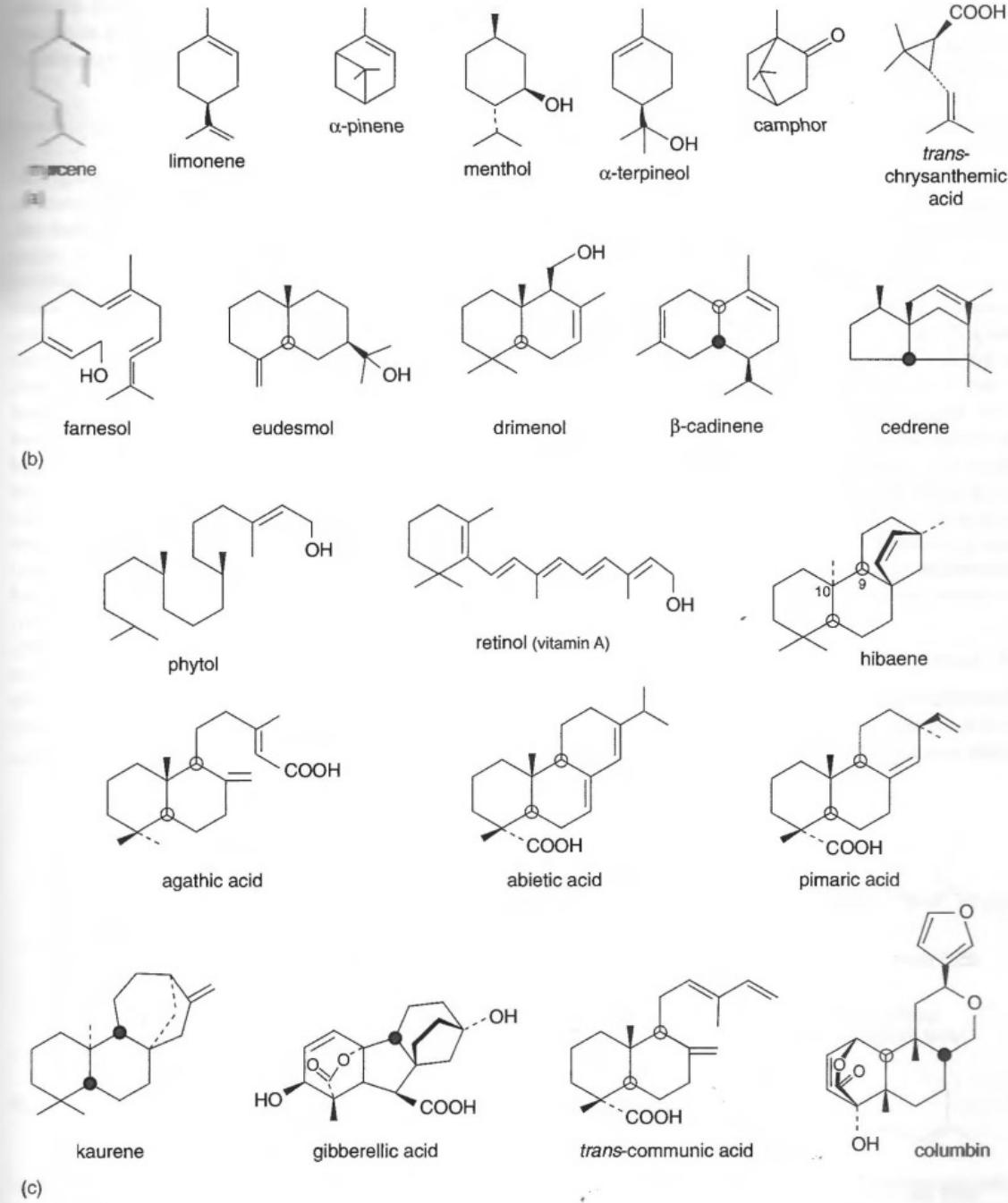
# Terpenes and Terpenoids

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- ▶ The **terpenoids**, sometimes called **isoprenoids**, are a large and diverse class of naturally occurring organics similar to [terpenes](#), derived from five-carbon [isoprene](#) units assembled and modified in thousands of ways.
  - ▶ Terpenoids can be thought of as modified terpenes, wherein methyl groups have been moved or removed, or oxygen atoms added.
- ▶ Plant terpenoids are used extensively for their aromatic qualities. They play a role in traditional herbal remedies and are under investigation for [antibacterial](#), [antineoplastic](#), and other [pharmaceutical](#) functions. Terpenoids contribute to the scent of [eucalyptus](#), the flavors of [cinnamon](#), [cloves](#), and [ginger](#), the yellow color in [sunflowers](#), and the red color in [tomatoes](#).
- ▶ Terpenoids can be classified according to the number of isoprene units used:
  - ▶ [Hemiterpenoids](#), 1 isoprene unit (5 carbons)
  - ▶ [Monoterpenoids](#), 2 isoprene units (10C)
  - ▶ [Sesquiterpenoids](#), 3 isoprene units (15C)
  - ▶ [Diterpenoids](#), 4 isoprene units (20C) (e.g. [ginkgolides](#))
  - ▶ [Sesterterpenoids](#), 5 isoprene units (25C)
  - ▶ [Triterpenoids](#), 6 isoprene units (30C) (e.g. [sterols](#))
  - ▶ [Tetraterpenoids](#), 8 isoprene units (40C) (e.g. [carotenoids](#))
  - ▶ [Polyterpenoid](#) with a larger number of isoprene units

From Wikipedia

# Terpenoids



**Fig. 2.17** Examples of (a) monoterpenoids, (b) sesquiterpenoids and (c) diterpenoids.

# Terpenoids, cont.

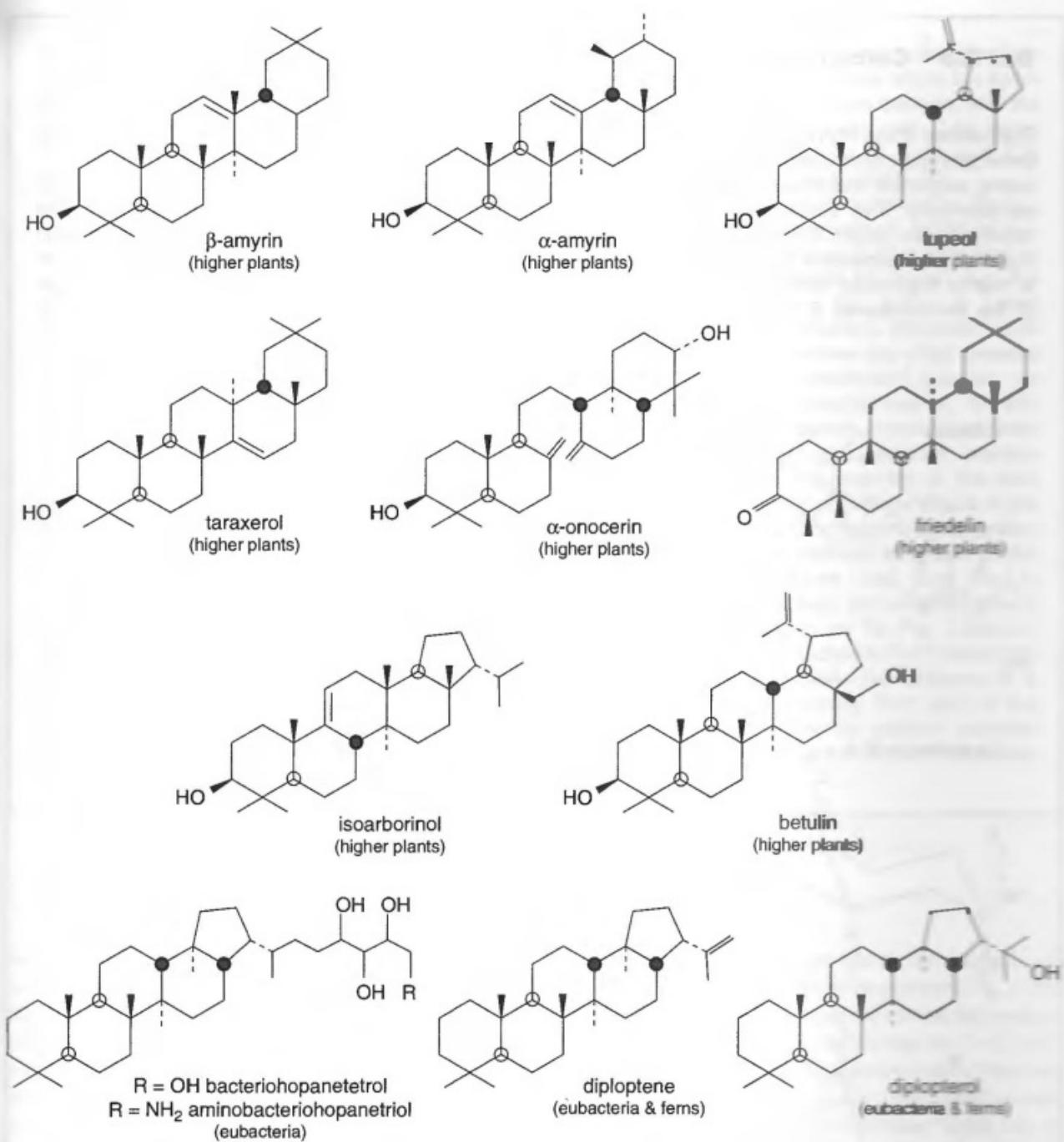
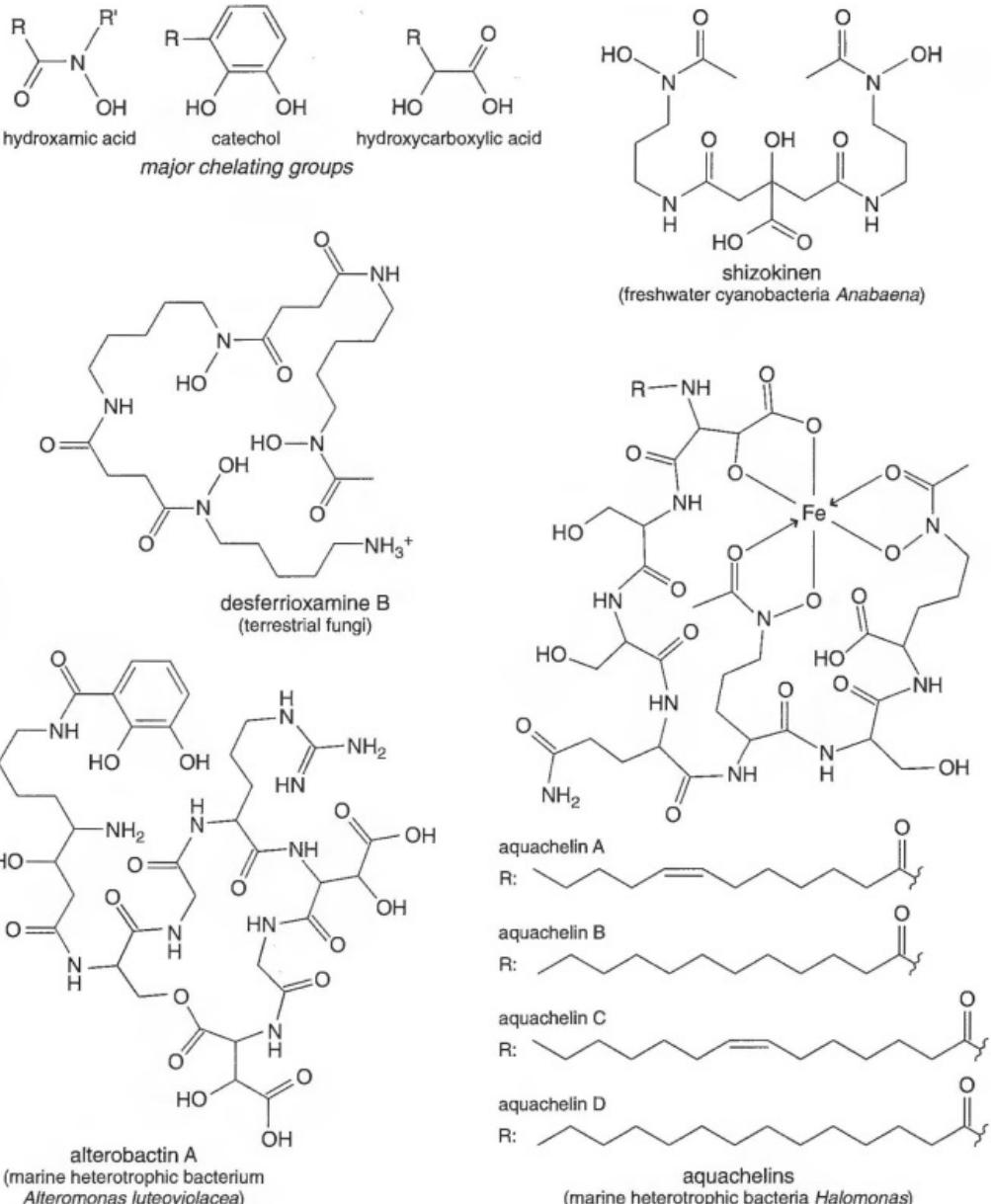


