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# CEE 370 Environmental Engineering Principles

## Lecture #20 Water Resources & Hydrology I: streamflow & water balance

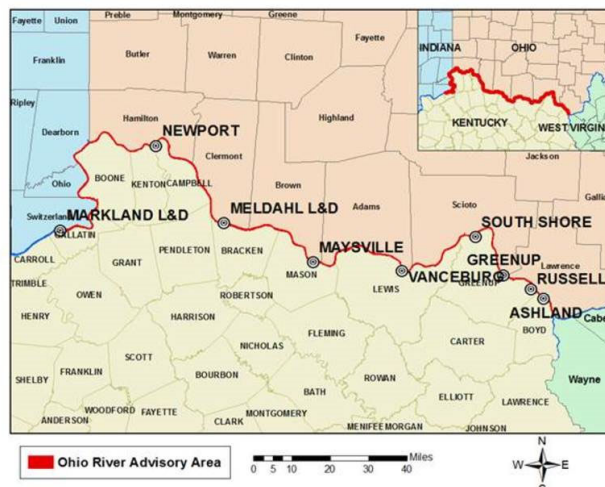
[Reading: Mihelcic & Zimmerman, Chapter 7](#)

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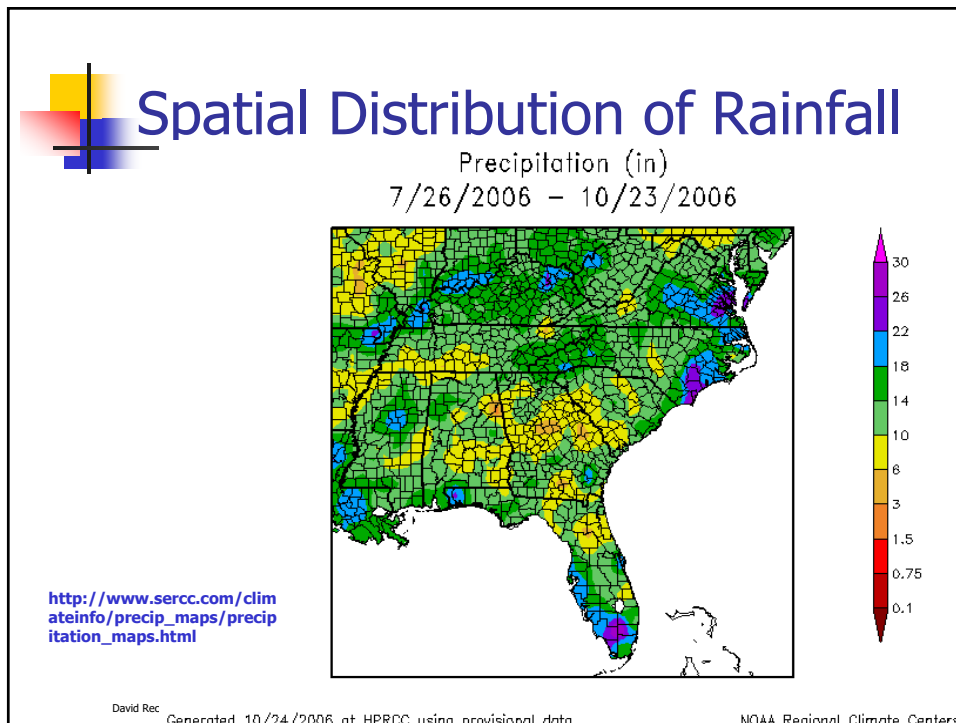
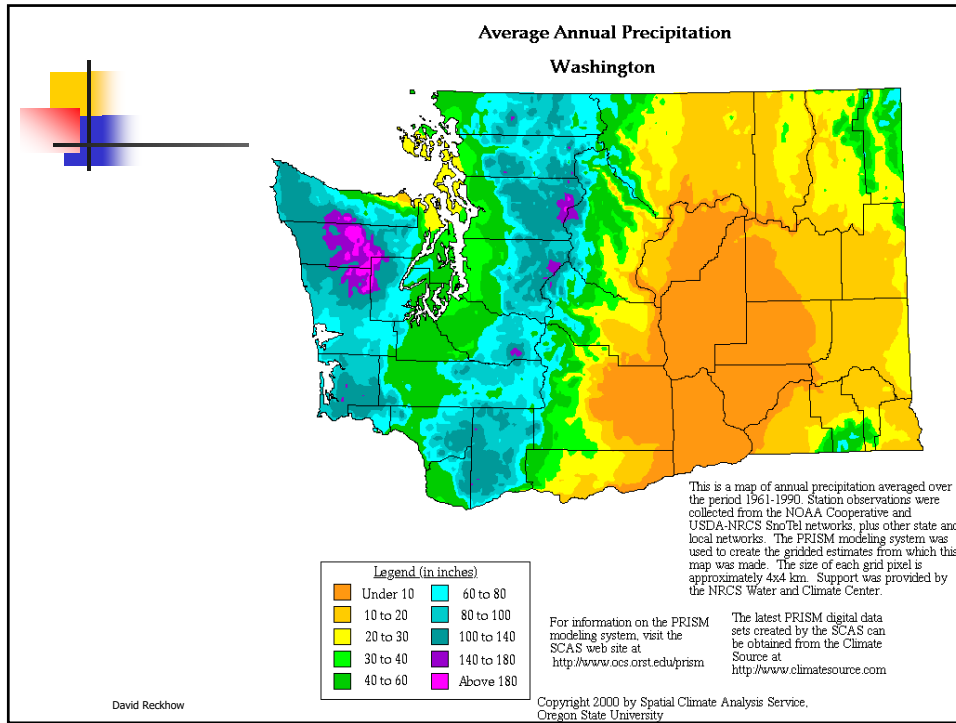
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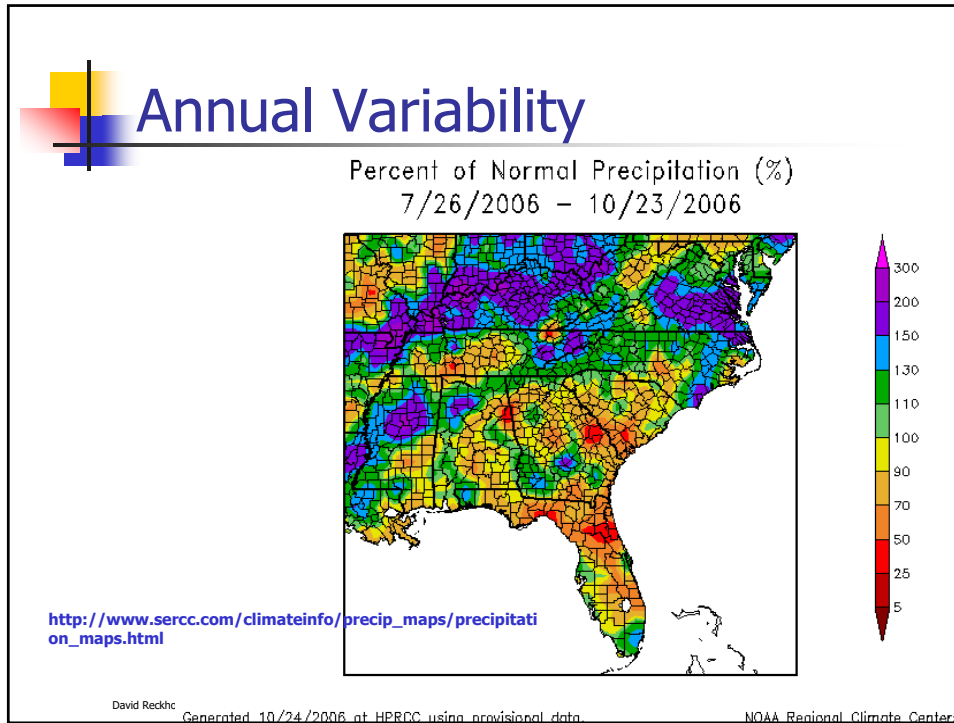
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### ■ Ohio River



