

Updated: 17 April 2013 [Print version](#)

CEE 577: Surface Water Quality Modeling

Lecture #36

Toxics: PCBs in the Great Lakes (Jeremiason et al., 1994)

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What are the PCBs

- Biphenyl
- 2,2' - Dichlorobiphenyl
- 2,3' - Dichlorobiphenyl

**Homologs (11)
Isomers (1-46)
Congeners (209)**

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History

- 1930: Monsanto is major US producer
- 1970: Monsanto decides to sell PCBs only for closed use
- 1975: NY State warns public about salmon and bass in Hudson
- 1979: PCB manufacture banned in US
- 1982: NY State begins dredging “hot spots”
- 1990: all PCB-containing equipment must be removed from US public buildings

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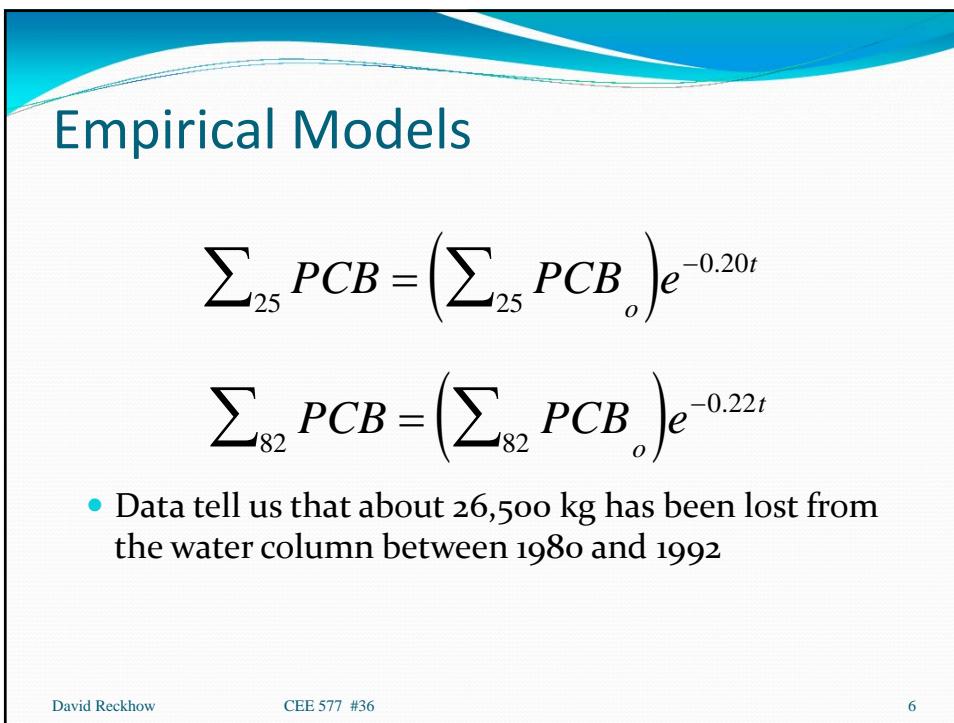
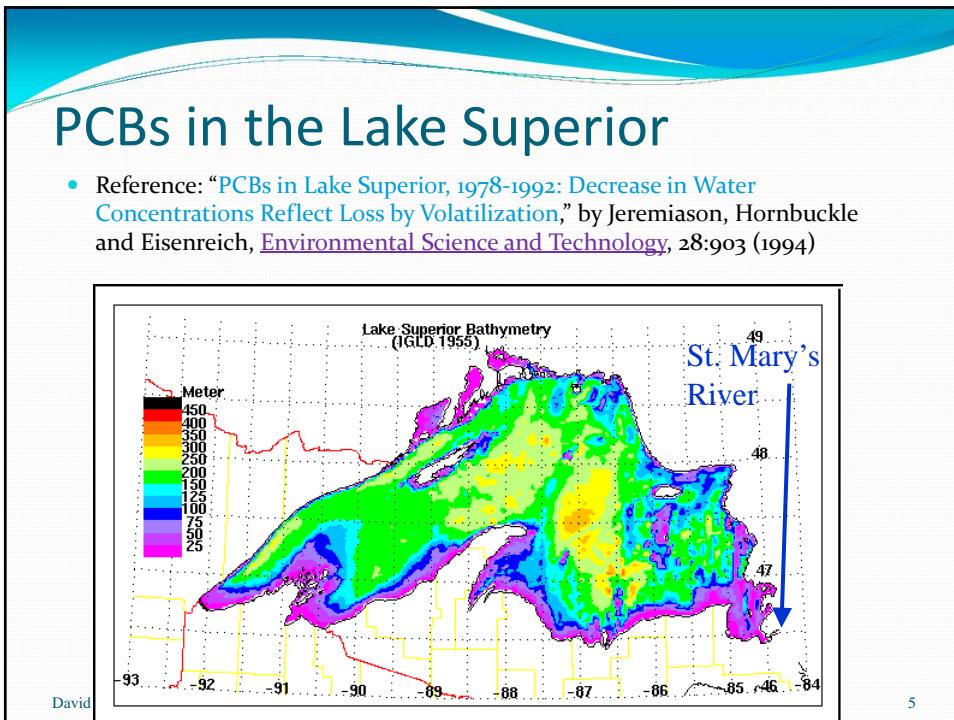
Arochlor Mixtures