Fig. 2.19 Some geochemically important polycyclic triterpenoids and their major sources.

# Iron Complexation

Iron is a vital component of the proteins involved in photosynthesis, respiration and nitrogen fixation, but under oxic conditions it is predominantly found in its extremely low solubility Fe(III) form and so is not

readily available for uptake by organisms. Unlike anoxic environments, in which soluble Fe(II) is generally abundant, the well oxygenated surface waters of the open oceans can contain as little as  $10^{-8}\text{ g l}^{-1}$  of

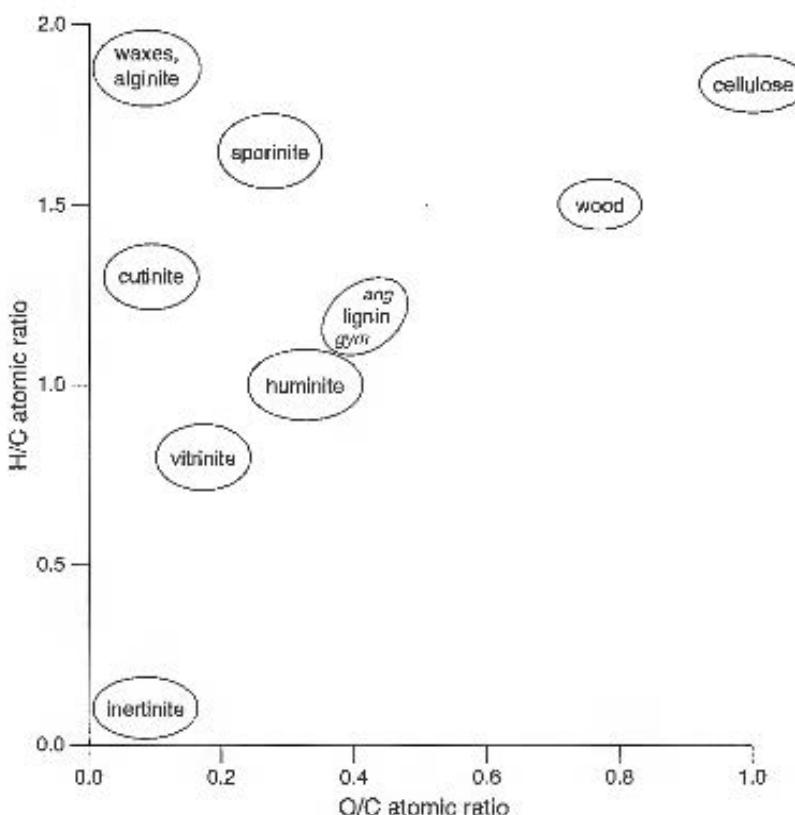


**Fig. 3.11** Major iron(III)-chelating groups in siderophores and structures of some siderophores.

# Van Krevelin Plot

**Table 4.5** Coal analysis (ASTM procedures; after Ward 1984)

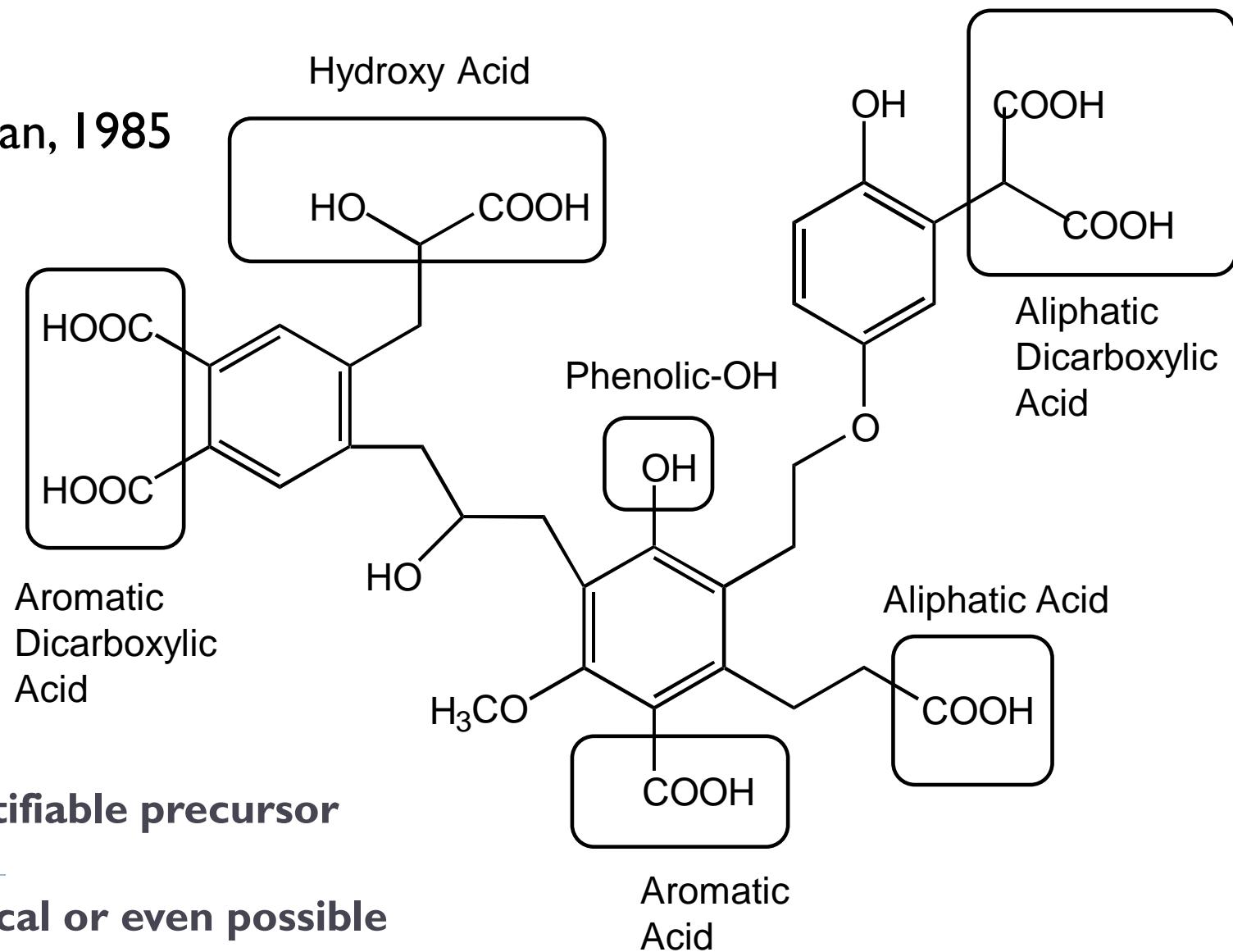
<i>proximate analyses</i>	
moisture	drying at 110°C (avoiding oxidation and decomposition)
volatile matter	volatiles liberated at 950°C in absence of air, excluding moisture
ash	inorganic residue from combustion
fixed carbon	C remaining after volatiles determination (i.e. coking potential)
<i>ultimate analyses</i>	
elemental analysis	C, H, N and S content
oxygen	determined by difference (i.e. total minus C, H, N, S content, and correcting for inorganic mineral content); more rarely determined directly
<i>examples of other analyses</i>	
forms of sulphur	organic, sulphide, native S, sulphate
other elements	e.g. trace elements, phosphorus, chlorine, carbonate CO <sub>2</sub>
relative density	depends on ash content and maturity
specific energy	energy liberated upon combustion



