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# CEE 577: Surface Water Quality Modeling

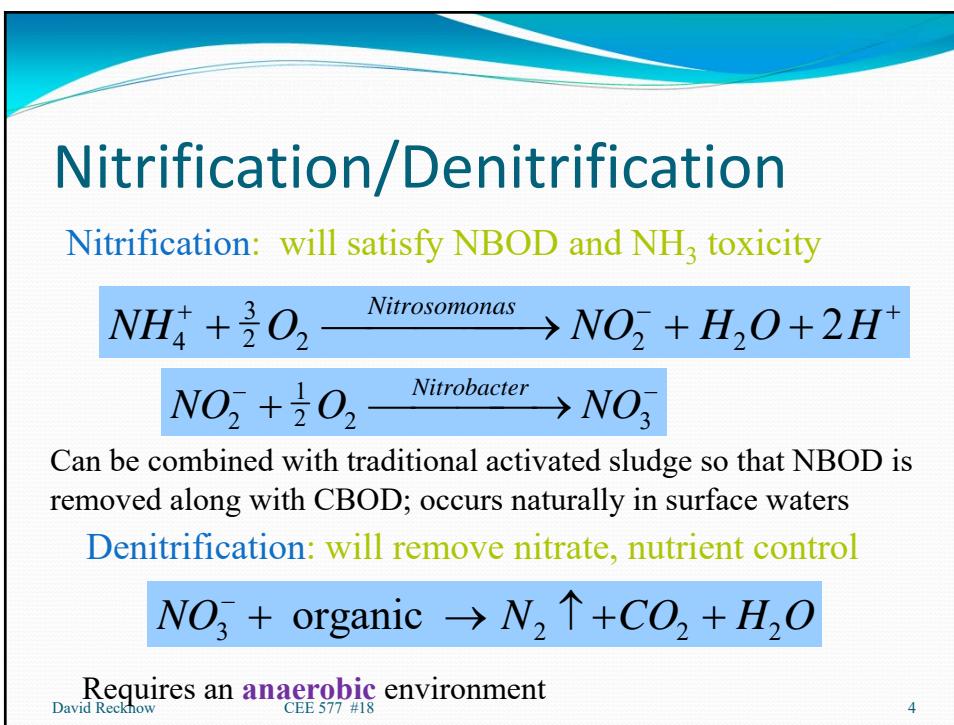
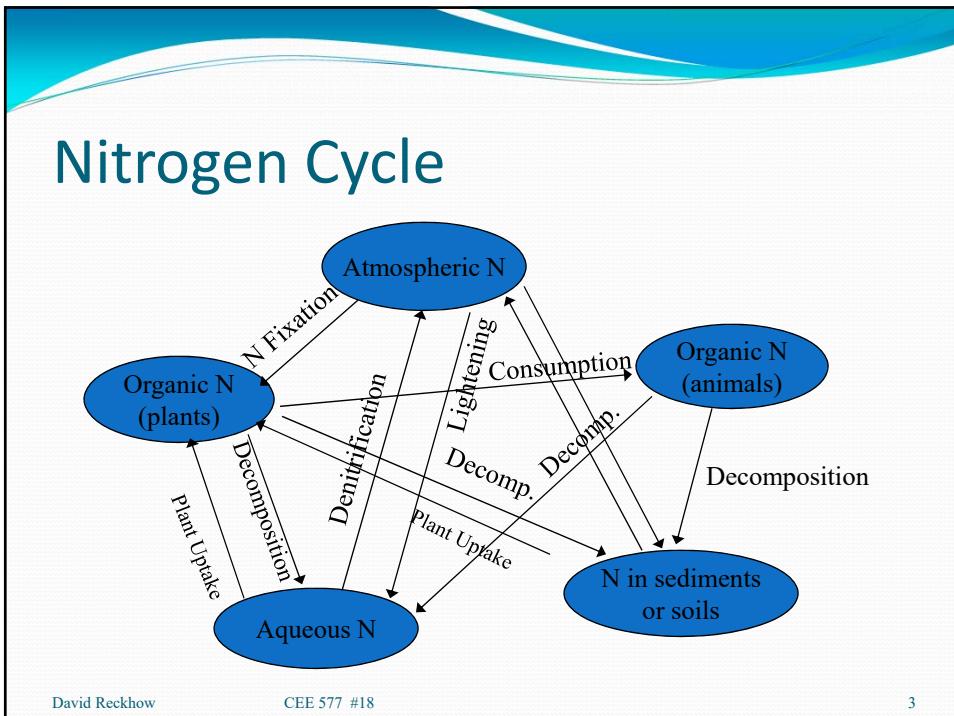
## Lecture #18 Streeter-Phelps: Distributed Sources & Nitrogen (Chapra, L22 & L23)