2





## Community Water Use

- Table 1 shows on a percent basis the use of water for community systems in the USA. The percentages are average values for USA.
- Public: municipal buildings, pools, etc.
- Loss: unaccounted-for

**Table 1. Types of Community Water Use**

Category	%
Domestic	45
Industrial	24
Commercial	15
Public	9
Loss	7
Total	100

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## Home Use question

- What fraction of total home water use is devoted to showers & baths?
  - A. 10%
  - B. 20%
  - C. 30%
  - D. 40%
  - E. 50%

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## Home water use

- Table 2 shows the percent indoor use for the domestic category.
  - These data are average values from a survey (year 1998) for Boulder, CO; Denver, CO; Eugene, OR; Seattle, WA; San Diego, CA; Phoenix, AZ; Tempe and Scottsdale, AZ; Waterloo, Ontario; Walnut Valley Water District, CA; Municipal Water District, CA; and Lumpoc, CA
  - For these communities, the average indoor use was **71** gallons per capita per day (gpcd) and outdoor use was **101** gpcd for total domestic water use of **172** gpcd. You would expect much lower domestic water use in the Northeast because of less outdoor water use.
  - Northeast domestic water use is about **100 gpcd**.

Table 2

Compare with M&Z, Table 7.8

Category	%
Flushing Toilets	27
Washing Clothes	22
Shower/Bath	19
Faucet	16
Leak	14
Other	2

Y=37X+69

Best Targets for reduced use

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## Summation

- Summary: for design of public water facilities we are interested in the following demands:
  - Average Daily Demand/flow
  - Maximum Daily Demand/flow
  - Peak Hourly Demand/flow
  - Fire Demand
  - Inflow

$Q_{design} = Q_{avg} \times PF$

For PFs see M&Z Table 7.14 McGuire, 1991

**Diurnal Demand Hydrograph**

- **Hourly Variation in Water Demand on the Maximum Day**

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## Hydraulics of water systems

- Used to size hydraulic aspects of water systems
  - Under economic and various physical constraints
  - Focus: transmission mains, distribution storage, distribution pipe network
  - Relate: flow (or velocity), pipe diameter, roughness, pipe length, head loss

Transmission main      WTP      Distribution Storage and Distribution main      Distribution System

## Pump Head ( $h_T$ )

- $h_T(Q)$  = energy (head) that must be supplied to achieve desired  $Q$  (= system head)

$$h_T = h_s + h_f + h_m$$

net static lift (elevation difference)  
=  $z_{\text{discharge}} - z_{\text{suction}}$

friction losses on long straight pipe (intake and discharge)  
=  $f_n(Q, d)$

minor losses for pipe system entrance, pump station elements, exit

Compare with M&Z figure 7.20

## Multiple Pumps

- Parallel operation (a)
- Head-discharge curves for various combinations (b)

H&H, Figs 4-17, pg 109

(a)

(b)

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## T/F Question

- Consider 2 cities of the same size, both having the same maximum day water demands, and both pumping at that rate for 24 hours.
- The city with the more uniform hourly water demand will have higher system storage needs

A. True  
B. False

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## Demand Hydrograph

- analysis for 24 hr cycle

Area in shaded portion converted to volume is equal to the required storage of 1.47 mg

Tank is filling

Tank is draining

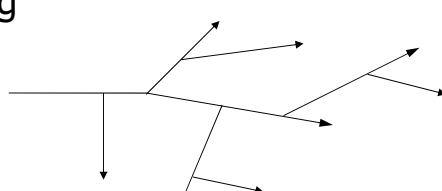
24-hr pumping rate = average hourly demand rate = 4775 gpm