**Fig. 4.3** Chemical composition of major coal macerals and plant tissues plotted on a van Krevelen diagram (after Tissot & Welte 1984; Hedges et al. 1985). Lignin incorporates differences between angiosperms (*ang*) and gymnosperms (*gym*) (after Hatcher 1990).

# Putting it all together?

From Thurman, 1985



- ▶ Many identifiable precursor structures
- ▶ Not practical or even possible

# Concentrations: Pedogenic

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- ▶ Land Sources
  - ▶ From Woody & non-woody plants, lignin, etc.
  - ▶ Depends on vegetation, soil, hydrology
- ▶ Attenuated by adsorption to clay soils
  - ▶ Parallel watersheds in Australia (Cotsaris et al., 1994 [Chamonix proceedings])
    - ▶ Clearwater Creek, high clay content: 2.5 mg/L TOC
    - ▶ Redwater Creek, sandy soil: 31.7 mg/L TOC

# Concentrations: Aquagenic

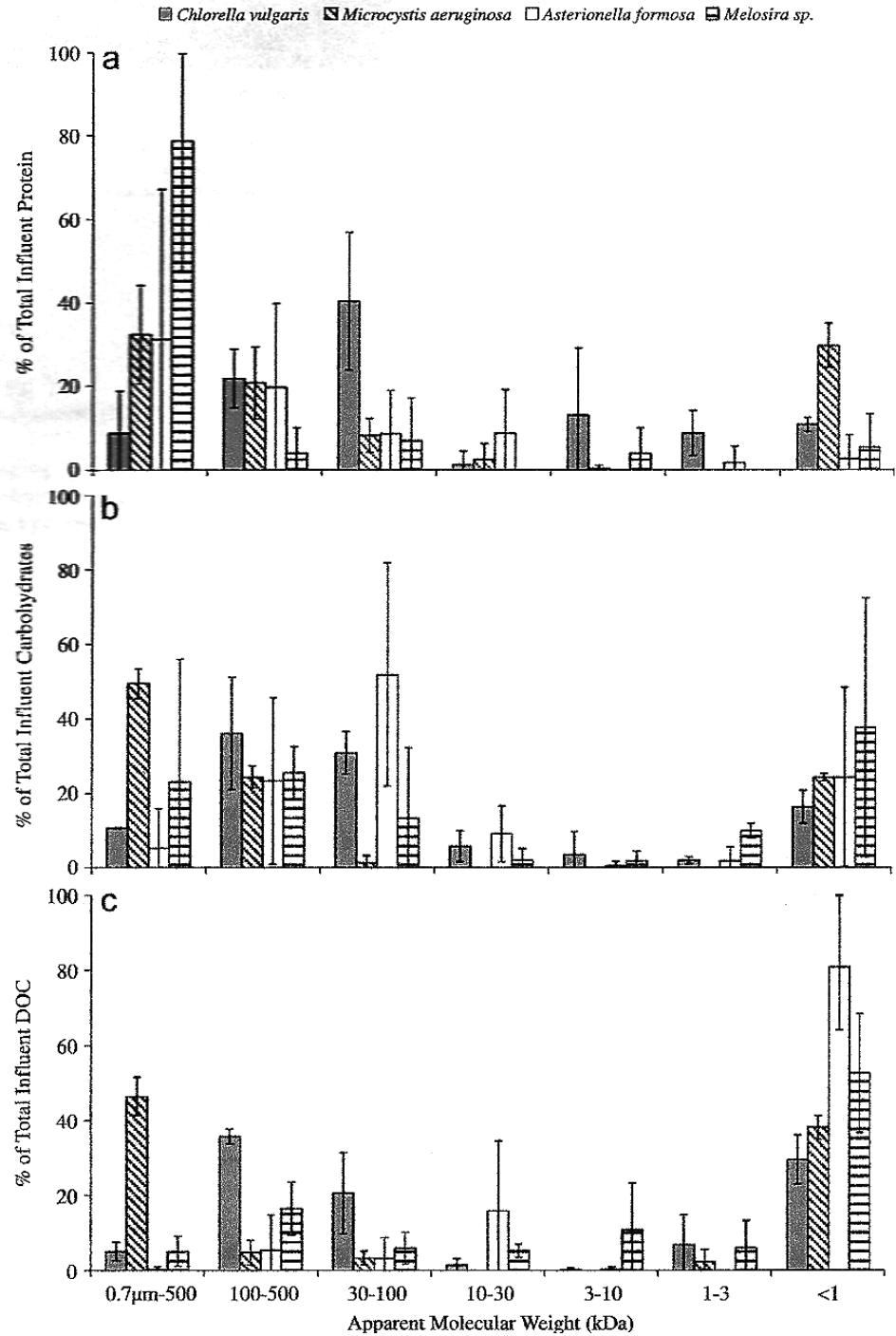
- ▶ Algal & aquatic plant Sources
  - ▶ Depend on nutrient levels / trophic state
- ▶ Concentrations in Lakes (mg/L) (Thurman, 1985)

- ▶ Groundwater
  - ▶ No algae

Trophic State	Mean DOC	Range
Oligotrophic	2	1-3
Mesotrophic	3	2-4
Eutrophic	10	3-34
Dystrophic	30	20-50

# MW vs type

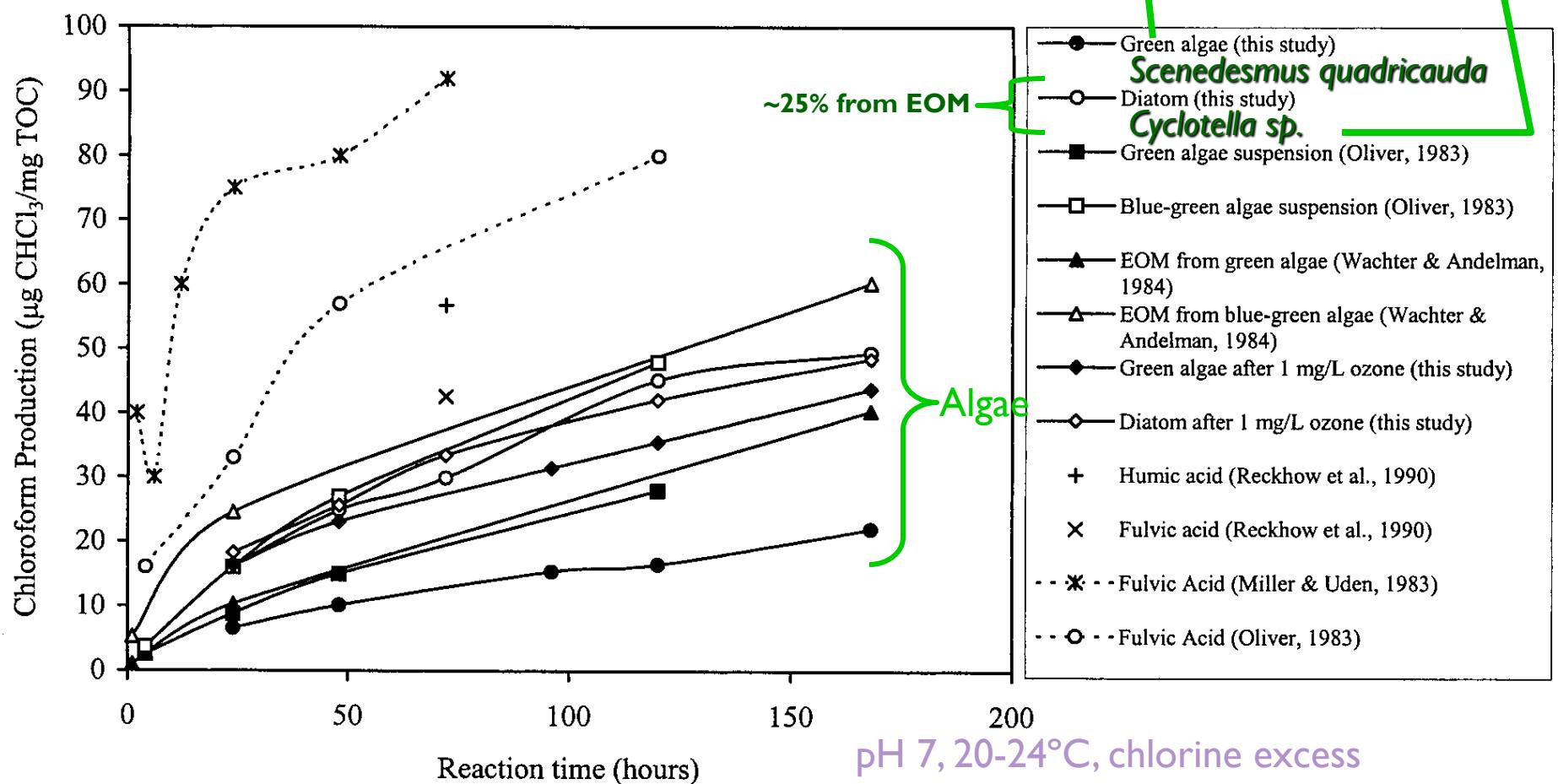
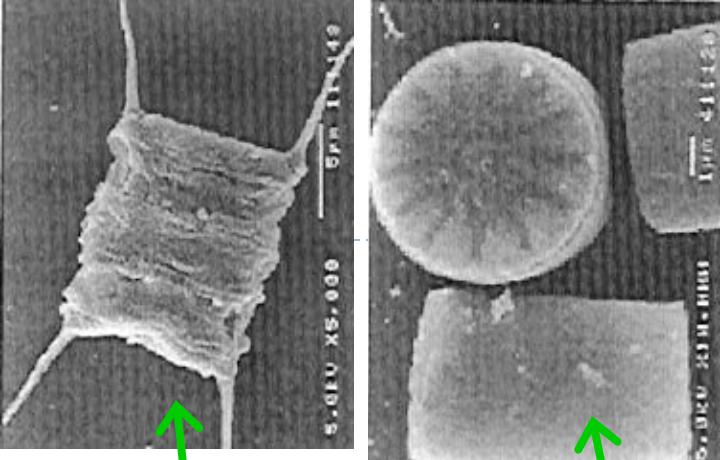
- ▶ Algogenic organic matter (AOM)
  - ▶ Proteins & carbohydrates
  - ▶ Large polymers with monomers