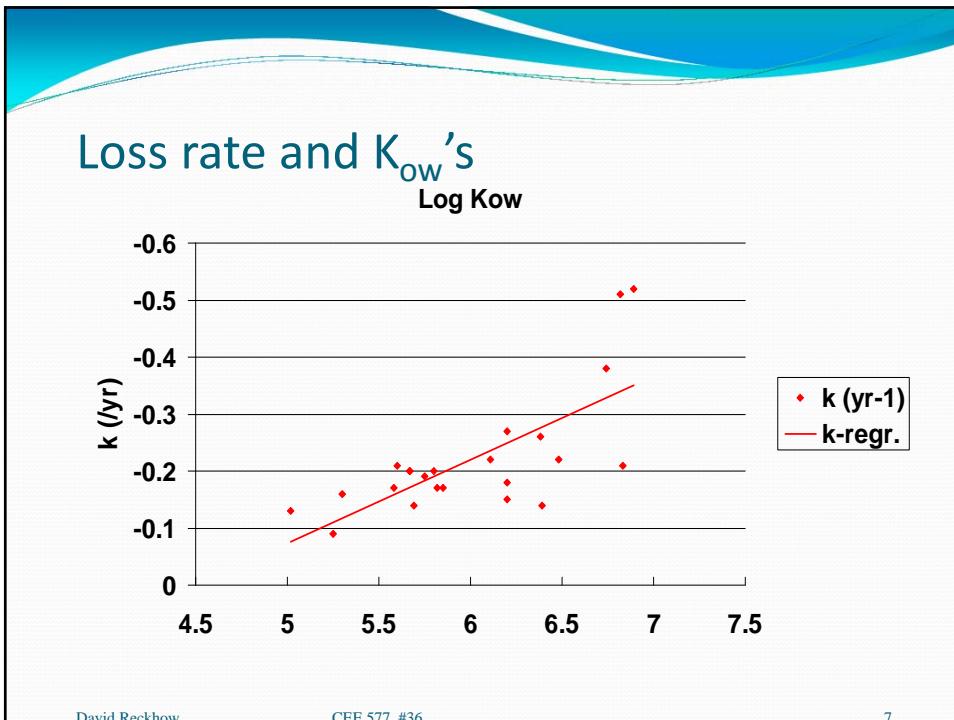
- Arochlor 12xx (xx=% chlorine)
 - 1221: 50% Cl₁, 35% Cl₂
 - 1232: 26% Cl₁, 29% Cl₂, 24% Cl₃
 - 1242: 13% Cl₂, 45% Cl₃, 31% Cl₄
 - 1248: 49% Cl₄, 27% Cl₅
 - 1254: 15% Cl₄, 53% Cl₅, 26% Cl₆
 - 1260: 12% Cl₅, 42% Cl₆, 38% Cl₇
 - 1262: no data
 - 1268: no data

