





















Rank	Chemical	Use	Chemical Formula
1	Dichloromethane	Paint stripping, solvent degreaser, blowing agent in foams	CH ₂ Cl ₂
2	Trichloroethene	Dry cleaning agent, metal degreaser solvent	C ₂ Cl ₃ H
3	Tetrachloroethene	Dry cleaning, metal degreaser, solvent, paint remover	C ₂ Cl ₄
4	trans 1,2- Dichloroethene	Solvent, additive to lacquer, low- temperature solvent for caffeine	$C_2H_2Cl_2$
5	Chloroform	Solvent, electronic circuit manufacturing	CHCl ₃
6	l,1-Dichloroethane	Paint and varnish remover, metal degreaser, ore flotation	C ₂ C1 ₂ H ₄
7	1,1-Dichloroethene	Paint and varnish remover, metal de- greaser	C ₂ C1 ₂ H ₂
8	1,1,1-Trichloroethane	Solvent	C ₂ Cl ₃ H ₃
9	Toluene	Gasoline component, solvent thinner, adhesive solvent	C ₇ H ₈
10	1,2-Dichloroethane	Paint and varnish remover, metal degreaser, fumigant	C ₂ C ₁₂ H ₄

benzene	Component of gasoline, used in chemical synthesis Used in styrene manufacturing, solvent, asphalt construction	C ₆ H ₆ C ₈ H ₁₀
benzene	Used in styrene manufacturing, solvent, asphalt construction	C ₈ H ₁₀
ol	D 1 1 1 1 1 1	
1	Disinfectant, pharmaceutical aid	C ₆ H ₅ OH
obenzene	Used in chemical synthesis	C ₆ H ₅ Cl
chloride	Refrigerant, used in plastics industry	C ₂ ClH ₃
on tetrachloride	Dry cleaning, metal degreasing, veterinary medicine	CCl ₄
- hexyl)phthalate	Used in vacuum pumps	C ₂₄ H ₃₈ O ₄
thalene	Used in manufacturing mothballs and motor fuel, component of coal tar	C ₁₀ H ₈
-Trichloroethane	Solvent	C ₂ Cl ₃ H ₃
oethane	Refrigerant, solvent, used to produce tetraethyl lead	C ₂ ClH ₅
. T	richloroethane ethane	Trichloroethane Solvent ethane Refrigerant, solvent, used to produce tetraethyl lead

How caphysica	ledge of rties?			
Table 1-2. Propertie	es of Selecte	ed Chlorinated A	liphatic Hyd	drocarbons*
Chemical	Vapor Pressure (mmHg)	Henry's Constant (atm-m ³ /mole)	Water Solubility (mg/L)	Chemical Half-life (Years)
Chemical Carbon tetrachloride	Vapor Pressure (mmHg) 90	Henry's Constant (atm-m ³ /mole) 0.0294	Water Solubility (mg/L) 785	Chemical Half-life (Years) 16–41
Chemical Carbon tetrachloride Chloroform	Vapor Pressure (mmHg) 90 160	Henry's Constant (atm-m ³ /mole) 0.0294 0.0040	Water Solubility (mg/L) 785 8,200	Chemical Half-life (Years) 16-41 742-3,000
Chemical Carbon tetrachloride Chloroform Tetrachloroethene	Vapor Pressure (mmHg) 90 160 14	Henry's Constant (atm-m ³ /mole) 0.0294 0.0040 0.0268 0.0117	Water Solubility (mg/L) 785 8,200 150	$\begin{array}{c} \text{Chemical} \\ \text{Half-life} \\ (\text{Years}) \end{array} \\ \hline 16-41 \\ 742-3,000 \\ 3.8 \times 10^8 - 9.9 \times 10^8 \\ 4.0 \times 10^5 + 1.2 \times 10^6 \end{array}$
Chemical Carbon tetrachloride Chloroform Tetrachloroethene Trichloroethene Vinyl chloride	Vapor Pressure (mmHg) 90 160 14 60 2,660	Henry's Constant (atm-m ³ /mole) 0.0294 0.0040 0.0268 0.0117 0.0224	Water Solubility (mg/L) 785 8,200 150 1,100 2,700	$\begin{array}{c} \mbox{Chemical} \\ \mbox{Half-life} \\ \mbox{(Years)} \end{array} \\ \hline 16-41 \\ 742-3,000 \\ 3.8 \times 10^8 - 9.9 \times 10^8 \\ 4.9 \times 10^5 - 1.3 \times 10^6 \\ > 10 \end{array}$
Chemical Carbon tetrachloride Chloroform Tetrachloroethene Trichloroethene Vinyl chloride From Barbee, 1994.	Vapor Pressure (mmHg) 90 160 14 60 2,660	Henry's Constant (atm-m ³ /mole) 0.0294 0.0040 0.0268 0.0117 0.0224	Water Solubility (mg/L) 785 8,200 150 1,100 2,700	$\begin{array}{c} \text{Chemical} \\ \text{Half-life} \\ (\text{Years}) \end{array} \\ \hline 16-41 \\ 742-3,000 \\ 3.8 \times 10^8 - 9.9 \times 10^8 \\ 4.9 \times 10^5 - 1.3 \times 10^6 \\ > 10 \end{array}$