Henderson et al., 2008

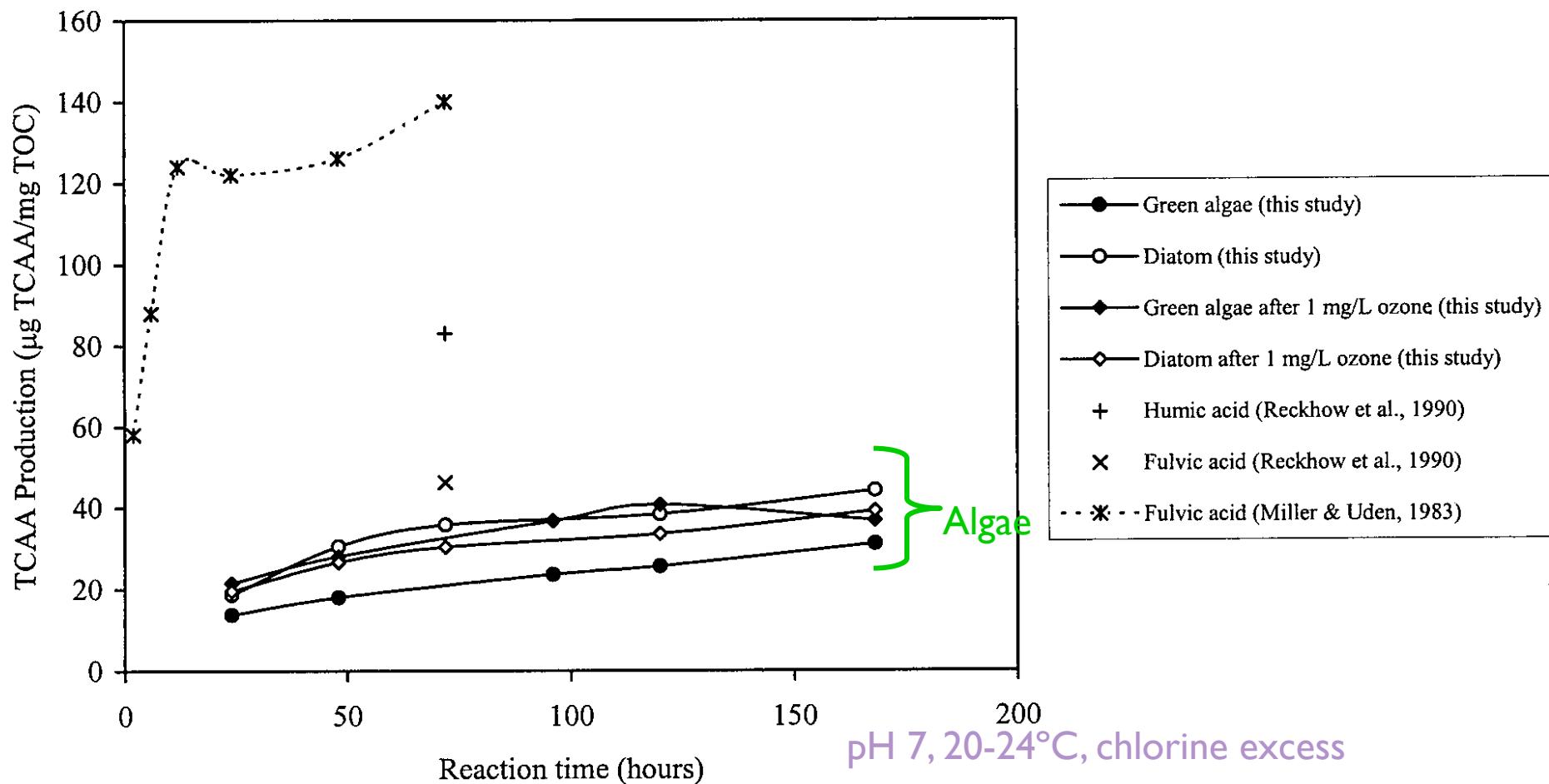
# Algae as THM Precursors

- ▶ From: Plummer & Edzwald, 2001
- ▶ [ES&T:35:3661]



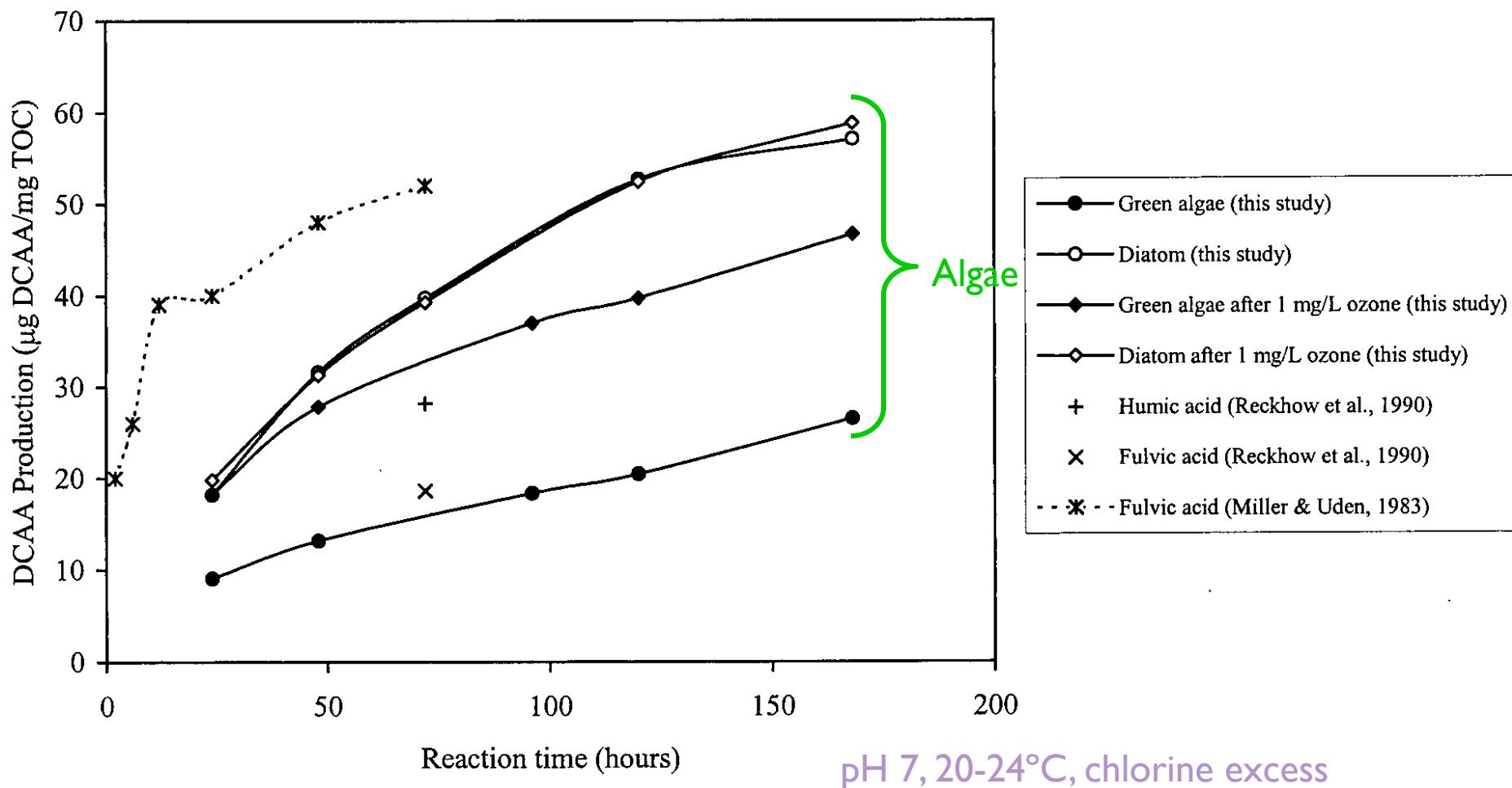
# Algae as TCAA Precursors

▶ Not much impact?