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## Nitrogen (Chapra L23)

- Nitrification/Denitrification
  - nitrification: oxygen consuming
  - denitrification: anaerobic, form  $N_2$
- Eutrophication
  - stimulates plant growth
- Nitrate pollution
  - from fertilizers and nitrification
- Ammonia toxicity ( $NH_3$  form)

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## Modeling Nitrification

$$L_N = L_{N_o} e^{-k_n t} + \frac{S_D}{k_n} (1 - e^{-k_n t})$$

↓  
Point                      ↓  
                                Distributed

$$L_N = 4.57 * TKN$$

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## In-class problem I

- A poorly treated municipal wastewater is discharged into Evergreen Brook at milepoint zero. In addition there is a continuous discharge of soluble BOD from a series of hog farms extending from milepoint zero to mile 25.
  - The WWTP discharges sufficient BOD such that there is 12 mg/L BOD at the point of mixing, 30% of which is particulate
  - The hog farms release 5 g-BOD/d for every foot of stream length
  - The stream can be considered to have a uniform depth of 4 ft, a width of 22 ft and a rocky bottom. Velocity is 3 mi/d.
  - Assume a BOD settling rate of 1.2 d<sup>-1</sup>
- What is the BOD concentration 10 miles downstream and how much originates from each source (WWTP vs hog farm)?

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## In-class problem II

- First determine distributed loading term

$$S_d'' = \frac{S_d''}{A_c} = \frac{S_d'}{H} = \frac{5 \text{ g}/\text{ft} - d}{4 \text{ ft}(22 \text{ ft})} \left( \frac{1 \text{ ft}^3}{28.3 \text{ L}} \right) = 2.0 \frac{\text{mg}}{\text{L} - d}$$

- Next estimate deoxygenation rate

$$k_d = C \left( \frac{H}{8} \right)^{-0.434} = 0.3 \left( \frac{4}{8} \right)^{-0.434} = 0.405 \text{ d}^{-1}$$

- Then formulate BOD model

$$\begin{aligned} L &= L_o e^{-k_r t} + \frac{S_D}{k_r} \left( 1 - e^{-k_r t} \right) && \text{deoxygenation only, no settling} \\ &= 12(0.3)e^{-1.605 \text{ mg/L}} + 12(0.7)e^{-0.405 \text{ mg/L}} + \frac{2}{0.405} \left( 1 - e^{-0.405 \text{ mg/L}} \right) \end{aligned}$$

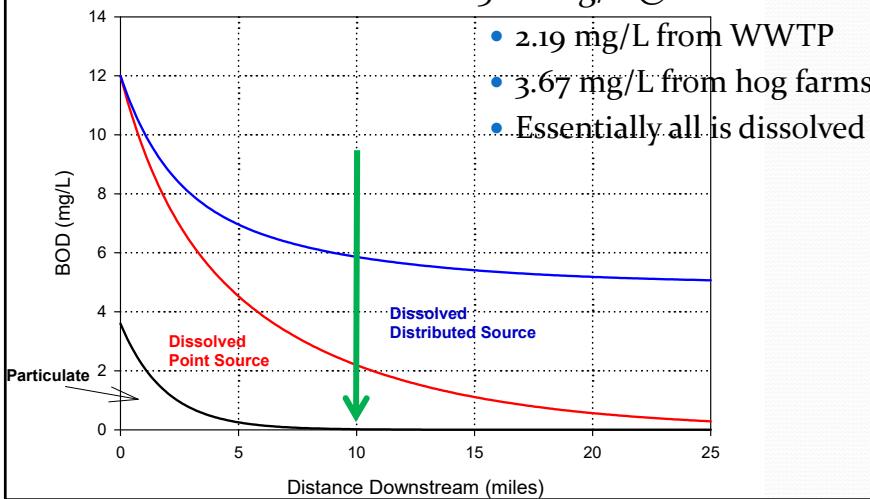
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## In-class problem II

