

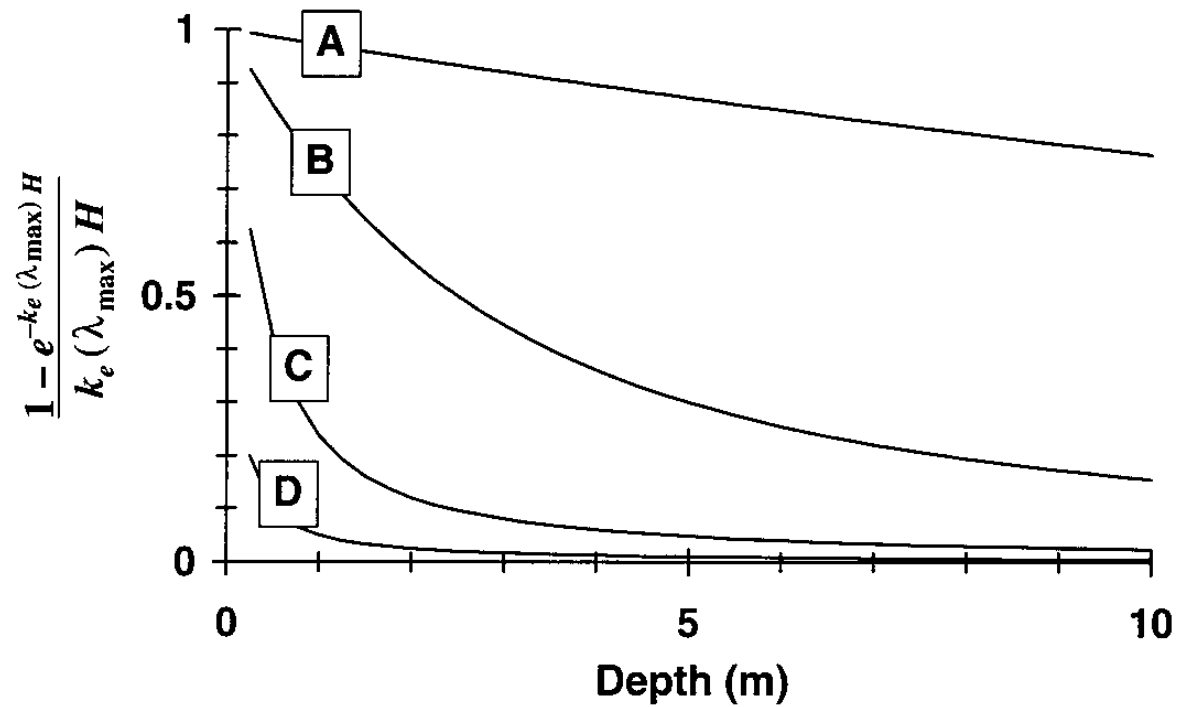
CEE 577: Surface Water Quality Modeling

Lecture #34

Toxics: Hydrolysis and Biodegradation:
Recapitulation and Simplified Forms
(Chapra, L42, L43 & L44)

Water type	Chla (mg L ⁻¹)	DOC (mg L ⁻¹)	ss (mg L ⁻¹)
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A	Pure water	0	0	0
B	Lake Tahoe	0.001	0.1	0.5
C	Eutrophic	0.01	0.5	5
D	Hypereutrophic	0.1	2	20



Biotransformation

- Microbially mediated transformation of organic and inorganic contaminants
- Biochemical processes:
 - Metabolism: toxicant is used for synthesis or energy
 - Cometabolism: not “used”, but transformed anyway
- Chemical Effects:
 - Detoxication: Toxic to Non-toxic
 - mineralization
 - Activation: Non-toxic to Toxic

Bio kinetics

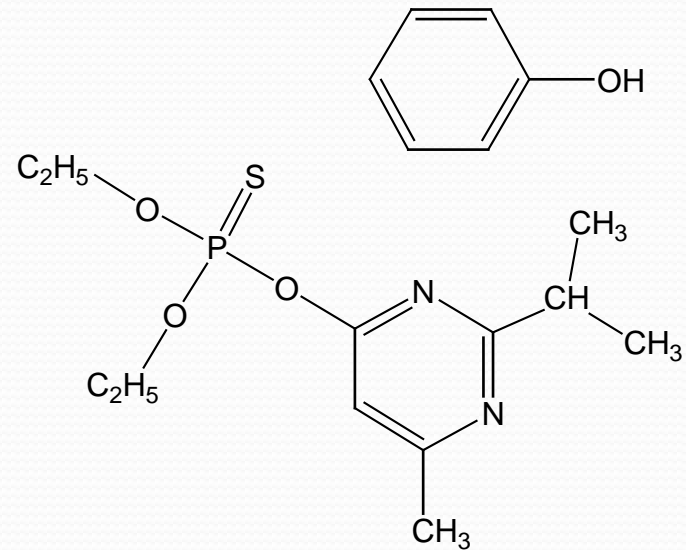
- Michaelis-Menten equation:
 - μ_{\max} = maximum growth rate (yr^{-1})
 - X = microbial biomass ($\text{\#cells}/\text{m}^3$)
 - Y = yield coefficient (cells produced per mass toxicant removed, $\text{\#cells}/\mu\text{g}$)
 - k_s = half-saturation constant ($\mu\text{g}/\text{m}^3$)
 - k_b = rate of biotransformation (yr^{-1})
- If $c \ll k_s$, then:

$$k_b = \frac{\mu_{\max} X}{Y(k_s + c)}$$

$$k_b = \frac{\mu_{\max} X}{Yk_s} = k_{b2} X$$

Bio kinetics (cont.)