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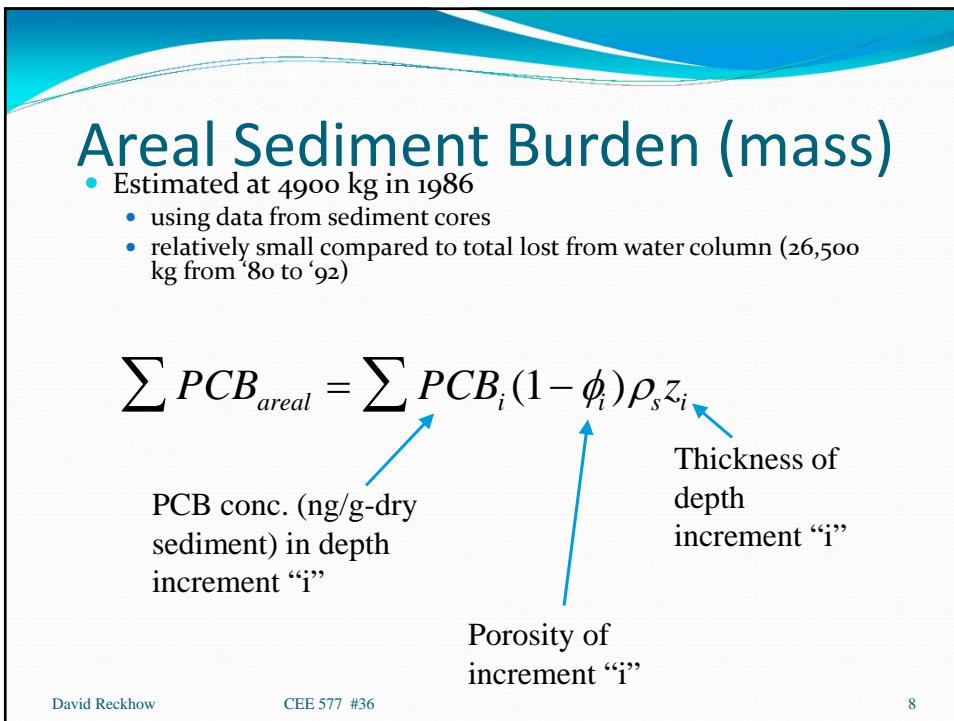




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Inputs

- Riverine
 - Known Q
 - Estimate c from analysis of pristine rain
- Other
 - estimates from industrial, municipal, (urban) runoff and storm sewer flows gives a combined total of about 40 kg/yr

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Inputs (cont.)

- Direct Atmospheric deposition
 - wet deposition
 - dry deposition
 - calculated for 4 seasons, then averaged

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Outputs

- Outflow
 - St. Mary's River
$$W_{outflow} = 7.1 \times 10^{13} L / yr (0.84 ng / L) = 60 kg / yr$$
- Burial (net loss to sedimentation)
 - estimated at 110 kg/yr from sediment cores collected in 1986 and 1990
- Net Volatilization
 - true volatilization minus gas absorption
 - assumed to account for missing flux