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Compare with M&Z figure 7.19

## Pipe Patterns I

- Branching



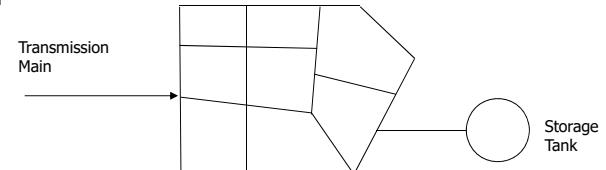
- Avoid this system except where necessary such as on the outskirts of a community
- Have "dead ends" where water may be stagnant and lead to water quality problems
- When a pipe break occurs, isolating break leads to interruption of service to the area beyond the break (only one path to a point of use)

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Compare with M&Z figure 7.19

## Pipe Patterns II

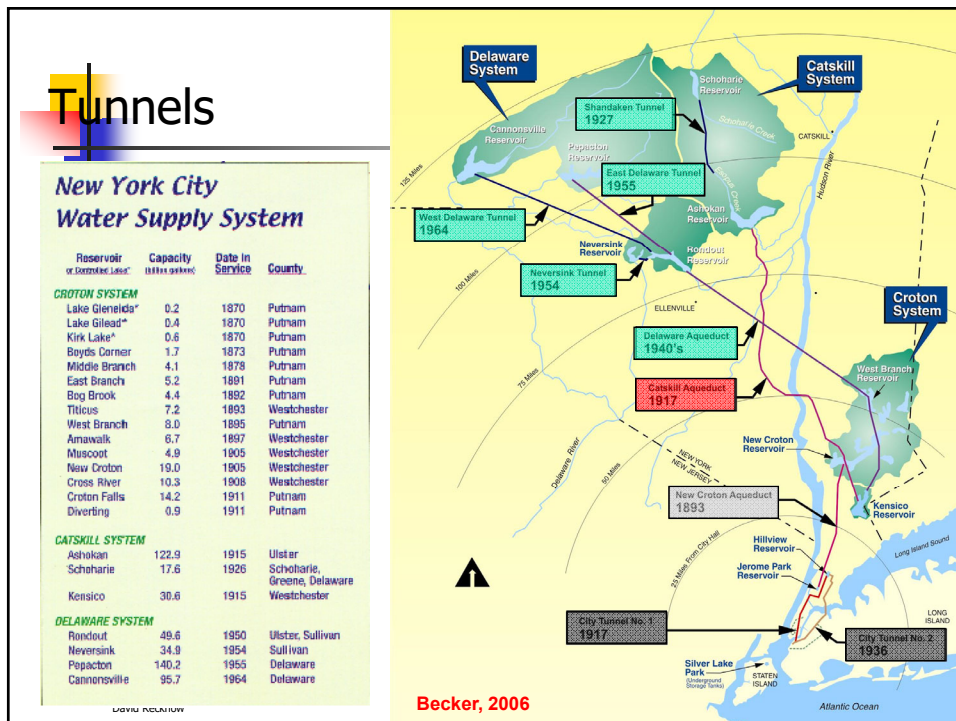
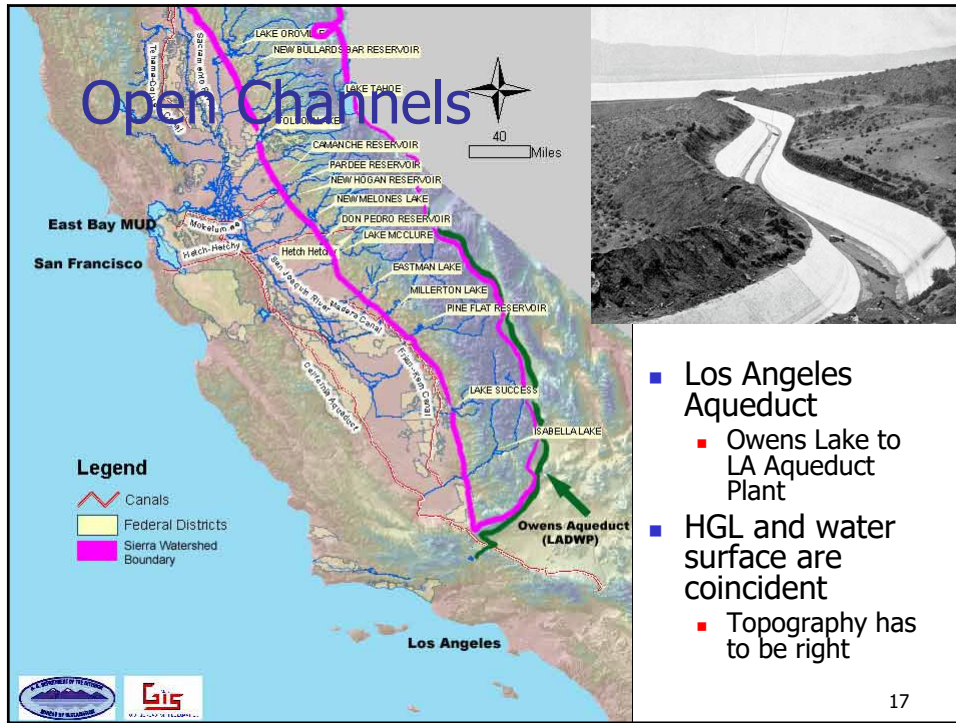
- Grid



- Head loss is minimized by multiple parallel pipe paths
- Can isolate breaks and maintain service to most of water system due to parallel routes
- Avoids dead ends and deterioration in water quality which can occur at dead ends
- 6 inch minimum diameter for pipe in grid system (8 inch for dead end pipe)

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## NYC Tunnel

- Well suited for mountain terrain or river crossings
  - An arch is constructed to prepare the tunnel to be lined with concrete.