		F	r
	Field	Journal	Publisher
	Environmental quality	Environmental Science and Technology	American Chemical Society
	-	Water Resources Research	American Geophysical Union
Ŭ		Water, Air and Soil Pollution	Kluwer Academic Publications
n	Water treatment	Journal of the American Water Works Association	American Water Works Association
0		Aqua	International Water Assn.
Ň		Journal of the Environmental Enaineering Division	American Society of Civil Engineers
U U	Wastewater treatment	Water Environment Research	Water Environment Federation
atic		Journal of the Environmental Engineering Division	American Society of Civil Engineers
	Solid waste	BioCycle	J. G. Press, Inc.
4	Hazardous waste	Hazardous Waste and Hazardous Materials	Mary Ann Liebert, Inc.
ō		Ground Water	Ground Water Publications, Inc.
IJ	Air pollution and control	Journal of the Air and Waste Management Association	Air and Waste Management Association
	General	Chemical and Engineering News	American Chemical Society
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_	Andrew W	heeler
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Region 1 Boston Region 2 New York Philadelphia	Region 4 Atlanta Chicago	
Region 6 Dallas Kansas City Denver	Region 9 San Francisco Seattle	















Cor	ntrolling Air Pol	lution in Cities		
Source	Pollutants	Methods of Control		
Industries	Volatile organics	Require reduced emissions		
	Volatile chlorofluorocarbons	Require reduced emissions		
	Particulate inorganics	Require reduced emissions		
Automobiles	Hydrocarbons	Improved discharge nozzles at filling stations, improved ventilation within the gasoline tank		
	Products of incomplete combustion	Improved combustion by requiring improved combustion efficiency (auto manufacturer), regular engine maintenance by requiring vehicle emission testing, requiring gasoline stations to provide only oxygenated fuels.		
	Chlorofluorocarbons from air conditioners	Require the redesign of the air conditione so that future automobiles can use other refrinerants		









All increase chance of deat	h in any year by 0.000001							
Smoking 1.4 cigarettes	Cancer, heart disease							
Spending 1 hr. in a coal mine	Black lung disease							
Living 2 days in NYC or Boston	Air pollution							
Living 2 months in Denver	Cancer caused by cosmic radiation							
One chest X-ray	Cancer caused by radiation							
Eating 40 tbs. of peanut butter	Liver cancer caused by Aflatoxin B							
Drinking 30 12-oz. cans of diet soda	Cancer caused by saccharin							
Living 150 yrs. within 20 miles of a nuclear power plant	Cancer caused by radiation							