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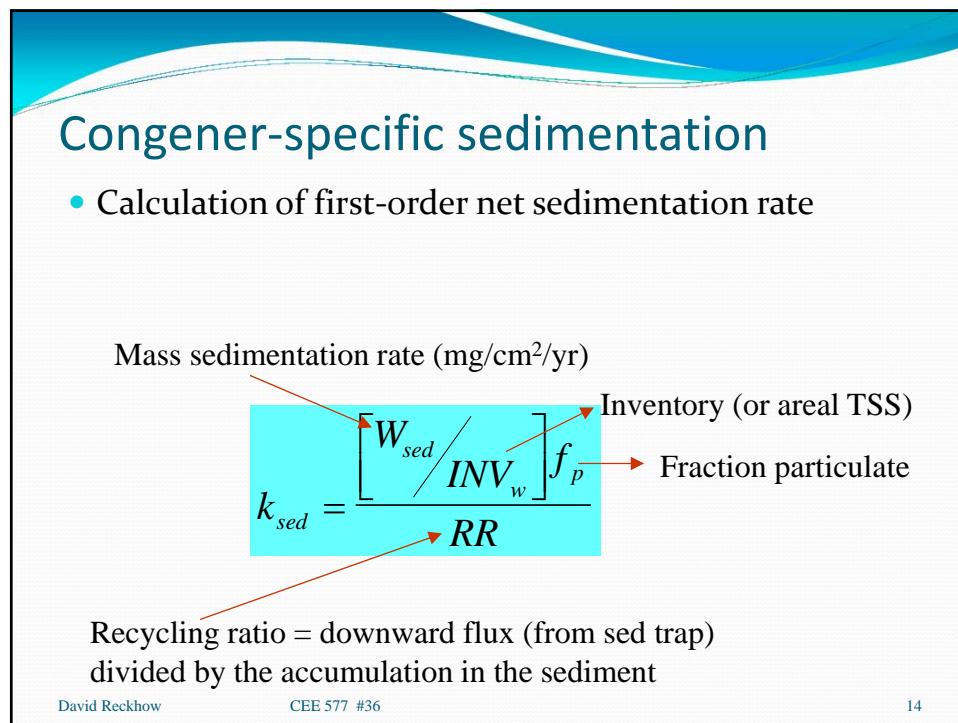
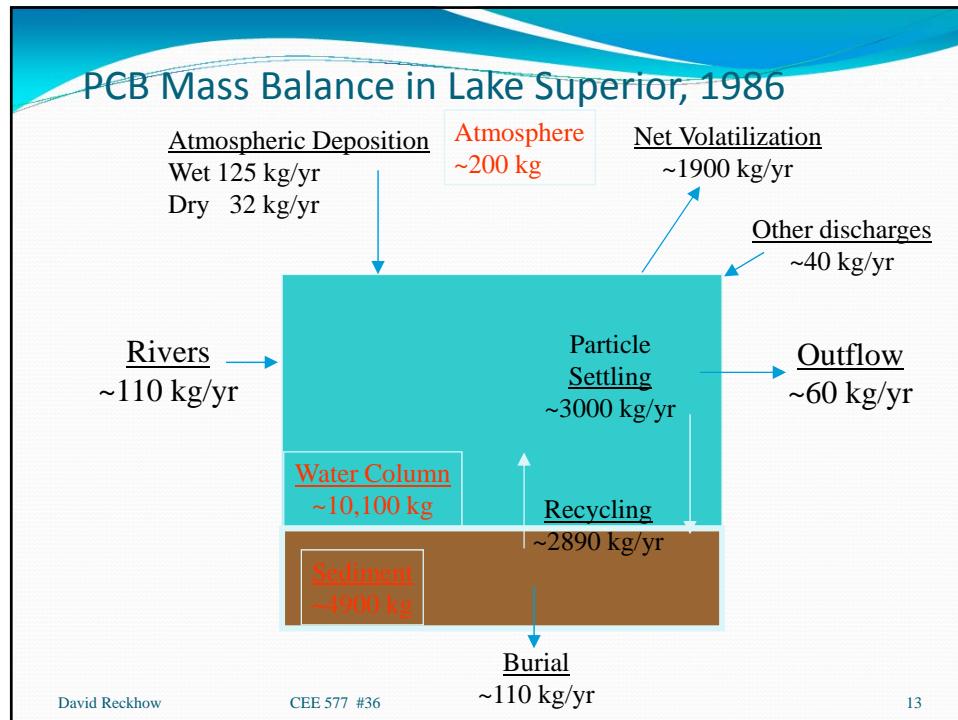
Reactions