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**Videos**

Tunnel #3 intro  
<https://www.youtube.com/watch?v=YWwgcBodAFo>

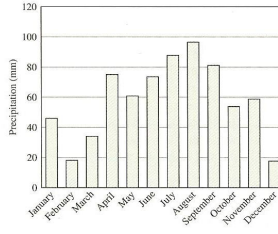
Tunnel #3: sandhogs (1:32)  
<https://www.youtube.com/watch?v=dShvdsRTNtY>

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## Rainfall: temporal variation

**FIGURE 6-9**

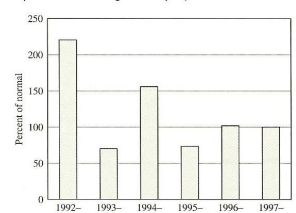
Seasonal variations in precipitation in the Michigan-Huron basin (for 1994). [Source: Data from Yee, P. A Report: On the 1994 Water Levels of the Great Lakes and St. Lawrence River. Environment Canada, Cornwall, Ontario, 1995.]



Month	Precipitation (mm)
January	45
February	20
March	35
April	75
May	60
June	70
July	95
August	100
September	80
October	55
November	60
December	20

**FIGURE 6-10**


Annual variations in precipitation in the Mojave Desert. [Source: Hereford, R., C. Longpré. Climate History of the Mojave Desert Region, Including Data from 48 Long-term Weather Stations and an Overview of Regional Climate Variation. Flagstaff, Arizona. <http://wgsc.wr.usgs.gov/mojave/climate-history> <http://atmos.washington.edu/pdo>]



Year	Percent of normal
1992-1993	220
1993-1994	70
1994-1995	150
1995-1996	75
1996-1997	100
1997-1998	100

implications for water resource management. As shown in Figure 6-9, monthly precipitation rates in the Michigan-Huron basin varied by a factor of almost 5. This phenomenon is not unusual. Yearly variations can also be significant as shown in Figure 6-10. Here variations in precipitation rates vary as much as threefold. These yearly variations make it important to design reservoirs that are adequate during years of low rainfall and dams that have the capacity to ensure adequate flood control even in times of high precipitation rates.

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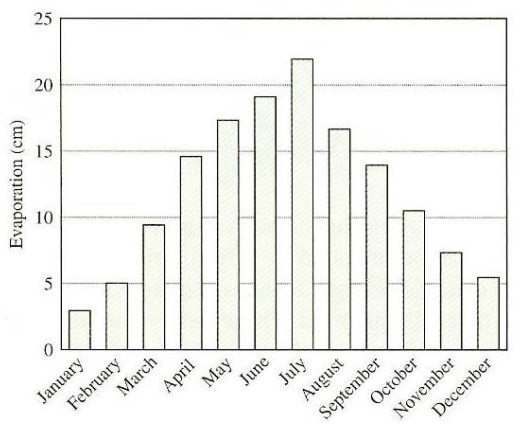


# Evaporation

\*Stochastic means that the occurrence of floods can be predicted from flood data that have been collected over time.


**FIGURE 6-11**

Average monthly pan evaporation rates for Athens University of Georgia (for 1970–1971). (Source: Data courtesy of South Carolina State Climatology Office [http://water.dnr.state.sc.us/climate/sco/pan\\_tables.html](http://water.dnr.state.sc.us/climate/sco/pan_tables.html))



Month	Evaporation (cm)
January	3
February	5
March	9
April	14
May	17
June	19
July	22
August	16
September	14
October	10
November	7
December	5

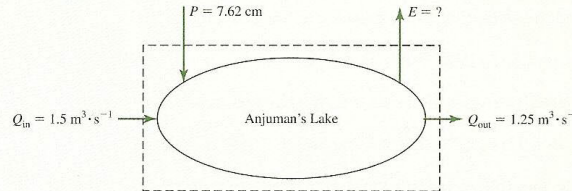
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# Example 3

**EXAMPLE 6-3** Anjuman’s Lake has a surface area of 70.8 ha. For the month of April the inflow was  $1.5 \text{ m}^3 \cdot \text{s}^{-1}$ . The dam regulated the outflow (discharge) from Anjuman’s Lake to be  $1.25 \text{ m}^3 \cdot \text{s}^{-1}$ . If the precipitation recorded for the month was 7.62 cm and the storage volume increased by an estimated  $650,000 \text{ m}^3$ , what is the estimated evaporation in cubic meters and centimeters? Assume that no water infiltrates into or out of the bottom of Anjuman’s Lake.


**Solution** Begin by drawing the mass-balance diagram.



The mass-balance equation is.

Accumulation = input – output

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The accumulation (change in storage) is  $650,000 \text{ m}^3$ . The input consists of the inflow and the precipitation. The product of the precipitation depth and the area on which it fell (70.8 ha) will yield a volume. The output consists of outflow plus evaporation. The change in storage can be represented by the equation

$$\Delta S = Q_{in} + P - E - Q_{out}$$

Make sure that all parameters are in the same units. The flow rates are expressed in cubic meters per second whereas  $E$  and  $P$  are shown in centimeters. Because we want to calculate the change in storage, we should convert all units to either units of volume per month ( $\text{m}^3 \cdot \text{month}^{-1}$ ) or units of length per month ( $\text{cm} \cdot \text{month}^{-1}$ ). Although hydrologists often calculate changes in storage in units of length per unit time, you should recognize that length is not conserved, rather mass (and therefore, volume, assuming a constant density) is. As such, we will solve the problem in units of volume and then calculate the change in depth. Remember also that April has 30 days.