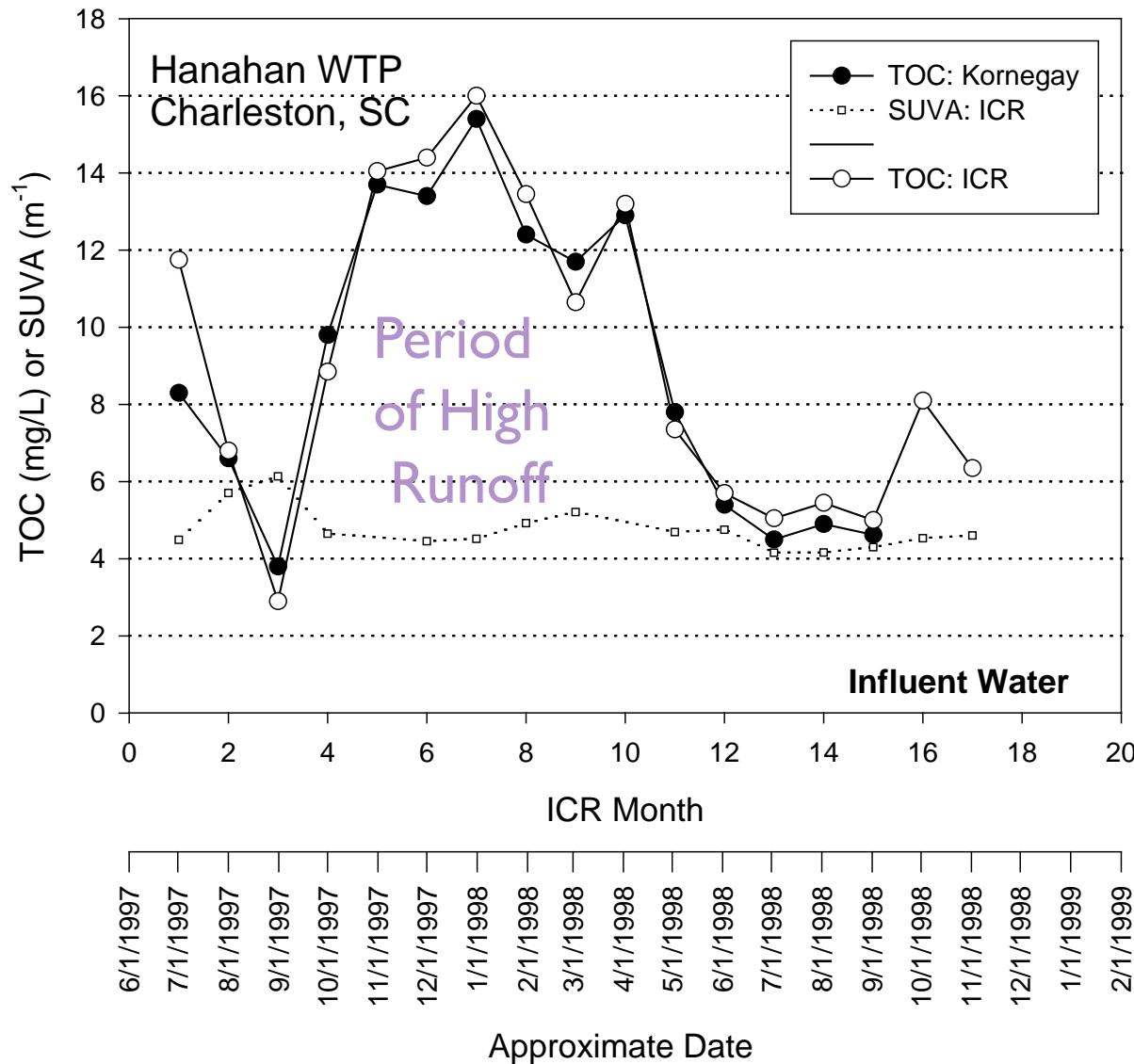
# Algae as DCAA Precursors

► Are Algae important sources of dihalo-AA precursors?



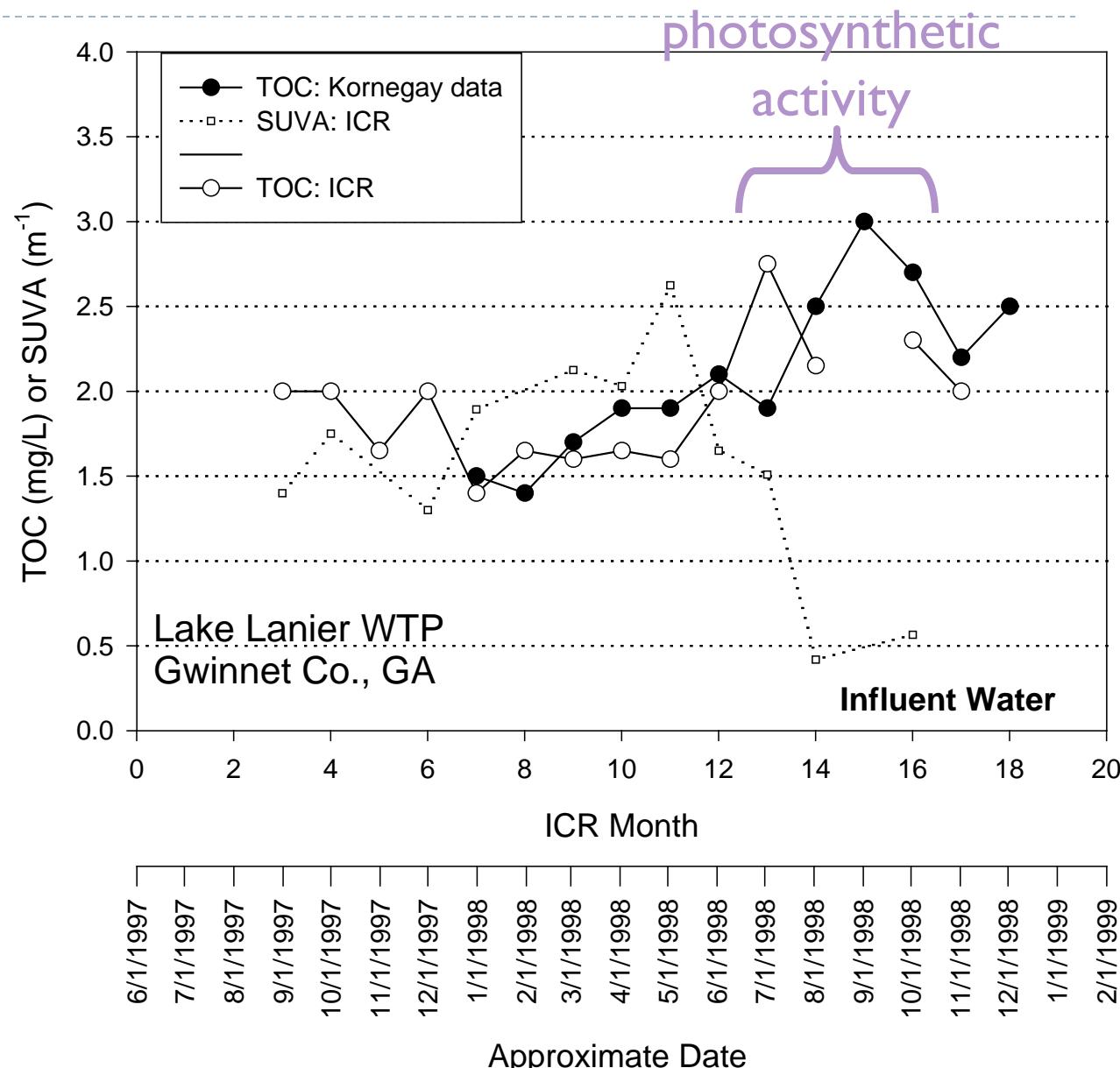
# Annual TOC Cycles

- ▶ Edisto River
  - ▶ Former source for Charleston's (SC) Hanahan WTP
  - ▶ Flushing of TOC during high rainfall months (cold period)