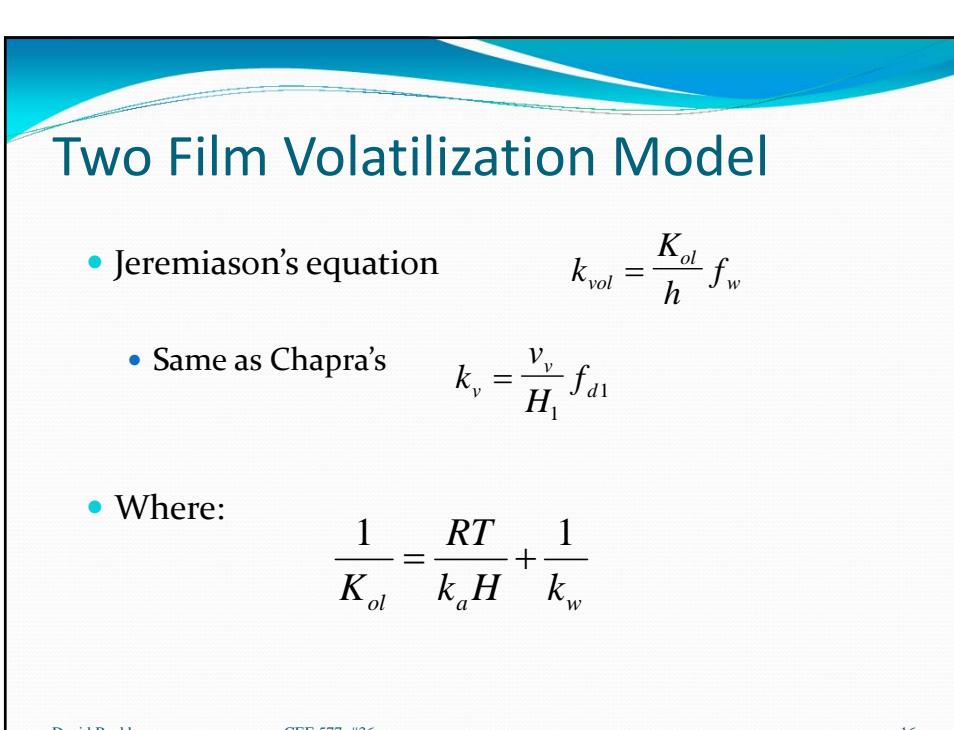
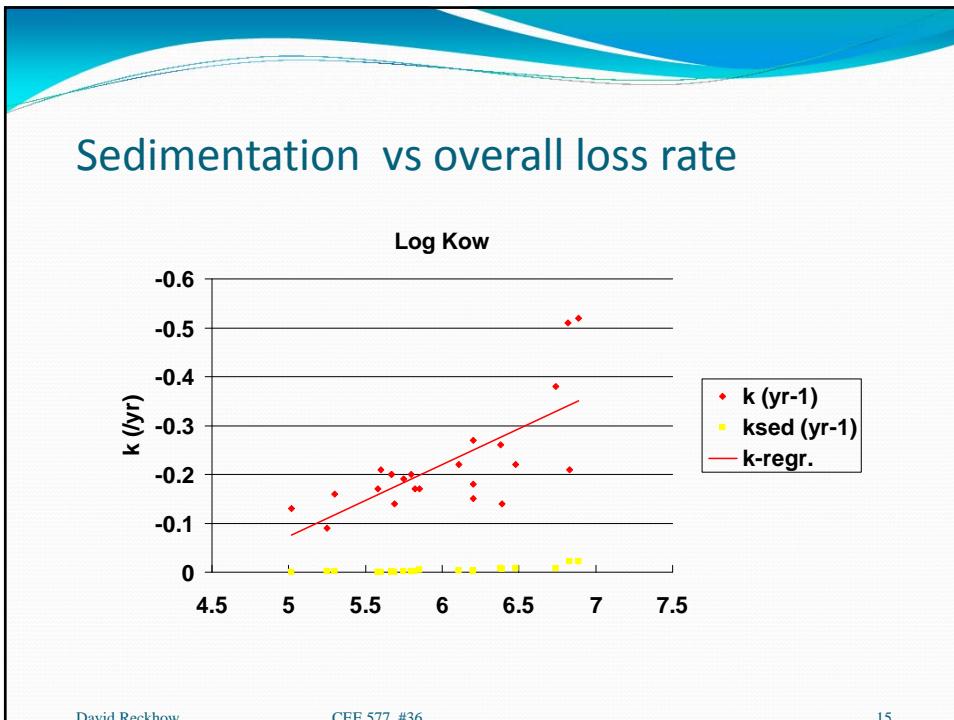
- NONE!
 - “evidence does not exist to support PCB degradation in Lake Superior or any other oligotrophic, aerobic system exhibiting low ambient concentrations”