Therefore,

$$650,000 \text{ m}^3 = (1.5 \text{ m}^3 \cdot \text{s}^{-1})(30 \text{ days})(86,400 \text{ s} \cdot \text{day}^{-1})$$

$$+ (7.62 \text{ cm})(70.8 \text{ ha})(10^4 \text{ m}^2 \cdot \text{ha}^{-1})(1 \text{ m} \cdot 100 \text{ cm}^{-1})$$

$$- (1.25 \text{ m}^2 \cdot \text{s}^{-1})(30 \text{ days})(86,400 \text{ s} \cdot \text{day}^{-1}) - E$$

Solving for  $E$ , we obtain

$$E = Q_{in} + P - Q_{out} - \Delta S$$


$$= 3.89 \times 10^6 \text{ m}^3 + 5.39 \times 10^4 \text{ m}^3 - 3.24 \times 10^6 \text{ m}^3 - 6.50 \times 10^5 \text{ m}^3$$

$$= 5.39 \times 10^4 \text{ m}^3$$

For an area of 70.8 ha, the evaporation rate in units of depth per month is

$$E = \frac{5.39 \times 10^4 \text{ m}^3}{(70.8 \text{ ha})(10^4 \text{ m}^2 \cdot \text{ha}^{-1})} = 0.076 \text{ m} = 7.6 \text{ cm} \cdot \text{month}^{-1}$$


23



## Estimating Evaporation

- Pan Evaporation
  - Land: direct measurement
  - Lake: multiply pan evaporation by 0.7
- Correlations: semi-empirical
  - Based on
    - Saturation vapor pressure ( $e_s$ ) in kPa
    - Vapor pressure in overlying air ( $e_a$ ) in kPa
    - Wind speed ( $u$ ) in m/s
  - Dalton's Equation  $E = (e_s - e_a)(a + bu)$
  - Lake Hefner Equation  $E = 1.22(e_s - e_a)u$

24



## Example 4

**EXAMPLE 6-4** During April, the wind speed over Anjuman's Lake was estimated to be  $4.0 \text{ m} \cdot \text{s}^{-1}$ . The air temperature averaged  $20^\circ\text{C}$ , and the relative humidity was 30%. The water temperature averaged  $10^\circ\text{C}$ . Estimate the evaporation rate using the Dalton's equation.

**Solution** From the water temperature and Table 6-2, the saturation vapor pressure is estimated as  $e_s = 1.227 \text{ kPa}$ . The vapor pressure in the air may be estimated as the product of the relative humidity and the saturation vapor pressure at the air temperature.

$$e_a = (2.337 \text{ kPa})(0.30) = 0.70 \text{ kPa}$$


The daily evaporation rate is then estimated to be

$$E = 1.22(1.227 - 0.70)(4.0 \text{ m} \cdot \text{s}^{-1}) = 2.57 \text{ mm} \cdot \text{day}^{-1}$$

The monthly evaporation would then be estimated to be

$$E = (2.57 \text{ mm} \cdot \text{day}^{-1})(30 \text{ days}) = 76.8 \text{ mm, or } 7.7 \text{ cm}$$

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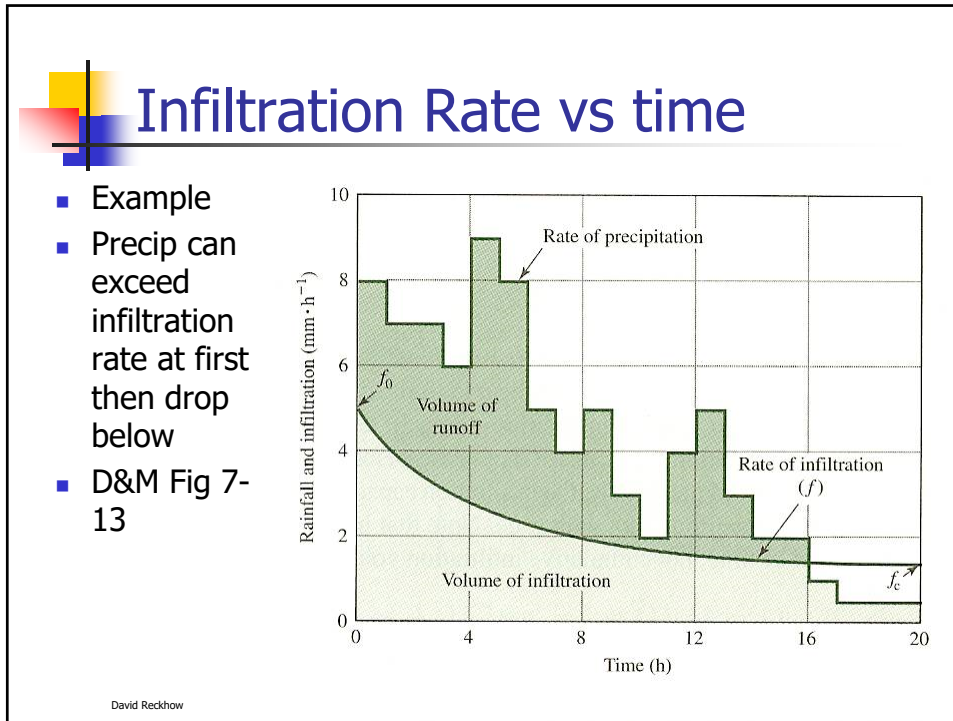


## Vapor Pressure

Temperature (°C)	Vapor Pressure (kPa)	Temperature (°C)	Vapor Pressure (kPa)
0	0.6113	25	3.1690
5	0.8726	30	4.2455
10	1.2281	35	5.6267
15	1.7056	40	7.3814
20	2.3388	50	12.344

Source: Lide, D. R. editor-in-chief, *CRC Handbook of Chemistry and Physics*. 76th ed., CRC Press, New York, (1995), pp. 6-15.