- Wide environmental range
 - phenol: $k_b = 4.0 \text{ d}^{-1}$
 - diazinon: $k_b = 0.016 \text{ d}^{-1}$
- Temperature correction
 - $\theta = 1.04 - 1.095$



$$(k_b)_T = (k_b)_{20} \theta^{T-20}$$

Hydrolysis

- Reaction with water and its constituents

- H_2O $k_h = k_n$

- OH^- $k_h = k_b [OH^-]$

- H^+ $k_h = k_a [H^+]$

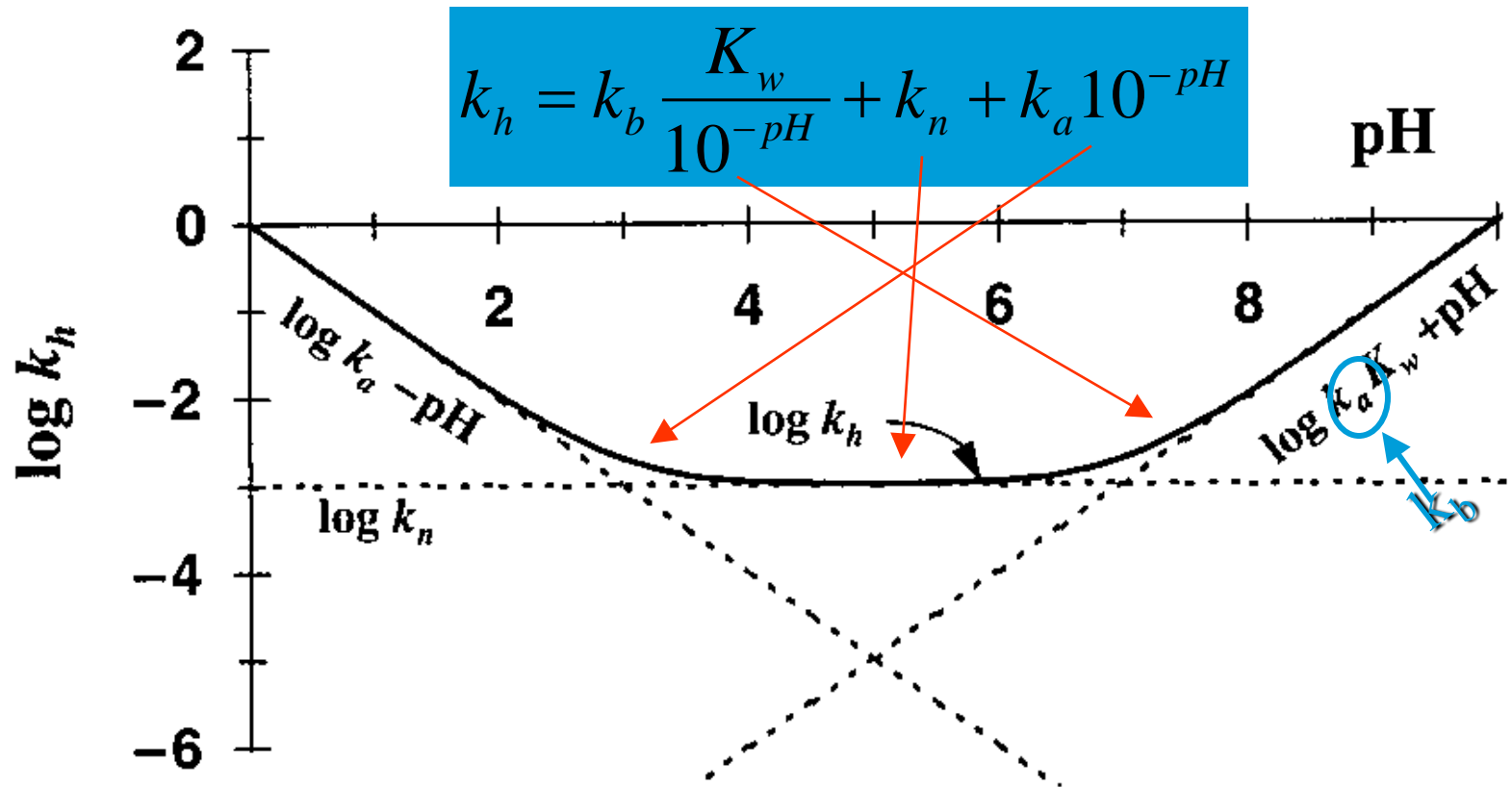
- Autodissociation $K_w = [OH^-][H^+]$

- Combining: $k_h = k_b [OH^-] + k_n + k_a [H^+]$

- or:

$$k_h = k_b \frac{K_w}{10^{-pH}} + k_n + k_a 10^{-pH}$$

Graphic Representation



Special Considerations for Metals

- In general they are not subject to decomposition
 - e.g., biodegradation, hydrolysis, photolysis
 - exception: radionuclides undergo radioactive decay
- Most do not volatilize (Hg is an exception)
- They speciate into many forms which differ in toxicity and behavior
- Natural background and non-point loadings may be quite high

- To next lecture