- 5.86 mg/L @ 10 mi

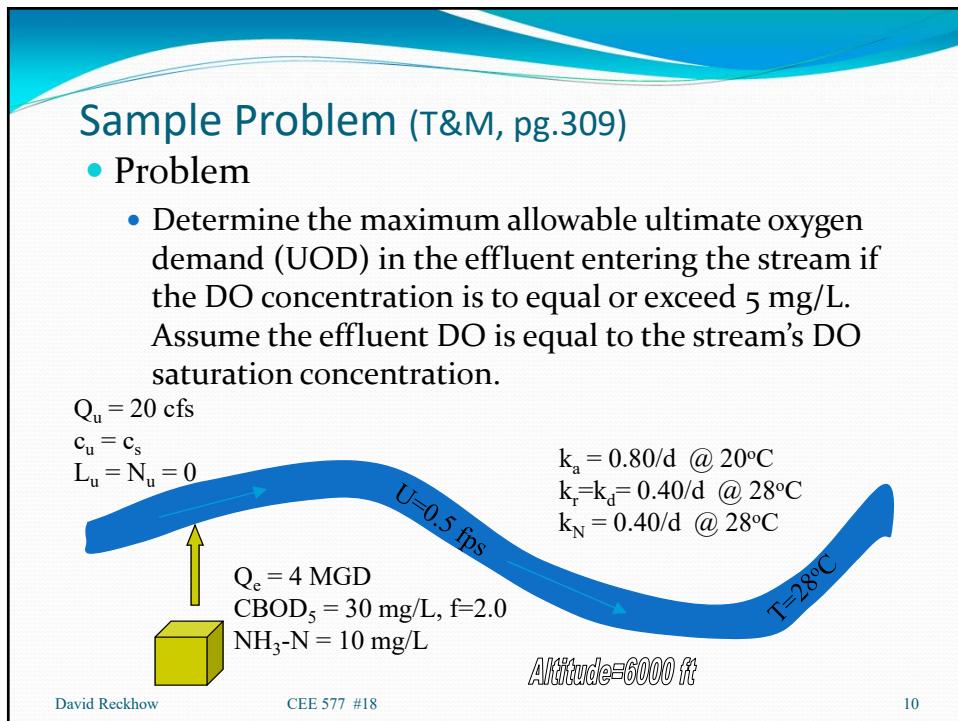


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**Full Equation**

$$\begin{aligned}
 D = & D_o e^{-k_a t} + \frac{k_d L_o}{k_a - k_r} (e^{-k_r t} - e^{-k_a t}) \\
 & + \frac{k_n L_{No}}{k_a - k_n} (e^{-k_n t} - e^{-k_a t}) \rightarrow \text{Point NBOD} \\
 & + \frac{-P + R + \left( S'_B / H \right)}{k_a} (1 - e^{-k_a t}) \\
 & + \frac{k_d S_d}{k_r k_a} (1 - e^{-k_a t}) - \frac{k_d S_d}{k_r (k_a - k_r)} (e^{-k_r t} - e^{-k_a t}) \\
 & + \frac{k_n S_{Nd}}{k_n k_a} (1 - e^{-k_a t}) - \frac{k_n S_{Nd}}{k_n (k_a - k_n)} (e^{-k_n t} - e^{-k_a t}) \quad \text{Distributed NBOD}
 \end{aligned}$$

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## Problem (cont.)

- Analysis of existing conditions

- Loading

- BOD and NBOD may be treated as one UOD load since the decay rates are the same in the stream. Assume only the ammonia is significant in the NBOD.

$$\begin{aligned} UOD(mg / L) &= fxCBOD_s + 4.57x(NH_3 - N) \\ &= 2.0 \times 30 + 4.57 \times 10 = 105.7 \text{ mg/L} \end{aligned}$$

$$W(UOD) = 4MGD \times 105.7 \frac{\text{mg}}{\text{L}} \times 8.34 = 3530 \text{ lb/d}$$

$$\begin{aligned} L_o &= \frac{W(UOD)}{Q_u + Q_e} = \frac{3530 \text{ lb/d}}{(20 \text{ cfs} + 4(1.548) \text{ cfs})5.4} \\ &= 25.0 \frac{\text{mg}}{\text{L}} \end{aligned}$$

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## Problem (cont.)

- Adjust Reaction rates to ambient temp.

$$\begin{aligned} k_a &= k_a(20^\circ\text{C}) \times \theta^{T-20} \\ &= 0.8(1.024)^{(28-20)} = 0.97 \text{ d}^{-1} @ 28^\circ\text{C} \end{aligned}$$

- Determine  $t_{crit}$

$$\begin{aligned} t_{crit} &= \frac{1}{k_a - k_r} \ln \left[ \frac{k_a}{k_r} \left( 1 - \frac{D_o(k_a - k_r)}{k_d L_o} \right) \right] \\ &= \frac{1}{0.97 - 0.40} \ln \left[ \frac{0.97}{0.40} \left( 1 - \frac{0(0.97 - 0.40)}{0.40(25.0)} \right) \right] \\ &= 1.55 \text{ d} \end{aligned}$$

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## Problem (cont.)