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## Example 5

**EXAMPLE 6-5** A soil has the following characteristics

$$f_0 = 3.81 \text{ cm} \cdot \text{h}^{-1} \quad f_c = 0.51 \text{ cm} \cdot \text{h}^{-1} \quad k = 0.35 \text{ h}^{-1}$$

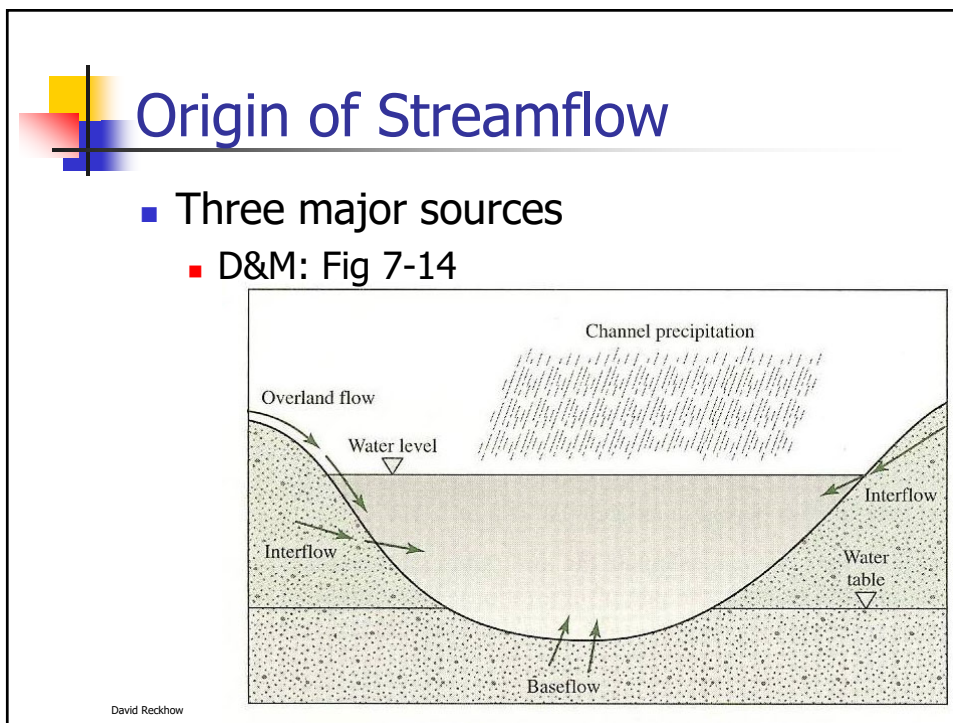
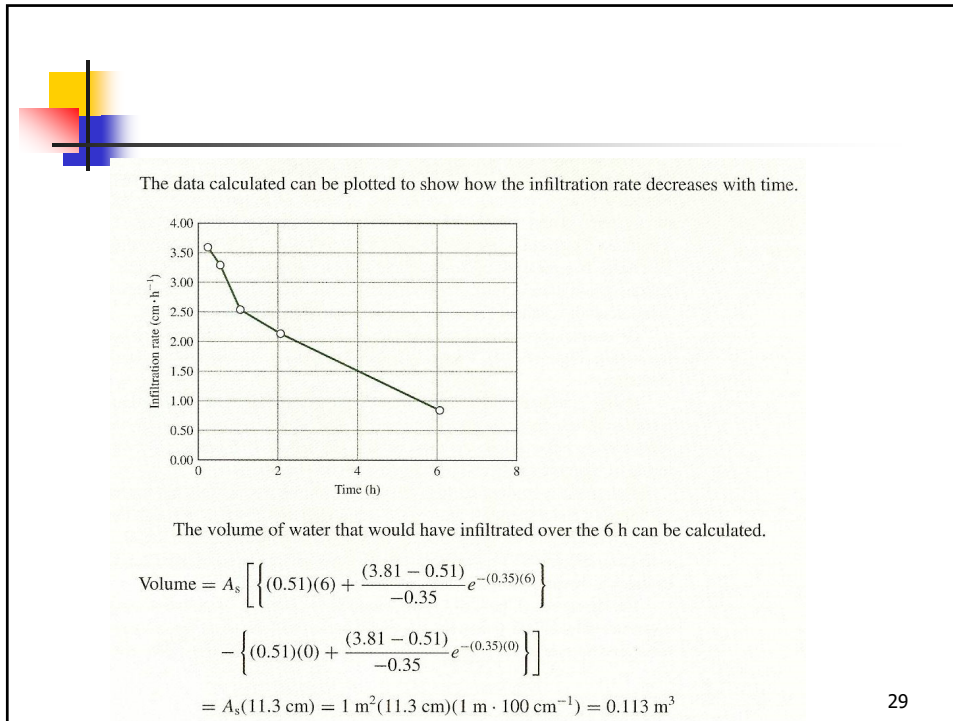
What are the values of  $f$  at  $t = 12 \text{ min}$ ,  $30 \text{ min}$ ,  $1 \text{ h}$ ,  $2 \text{ h}$ , and  $6 \text{ h}$ ? What is the total volume of infiltration over the 6-h period in an area that is  $1 \text{ m}^2$ ?

**Solution** Using the stated data, we can calculate the infiltration rates using Horton's equation if  $i > f$  (or rate of precipitation exceeds the rate of infiltration). The volume of precipitation that infiltrated can be calculated by integrating Horton's equation over the time interval being considered.

$$\text{Volume} = A_s \int_0^t f \, dt = A_s \int_0^t [f_c + (f_0 - f_c)e^{-kt}] \, dt = A_s \left[ f_c t + \frac{(f_0 - f_c)}{-k} e^{-kt} \right]_0^t$$

Using Horton's equation, we calculate the infiltration rate for each of the desired time intervals.