1	1 H Hydrogen 1 x 10 ⁻¹	2				ELI	EMEN	ITS C)F TH	1E HU	JMAI	N BO	DY	14	15	16	17	2 He Helum
2	3 Lithium 3.1 x 10 ⁻⁸	4 Be Berylliam		Mass Fraction = 10 ^{Color}							5 Boron 6.90 x 10 ⁻⁷	6 C Carbon 1.8 x 10 ⁻¹	7 N Nitrogen 3 x 10 ⁻²	8 0 0xygen 6.5 x 10 ⁻¹	9 F Fluorine 3.2 x 10 ⁻⁵	10 Ne Neon		
3	11 Na Sodium 1.5 x 10 ⁻³	12 Mg Magnesium 5.00 x 10 ⁻⁴	3	4	5	6	7	8	9	10	11	12	13 Aluminum 8.70 x 10 ⁻⁷	14 Silicon 2 x 10 ⁻⁵	15 P Phosphorus 1.1 x 10 ⁻²	16 S Sulfur 2.5 x 10 ⁻³	17 Cl Chiorine 1.5 x 10 ⁻³	18 Ar Argon
4	19 K Potassium 2 x 10 ⁻³	20 Ca Calcium 1.4 x 10 ⁻²	21 Sc Scandium	22 Ti ^{Titanium} 1.30 x 10 ⁻⁷	23 V Vanadium 2.60 x 10 ⁻⁷	24 Cr ^{Chromium} 2.4 x 10 ⁻⁸	25 Mn Manganese 1.70 x 10 ⁻⁷	26 Fe	27 Co ^{Cobalt} 2.1 x 10 ⁻⁸	28 Ni Nickel 1.40 x 10 ⁻⁷	29 Cu Copper 1 x 10 ⁻⁶	30 Zn ^{Zinc} 3.2 x 10 ⁻⁵	31 Ga Gallium	32 Germanium	33 As Arsenic 2.60 x 10 ⁻⁷	34 Seenium 1.90 x 10 ⁻⁷	35 Br Bromine 2.9 x 10 ⁻⁶	36 Kr Krypton
5	37 Rb Rubidium 4.6 x 10 ⁻⁶	38 Sr Strontium 4.6 x 10 ⁻⁶	39 Y Yttrium	40 Zr ^{Zirconium} 6 x 10 ⁻⁶	41 Nb Nicbium 1.6 x 10 ⁻⁶	42 Mo Nolybdenum 1.30 x 10 ⁻⁷	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver 1.0 x 10 ⁻⁸	48 Cd Cadmium 7.20 x 10 ⁻¹	49 In Indium	50 Sn ^{Tin} 2.40 x 10 ⁻⁷	51 Sb Antimony 1.10 x 10 ⁻⁷	52 Te Tellurium 1.20 x 10 ⁻⁷	53 Icdine 1.60 x 10 ⁻⁷	54 Xe Xenon
6	55 CS ^{Cesium} 2.1 x 10 ⁻⁸	56 Ba Barium 3.10 x 10 ⁻⁷	57-71	72 Hf Hafnlum	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Iridium	78 Pt Platinum	79 Au Gold 3 x 10 ⁻⁹	80 Hg Mercury 1.90 x 10 ⁻¹	81 TI Thellum	82 Pb Lead 1.7 x 10 ⁻⁶	83 Bismuth	84 Po Polonium	85 At Astatine	86 Rn Redon
7	87 Fr Francium	88 Ra Radium	89-103	104 Rf Rutherfordium	105 Db Dubrium	106 Sg Seeborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Dermstadtium	111 Rg Roentgenium	112 Copernicium	n Nihonium	114 Fl Flerovium	115 Mc Noscovium	116 Lv Livermorium	117 TS Tennessine	118 Og Oganesson
			57 L Lantf 1.37:	a Cer k 10 ⁻⁶ 5.70	e 1 1 10 ⁻⁷ Praseco × 10 ⁻⁷	f r N dymium Neod	d ymium Prome	m sthium Sama	nium 63 Euro	ju Gadol	d Ter	rbium 66 Dys)y prosium Hol	68 Mium Er	ir Num Th	alium 70 Ytte	rbium 71 Lute	.U etium
			89 Acti	IC T	h P Protax	a 92 Ura	J 93 Nepti	Piuto	u Arre	m Ci	ium 97 Ber	Sk (Cf E Ifornium	inium Fer	mium Hend	Id Nob	elium	.ľ encium