- Then get  $x_{crit}$
- And  $c_s$  is: 
$$x = Ut = 0.5 \text{fps} \left( \frac{16.4 \text{mpd}}{\text{fps}} \right) 1.55d = 12.7 \text{mi}$$
  - accounting for temp. & altitude
- Finally the  $c_{min}$  is: 
$$c_s = 6.19 \frac{mg}{L}$$

$$\begin{aligned} c_{min} &= c_s - L_o \frac{k_d}{k_a} e^{-k_r t_{crit}} \\ &= 6.19 - 25.0 \frac{0.40}{0.97} e^{-0.40(1.55)} \\ &= 0.64 \frac{mg}{L} \end{aligned}$$

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## Problem (end)

- Determine allowable load, if WQC require 5.0 mg/L minimum D.O.

- Recognize that the loading:deficit relationship is linear

$$D_{crit} \equiv c_s - c_{min} = L_o \frac{k_d}{k_a} e^{-k_r t_{crit}}$$

Also,  $t_{crit}$  is independent of  $L$  when  $D_o=0$

- so determine allowable  $L$

$$\begin{aligned} \frac{D_{crit}}{L_o} &= \frac{D_{crit(allowable)}}{L_{o(allowable)}} \\ L_{o(allowable)} &= \frac{D_{crit(allowable)} L_o}{D_{crit}} = \frac{1.19(105.7)}{5.55} \\ &= 23 \frac{mg}{L} \end{aligned}$$

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## Additional notes on WLA

Not from Chapra

- Selecting a model
  - number of dimensions
    - usually 1, major gradients are longitudinal, very minor gradients in lateral and vertical directions
    - sometimes 2, deep rivers or river-run impoundments; use should be justified
    - never 3, except research and a few extraordinary cases

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## Additional notes on WLA (cont.)

- Loads, sources & sinks
  - Categorize
    - category I - major sources controlling water quality
      - thorough data collection - temporal variation
    - category II - background sources
      - small to moderate data collection
  - necessary data
    - long-term BOD, with nitrification inhibition
    - analysis of all forms of nitrogen
      - org-N, NH<sub>3</sub>, NO<sub>2</sub>, NO<sub>3</sub>

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## Additional notes on WLA (cont.)

- Time scale
  - steady state
  - quasi-steady state
    - const. loads, constant  $Q$ , diurnal DO variations due to photosynthesis
    - const. loads, variable  $Q$
    - variable loads, constant  $Q$
    - others
  - Fully time-variable analysis

## Additional notes on WLA (cont.)

- Design Conditions
  - $7Q_{10}$  - summer
    - generally endorsed by USEPA
  - Spring Floods - large event
    - storm intensity, sequences, recessional hydrograph
  - Ice cover - winter
- Spatial Extent
  - well into the zone of recovery

## Additional notes on WLA (cont.)

- Dispersion (is it significant?)
    - calculate E
      - from slope
      - from dye studies
    - calculate dimensionless estuary #
    - use Chapra's criteria
      - $n < 0.1$ , advection predominates
      - $n > 10$ , diffusion predominates
    - or calculate reaeration/deoxygenation ratio & use O'Connor figure
- $$\phi = \frac{k_a}{k_d}$$

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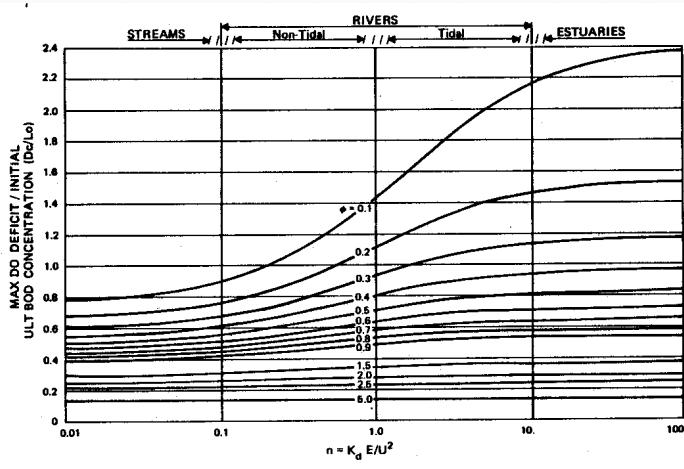
Figure 3-0. Dissolved oxygen response as a function of  $K_d E/U^2$ .