Time (h)	Infiltration Rate (cm/h)
0.2	3.58
0.5	3.28
1	2.54
2	2.16
6	0.91



## Watershed & Hydrogeometric Parameters

- Geometry
  - Width and Depth
  - Slope
- Hydrology
  - Velocity and Flow
  - Mixing characteristics (dispersion)
- Drainage Area
- Dams, Reservoirs & flow diversions
- Geographical location of basin

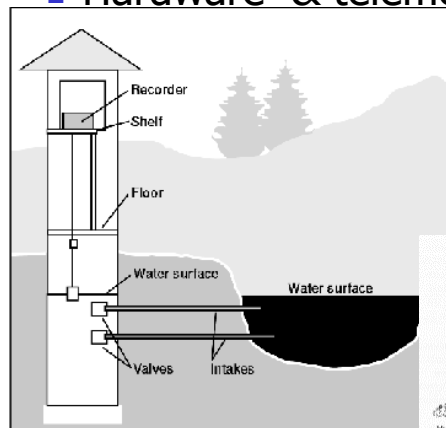
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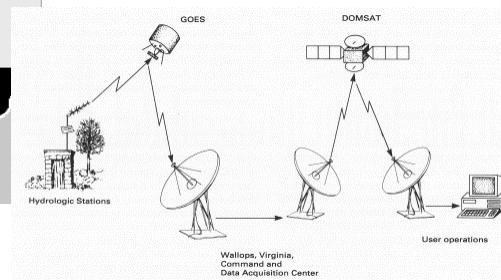
31

## USGS Gaging Stations

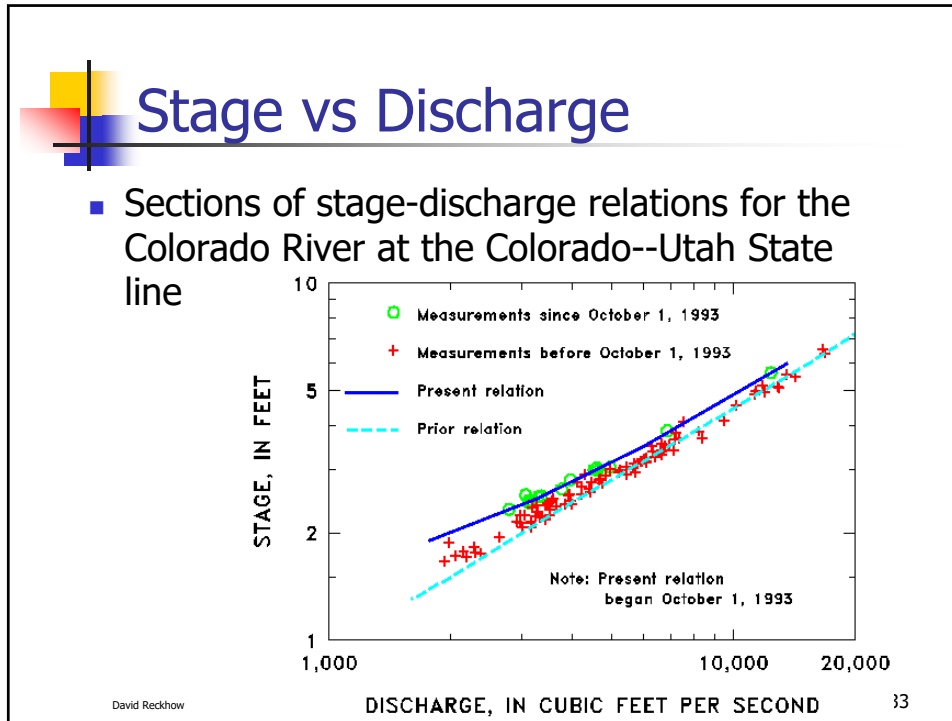
- Hardware & telemetry



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## Mass Transport Processes

- Processes that move chemicals through the air, surface water, subsurface environment or engineered systems
  - e.g., From point of generation to remote locations
- Very important to:
  - design of treatment systems
  - prediction of pollutant impacts in the environment
  - determination of waste load allocations
  - determination of sources of pollutants.


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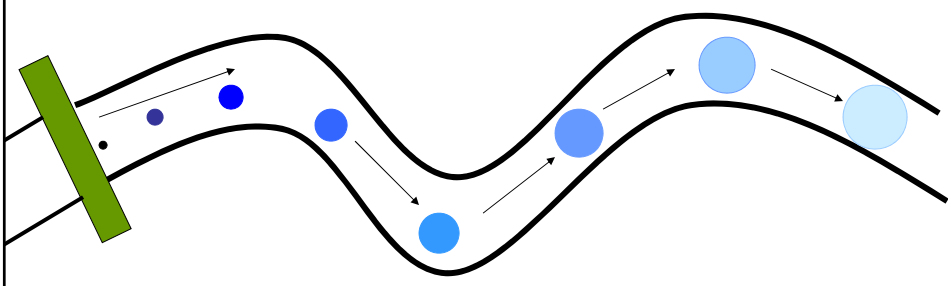
## Advection and Dispersion

- Advection
  - Transport with the mean fluid flow
- Dispersion
  - Transport in directions other than that of the mean fluid flow
    - Some is due to "random" motions

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- Blue dye dropped in a flowing river



- Dispersion occurs along with clear advection

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## Assessing Hydrogeometry

- Point Estimates vs. Reach Estimates
- Flow
  - often requires velocity
  - May use stage
    - USGS gaging stations

$Q = UA_c$


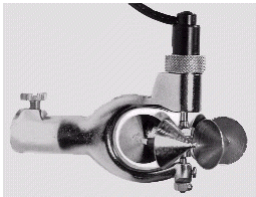

$U = \frac{Q}{A_c}$


- Velocity
  - Current Meter
  - Weighted Markers or Dye

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## Current Meters

- Price
- Pygmy




<http://advnmc.com/Rickly/currmet.htm>

<http://www.swoffer.com/2200.htm>

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38

## Current Meter Deployment

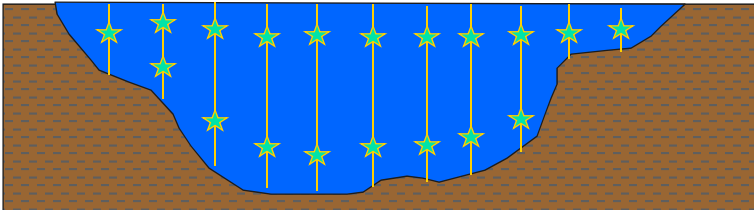
- Current meter and weight suspended from a bridge crane
- Wading rod and current meter used for measuring the discharge of a river