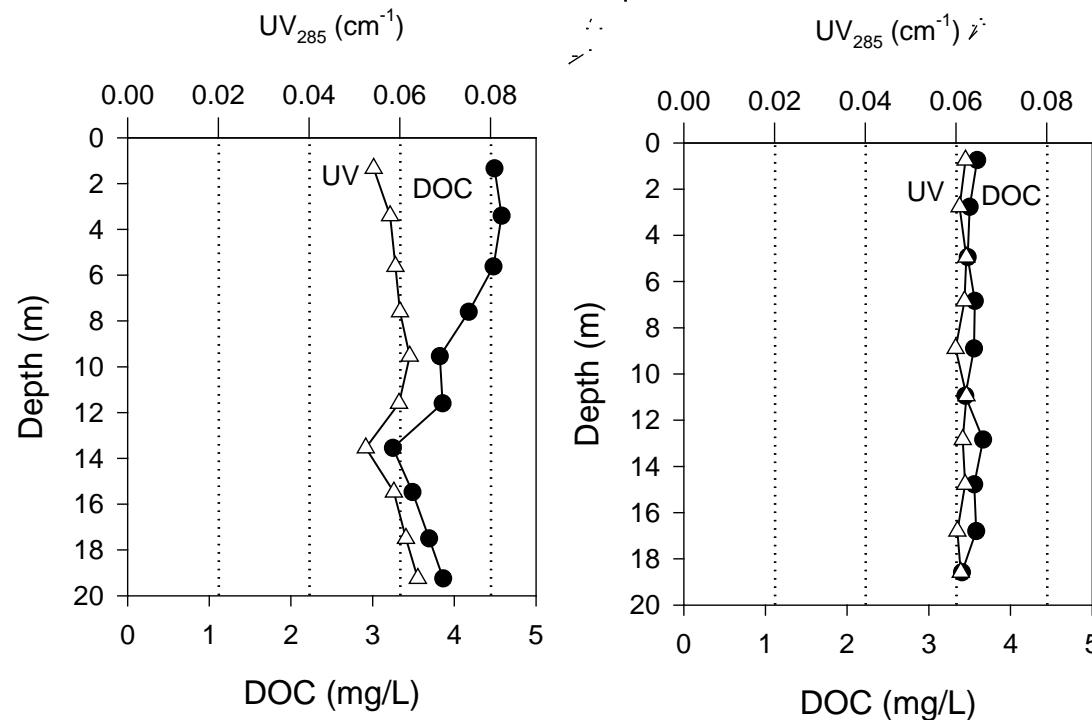
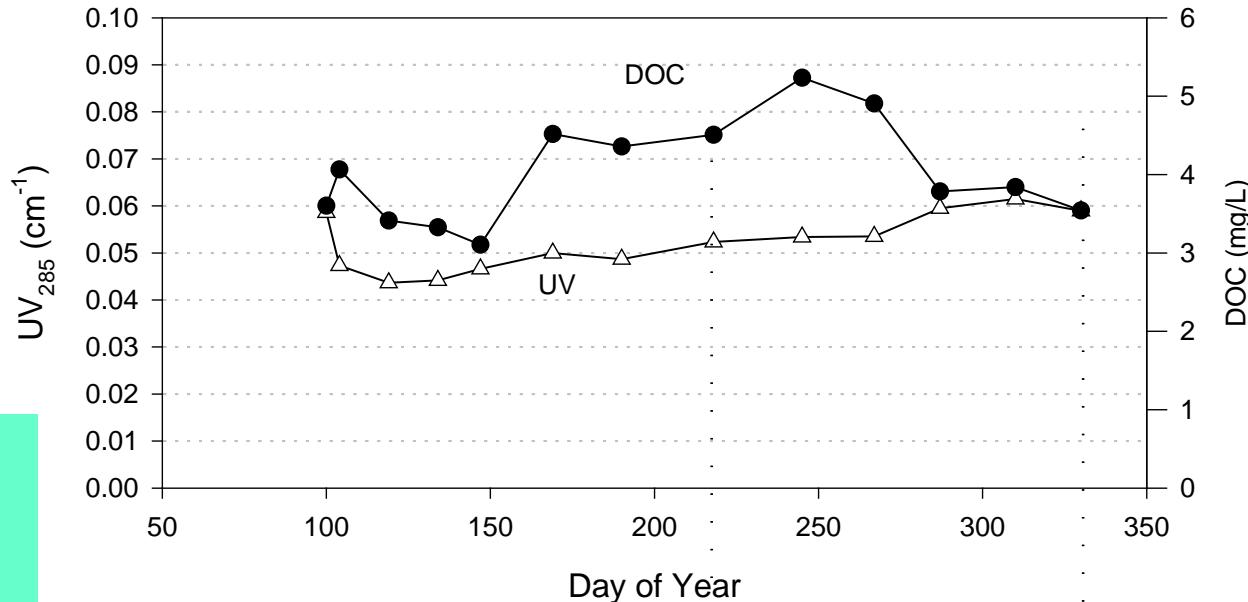


# Annual TOC Cycles

- ▶ Lake Lanier
  - ▶ Source for Gwinnett Co.'s (GA) Lanier WTP
  - ▶ High clay content in watershed

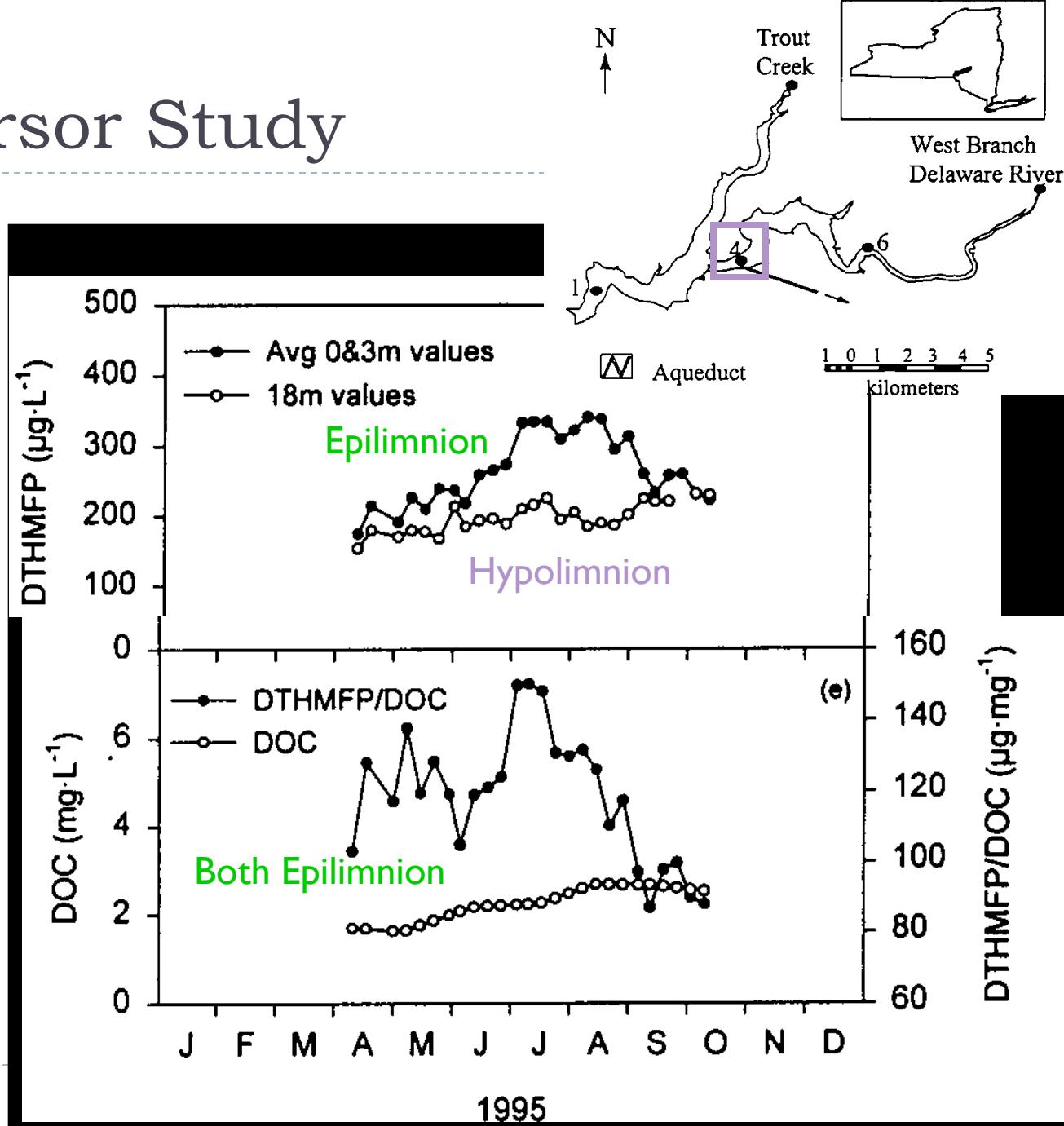


*Spatial and Temporal  
Distribution of DOC  
and UV absorbing  
Substances in Lake  
Bret*  
**(from Zumstein  
& Buffle, 1989;  
and Krasner et  
al., 1996)**



# THM Precursor Study

- ▶ Cannonsville Reservoir
  - ▶ Catskill-Delaware water supply for NYC
- ▶ Stepczuk et al., 1998
  - ▶ J. Lake Res. Mgmt. 14(2-3)356



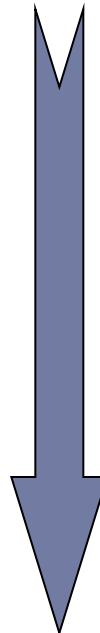
# Metabolic

## biochemicals

- ▶ Sugars, starches
- ▶ Proteins
- ▶ Cellulose
- ▶ Hemicellulose
- ▶ Fats & waxes
- ▶ Lignins & phenolics