Figure prepared by  
O'Connor  
From: Technical  
Guidance Manual  
for Performing  
Waste Load  
Allocations: Book  
II, Chapter 1

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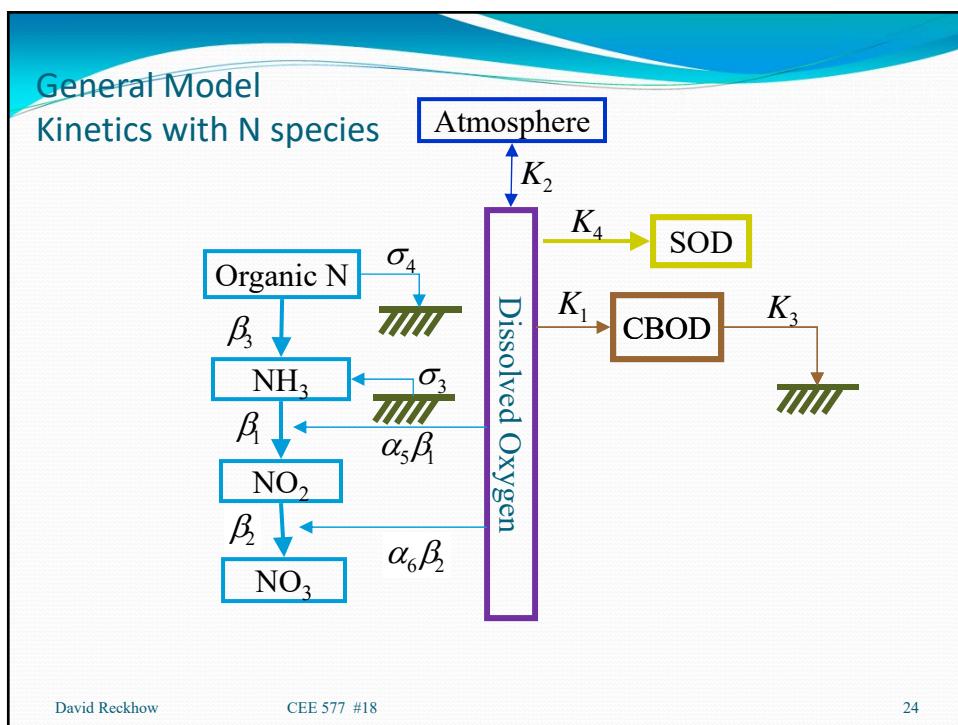
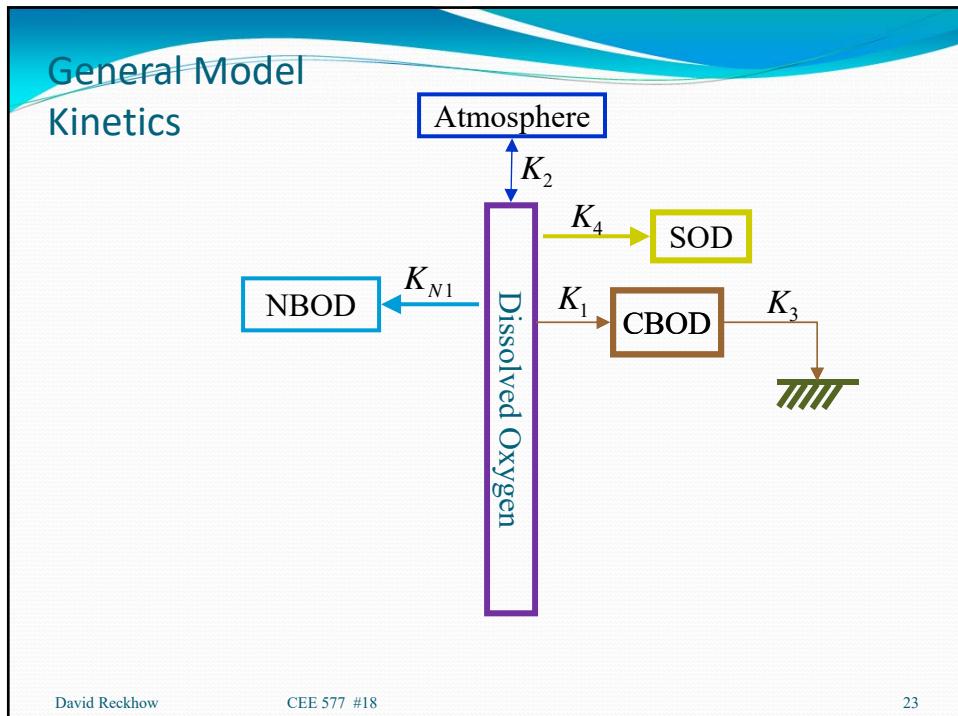
**Full Equation**