SI U	nit pre	fixes (l	arge)	
-	Factor	Prefix	Symbol	
	10 ¹	deka	da	
	10 ²	hecto	d	
	10 ³	kilo	k	
	10 ⁶	mega	М	
	10 ⁹	giga	G	
	1012	tera	Т	
	10 ¹⁵	peta	Р	
	1018	еха	E	
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SI U	Jnit pro	efixes ((small)	_
	Factor	Prefix	Symbol	
	10-1	deci	d	
	10-2	centi	С	
	10-3	milli m		
	10 ⁻⁶	micro	μ	
	10-9	nano	n	
	10-12	pico	р	
	10-15	femto	f	
	10-18	atto	а	
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	1 6		100 at 100				
what Tom is Ma	de of wor	ald fill a cu	ibe				
	Element	of length	mass				
	- oxygen	33.5 cm	43 kg				
Oxygen-	carbon	19.2 cm	16 kg				
found	hydrogen	46.2 cm	7 kg				
mostly	calcium	8.64 cm	1.0 kg				
component	phosphorus	7.54 cm	780 g	lom	VC	Mucro	
of water,	potassium	5.46 cm	140 g		V D	PICO	
which makes	sodium	4.07 cm	140 g				/
weight	chlorine	3.98 cm	95 g				
	magnesium	2.22 cm	19 g				
	iron fluorine	8.1 mm	4.2 g				
	zinc	6.9 mm	2.3 g		Flome	nt mM/I M/M-I	2
	a	7.5 mm	1.0 g		Lieme		
most abundant	rubidium	7.6 mm	0.68 g		C	11000	14/
element in the body	strontum	4.0 mm	0.32 g		N	1400	16
biological role	- lead	2.2 mm	0.12 g		P	120	1
	copper	2.0 mm	72 mg		1	120	1 2
	 aluminum cadmium 	2.8 mm	60 mg		5	170	1.3
	- cerium	1.7 mm	40 mg		K	250	1.7
	barium	1.8 mm	22 mg		Ma	100	0.56
	a iodine	1.6 mm	20 mg		Ca.	3100	23
	titanium	1.6 mm	20 mg		cu	5100	2.5
	o —— boron	2.0 mm	18 mg		Sr	/ 0	.054
	 —— nickel selenium 	1.2 mm	15 mg		Fe	0.7 0.0	0075
	 chromium 	1.3 mm	14 mg		Mn	0.42 0.0	0038
	manganese	1.2 mm	12 mg		Zn	0.08 0.0	000
	 arsenic lithium 	1.1 mm	7 mg		211	0.08 0.0	0000
	 cesium 	1.5 mm	6 mg		Cu	0.035 0.00	0038
	· mercury	0.8 mm	6 mg		Co	0.024 0.00	0019
	germanium molybdenum	1.0 mm	5 mg		Cd	0.017 0.00	0021
	· cobalt	0.7 mm	3 mg		Mo	0.0021.0.00	022
	· antimony	0.7 mm	2 mg		MO	0.0031 0.000	1022
	silver	0.6 mm	2 mg				
	zirconium	0.54 mm	1 mg				
	lanthanium	0.51 mm	0.8 mg				
	- — gallium	0.49 mm	0.7 mg				
	· tellurium	0.46 mm	0.7 mg				
	bismuth	0.37 mm	0.5 mg				
	- — thallium	0.35 mm	0.5 mg				
	apid	0.36 mm	0.4 mg				
	scandium	0.41 mm	0.2 mg				
	tantalum	0.23 mm	0.2 mg				
element that has	vanadium thorium	0.26 mm (0.20 mm	0.1 mg				
a known biologic	· uranium	0.17 mm	0.1 mg				
Data credit: role.	samarium	0.19 mm	50 µg				45
http://web2.airmail.net/uthman/elements_of_body.html	beryllium	0.27 mm 0.10 mm	36 µg 20 µg	:E 3/U L#3			70