## Formation Potential

- ▶ Low
- ▶ Moderate
- ▶ Low
- ▶ Low
- ▶ Low
- ▶ high



Decreasing  
biodegradability

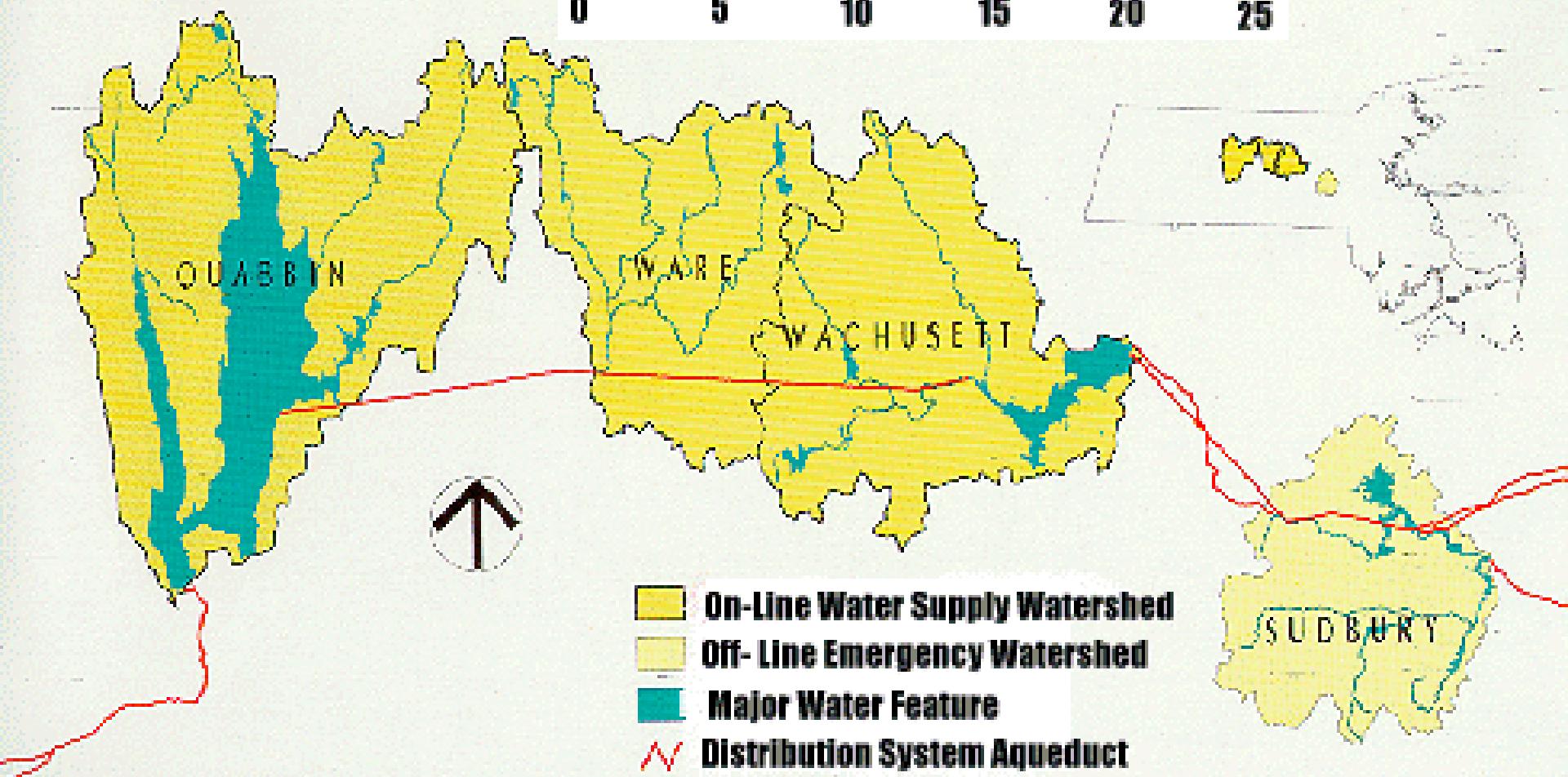
Terpenoids - ??

# Structural

- ▶ Simplification: Doesn't explicitly consider bacterial metabolites

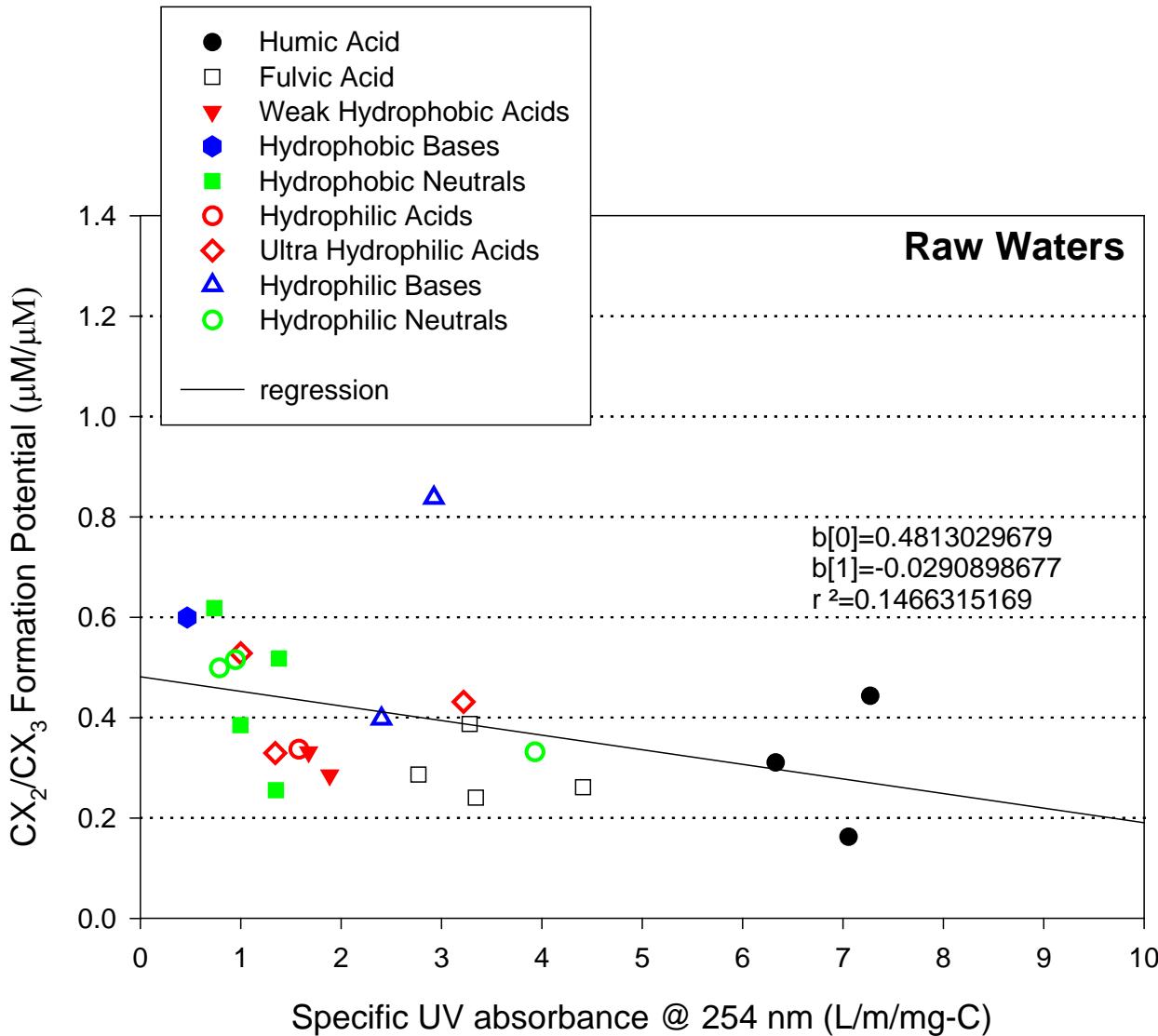
**MDC Division of Watershed Management  
Water Supply Watersheds**