$$\begin{aligned}
 D = & D_o e^{-k_a t} + \frac{k_d L_o}{k_a - k_r} (e^{-k_r t} - e^{-k_a t}) \\
 & + \frac{k_n L_{No}}{k_a - k_n} (e^{-k_n t} - e^{-k_a t}) \quad \xrightarrow{\text{Point NBOD}}
 \end{aligned}$$

$$\begin{aligned}
 & - P + R + \left( \frac{S'_B / H}{k_a} \right) (1 - e^{-k_a t}) \\
 & + \frac{k_d S_d}{k_r k_a} (1 - e^{-k_a t}) - \frac{k_d S_d}{k_r (k_a - k_r)} (e^{-k_r t} - e^{-k_a t}) \\
 & + \frac{k_n S_{Nd}}{k_n k_a} (1 - e^{-k_a t}) - \frac{k_n S_{Nd}}{k_n (k_a - k_n)} (e^{-k_n t} - e^{-k_a t}) \quad \xrightarrow{\text{Distributed NBOD}}
 \end{aligned}$$

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- Additional notes on WLA (cont.)**
- Nitrogen modeling
    - NBOD
      - measure and model TKN only
    - all 4 major species
      - org-N, NH<sub>3</sub>, NO<sub>2</sub>, and NO<sub>3</sub>
      - requires separate analysis of loadings, rate coefficients, etc.
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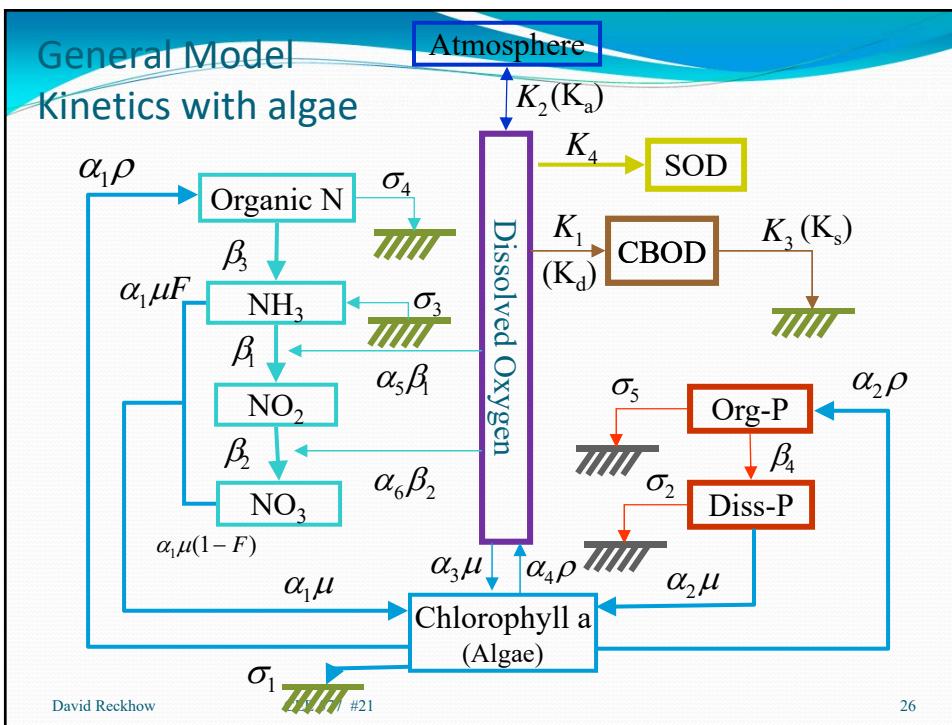
## Additional notes on WLA (cont.)

- Algal modeling
  - Level I
    - measure P-R: diurnal swings in D.O.
  - Level II
    - measure chlorophyll a, light, light extinction, nutrients "in-situ"
    - calculate P-R
  - Level III
    - assess nutrient loadings, light extinction
    - model nutrient conc., chlorophyll a, P-R

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