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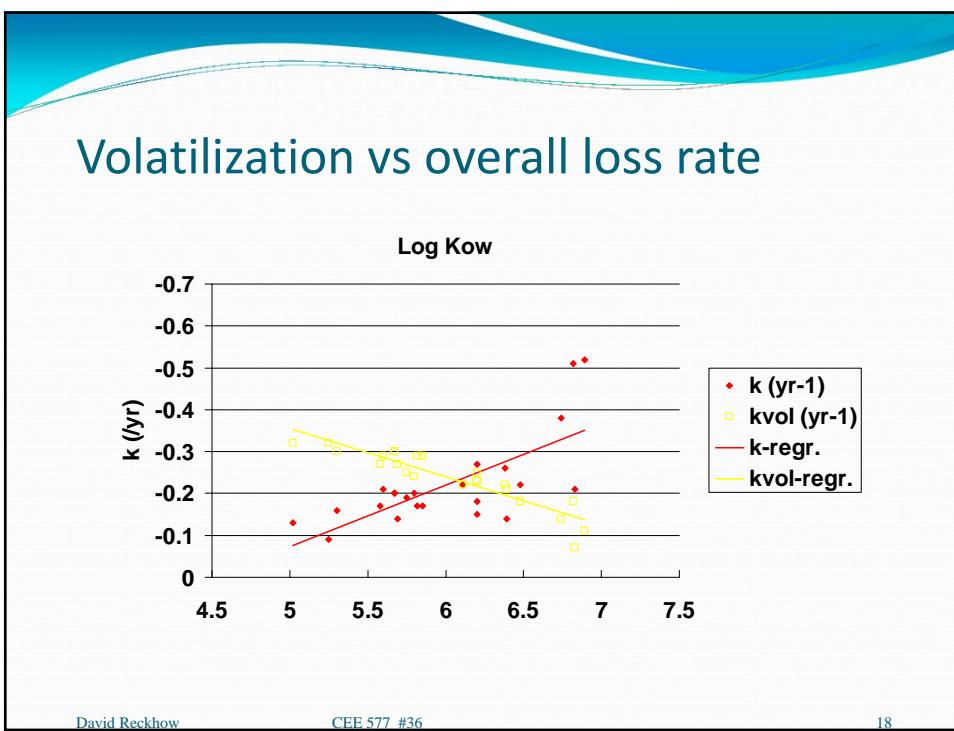
Estimating 2-film parameters

- The gas film coefficient
 $k_{a,H_2O} = 0.2u_{10} + 0.3$
 $k_{a,PCB} = k_{a,H_2O} \left(\frac{D_{PCB,air}}{D_{H_2O,air}} \right)^{0.61}$
- The liquid film coefficient
 $k_{w,CO_2} = 0.45u_{10}^{1.64}$
 $k_{w,PCB} = k_{w,CO_2} \left(\frac{Sc_{PCB}}{Sc_{CO_2}} \right)^{-0.5}$

Kinetic viscosity: molecular diffusivity

$\text{Dav } Sc = \frac{\mu}{\rho \cdot D_v}$ Schmidt Number

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- [To next lecture](#)