**Statute Miles**

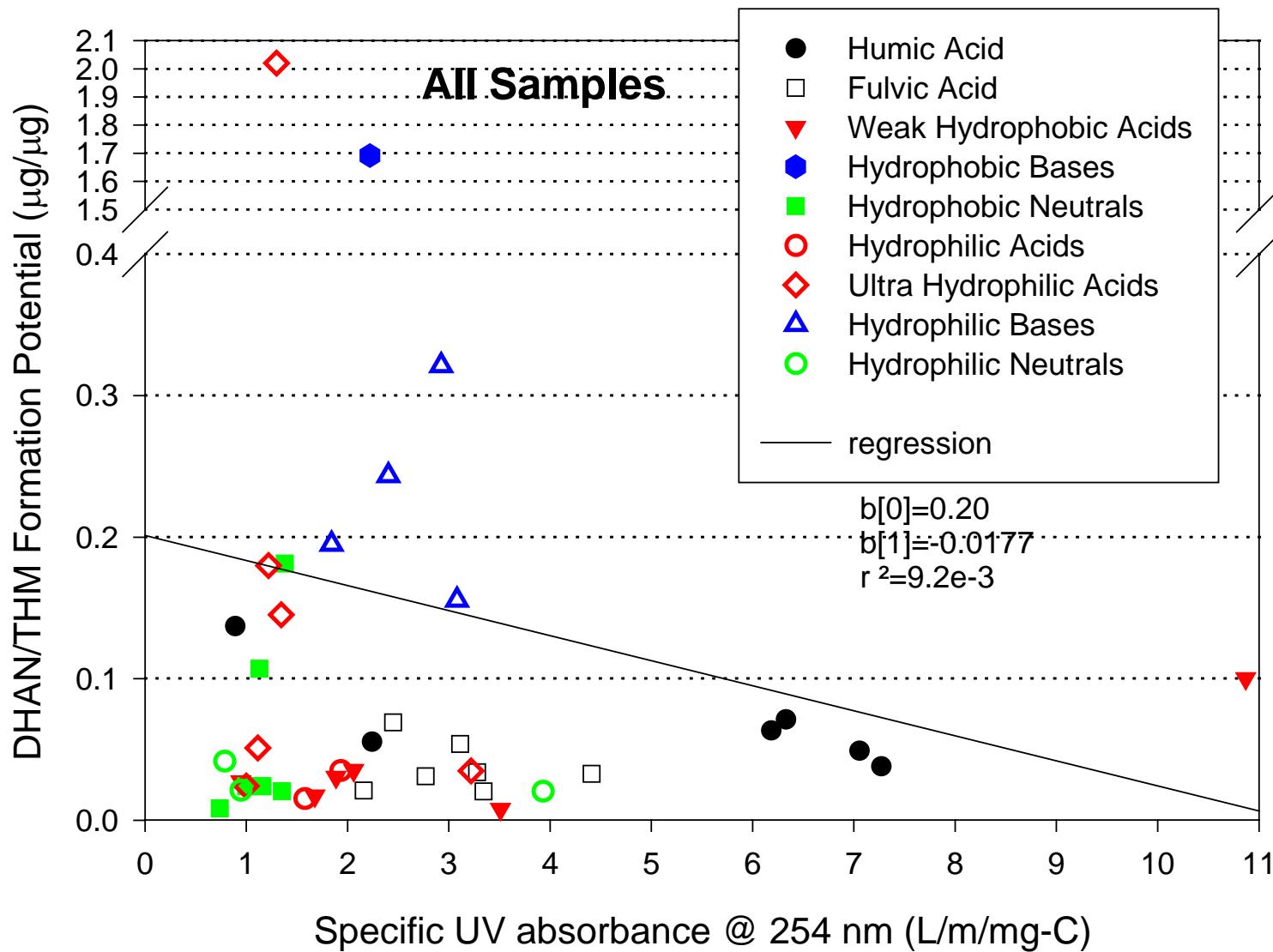


# Dihalo and Trihalo DBPs

- ▶ NOM Fractions
  - ▶ Evidence for greater importance of dihalo species in non-lignin based NOM

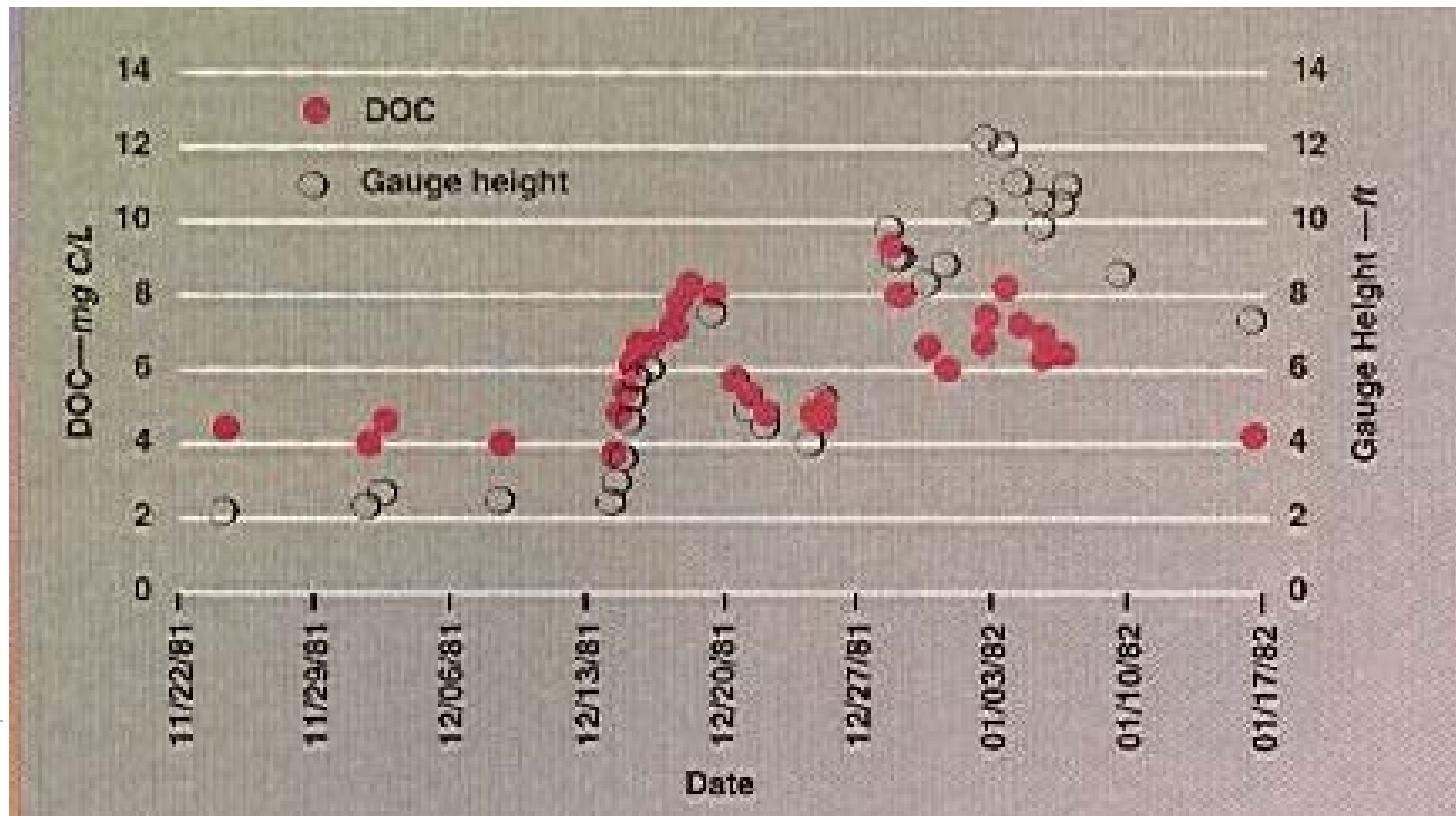


# DHAN/THM Ratio vs SUVA

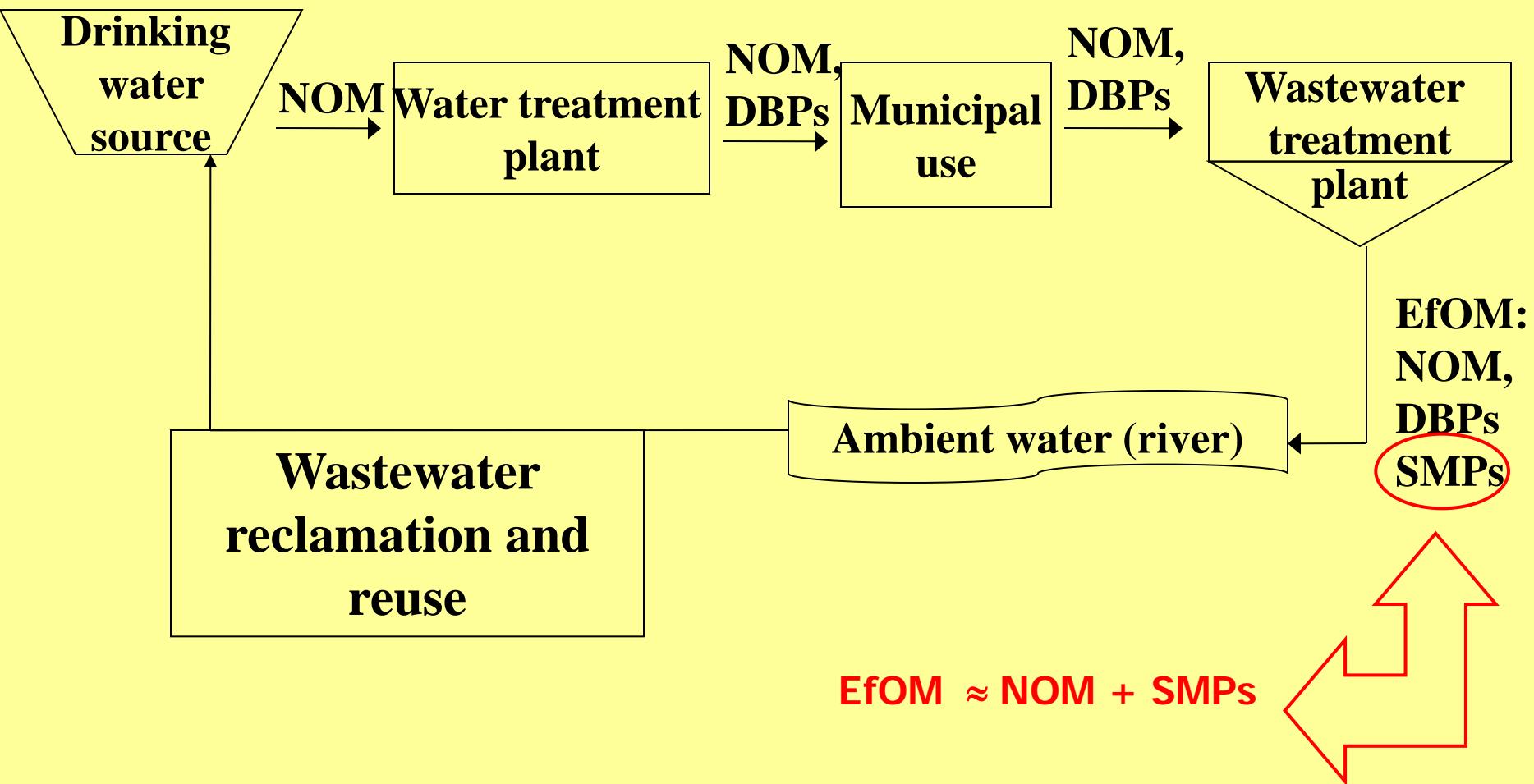


# DOC and runoff

- ▶ Ogeechee River (GA)
  - ▶ From Aiken & Cotsaris, 1995
    - [JAWWA 87(1)36]



# What is EfOM?



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► To next lecture