39

## Current Meter Method


- Divide stream cross section into transects
- Measure velocity in each with meter
  - at 60% depth in shallow water (<2ft)
  - or 20% and 80% depth in deep water



40

## Deployment cont.

- Crane, current meter, and weight used for measuring the discharge of a river from a bridge



From: U.S. GEOLOGICAL SURVEY CIRCULAR 1123; on the www at:  
<http://h2o.usgs.gov/public/pubs/circ1123/index.html>

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## Moving Marker Methods

- Best for low velocity (<0.2 ft/s)
- Several types
  - Drogues (current at depth)
  - Dye (mixing too)
  - Surface objects (Oranges, Frisbees)
- Velocity from change in location with time

$$U_{avg} = \frac{\Delta x}{t^*}$$

Time of travel

$$Q_{avg} = U_{avg} \left( \frac{A_1 + A_2}{2} \right)$$

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## Drogues

- Designed to move with the current at a specific depth
- Surface float with a plastic underwater sail set at a predetermined depth

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## Assumptions

- The ideal reactor is:
  - Completely mixed in cross-section
  - Without longitudinal (downstream) mixing
  - Steady State

Top view

Cross-section

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## Dye studies

Tracer Dye Injected Here

Stream Discharge Measurement -- Example of Tracer Dilution Technique

Water Sample Taken Here

Partial Mixing

Total Mixing

Water Surface

Stream Plan View

Stream Cross-Section View

Stream Profile View

Note: Gray-scale changes in the dye plume represent dye dilution and mixing, not color. The small amount of dye added to the stream flow is imperceptible to the eye and can only be detected by laboratory analysis of water sample.

Drawing courtesy of R. D. Mac Nish, University of Arizona, Tucson ([http://www.tucson.ars.ag.gov/salsa/research/research\\_1997/AMS\\_Posters/gw-sw\\_interactions/gw-sw\\_fl.html](http://www.tucson.ars.ag.gov/salsa/research/research_1997/AMS_Posters/gw-sw_interactions/gw-sw_fl.html))

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## Lateral Mixing: USGS guidance

- Lateral or transverse dispersion coefficient for a stream:

$$E_{lat} = 0.6HU^*$$

Mean depth

Shear velocity
- Length required for complete mixing:

Center discharge:

$$L_m = 0.10U \frac{B^2}{E_{lat}}$$

Width

$$U^* = \sqrt{gHS}$$

**For Fort River**  
~1000 ft  
or t=20 min

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## Liquid Water Transport

- Advection: unidirectional flow
- Diffusion: movement of mass that is not unidirectional flow; usually movement in an unorganized fashion
  - Dispersion
  - Eddy Diffusion
  - Molecular Diffusion

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## Mass Diffusion

Incorporates molecular movement and interfacial area

Bulk Diffusion (m<sup>2</sup>/yr)

$$V_1 \frac{dc_1}{dt} = D'(c_2 - c_1)$$

Concentration Gradient

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## Fick's First Law

- Mass flux is proportional to the concentration gradient and a diffusion coefficient

$$J_x = -D \frac{dc}{dx}$$

Units for diffusion coefficient:  
(Length<sup>2</sup>time<sup>-1</sup>)

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## Bulk Diffusion Coefficient

$V_1, c_1$

$V_2, c_2$

$V_1 \frac{dc_1}{dt} = -JA_c$

The mixing length

$J_x = -D \frac{dc}{dx}$

$\frac{dc}{dx} \cong \frac{c_2 - c_1}{\ell}$


And combining all three:

$$V_1 \frac{dc_1}{dt} = \underbrace{\frac{DA_c}{\ell}}_{D'} (c_2 - c_1)$$

$E' = \frac{EA_c}{\ell}$

Similar for Eddy Diffusion


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## Some diffusion coefficients

Compound	Temp ( C)	D (cm <sup>2</sup> s <sup>-1</sup> )
Methanol in H <sub>2</sub> O	15	1.26x10 <sup>-5</sup>
Ethanol in H <sub>2</sub> O	15	1.00x10 <sup>-5</sup>
Acetic Acid in H <sub>2</sub> O	20	1.19x10 <sup>-5</sup>
Ethylbenzene in H <sub>2</sub> O	20	0.81x10 <sup>-5</sup>
CO <sub>2</sub> in Air	20	0.151

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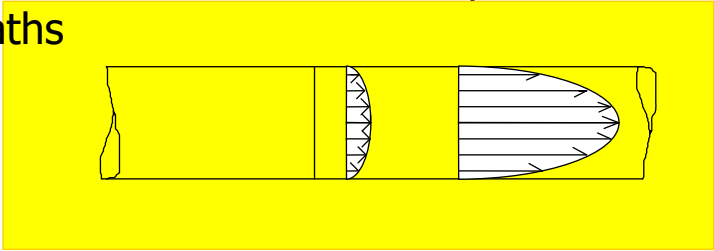
## Turbulent Dispersion

- Turbulent eddies
  - Large scale “random movement”
    - Whirlpools in a river
    - Circulatory flows in the ocean
  - Occurs only at flows above a “critical” level
    - Determined by the Reynolds number
  - Almost always dominates over molecular diffusion
    - Exception: transport across a boundary

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## Dispersion (Mechanical)

- Differences in velocities of parallel flow paths



- Different paths in porous media

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## To next lecture

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